

The Rotational Spectrum of Water between 650 and 50 cm^{-1} H_2^{18}O and H_2^{17}O in Natural Abundance¹

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In the far infrared spectrum of water, 121 H_2^{18}O and 48 H_2^{17}O pure rotational lines have been identified. From the line frequencies and a minimum number of literature values of near infrared rotation-vibration transitions and lines from microwave spectra, the ground-state energy levels have been calculated up to $J = 12$ and 11, respectively. For low J levels, the agreement with earlier calculations from other microwave lines and near infrared results is within the experimental error, but deviations up to 0.6 cm^{-1} are found for high J , high K_a levels.

INTRODUCTION

The far infrared spectrum of natural water is of considerable interest, because rotational lines of the various isotopic species of water are responsible for most of the atmospheric absorption in the region below 650 cm^{-1} . The transitions of the H_2^{16}O molecule are quite well known, partly from direct measurement (1-4), and partly as calculated from the ground-state levels of Flaud *et al.* (5) with an accuracy of a few thousandths of a cm^{-1} .

Contrary to this situation, no systematic investigation of the H_2^{18}O and H_2^{17}O rotational spectra has been made above 40 cm^{-1} . A few rotational lines of both isotopic species have been reported in the literature, generally in infrared test spectra (e.g., (6-8)) without frequency values or assignments. Two (unidentified) H_2^{18}O line frequencies have been given: $3_{21} \leftarrow 3_{12}$ at 37.913 cm^{-1} (9) and $7_{07} \leftarrow 6_{16}$ at 138.431 cm^{-1} (2). (Here and in the following the quantum numbers are given in the order J, K_a, K_c , for the water molecule equivalent to J, K_{-1}, K_{+1} .) In the region below 30 cm^{-1} , 12 and 7 lines, respectively, of H_2^{18}O and H_2^{17}O have been measured using microwave techniques (see, e.g., (11, 12)). After completion of the present work, Fleming and Gibson (10) published frequencies and assignments of 12 H_2^{18}O lines and 1 H_2^{17}O line between 13 and 40 cm^{-1} . Their spectra were obtained with a Michelson-type interferometer.

For low quantum numbers, the ground-state levels of H_2^{18}O have been calculated several times from rotation-vibration bands, most recently by Toth and Margolis (13). In their paper earlier determinations are listed for comparison. De Lucia and Helminger (12) have calculated a set of energy levels for H_2^{17}O based upon the microwave lines

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Table I
Assignment and Positions (cm^{-1}) of H_2^{18}O Rotational Lines

H_2^{18}O Lines									Overlapping Lines								
J'	K' _A	K' _C	J''	K'' _A	K'' _C	Observed	Calc.-Obs.	Wt. Inten.	J'	K' _A	K' _C	J''	K'' _A	K'' _C	Pos./Sep.	Isot.	Int.Rat.
2	2	1	2	1	2	54.498	-0.010	2 37	8	2	6	8	1	7	54.526	HDO	37/1
5	3	2	5	2	3	60.562	0.011	1 32									
3	2	2	3	1	3	63.192	-0.004	1 19	3	3	1	4	0	4	63.167	H_2^{16}O	1/1
5	2	3	4	3	2	65.878	-0.012	2 5.9	9	2	7	9	1	8	65.909	HDO	1/8
4	1	3	3	2	2	70.050	-0.003	2 10									
3	3	0	3	2	1	71.523	-0.014	2 43									
3	1	3	2	0	2	71.658	-0.017	2 55									
5	1	4	5	0	5	74.322	-0.008	1 44									
4	2	3	4	1	4	74.793	-0.001	1 61									
7	2	5	7	1	6	78.757	0.003	1 16									
4	0	4	3	1	3	79.665	0.001	2 61									
4	3	2	4	2	3	80.671	0.002	4 58									
4	1	4	3	0	3	87.491	0.002	8 239									
2	2	1	1	1	0	91.449	0.003	9 206									
2	2	0	1	1	1	98.064	-0.029	2 66	7	1	7	6	0	6	98.131	HDO	5/1
5	1	4	4	2	3	99.750		4 72	12	4	8	12	3	9	99.7763	H_2^{16}O	8/1
						(99.747)	-0.007										
5	0	5	4	1	4	100.198		13 255	10	3	7	10	2	8	100.1809	H_2^{16}O	1.4/1
						(100.211)	0.008										
8	2	6	8	1	7	100.751	-0.020	2 3.3									
4	4	0	4	3	1	101.965	0.007	2 12									
4	4	1	4	3	2	103.345	0.009	5 36	5	4	2	5	3	3	103.952	H_2^{18}O	7/1
5	1	5	4	0	4	104.005		10 94	10	4	7	9	5	4	104.0389	H_2^{16}O	1.6/1
						(103.984)	-0.001										
6	2	5	6	1	6	105.078		11 43	9	4	5	8	5	4	105.0690	H_2^{16}O	1/4
						(105.115)	-0.009										
7	4	4	7	3	5	109.176	-0.035	2 5.3									
3	2	2	2	1	1	109.973	-0.005	9 81	8	1	8	7	0	7	110.046	HDO	7/1
6	0	6	5	1	5	119.631	0.001	9 90	6	2	5	5	1	4	119.639	HDO	9/1
6	1	6	5	0	5	121.305	-0.004	17 279									
4	2	3	3	1	2	125.738	0.001	20 249									
6	5	1	6	4	2	127.907	0.025	1 5.5	7	2	6	6	1	5	127.920	HDO	1/1.6
5	5	0	5	4	1	128.901	-0.007	3 18	7	5	3	7	4	4	129.105	H_2^{18}O	4.8/1
6	5	2	6	4	3	129.044	0.004	3 17	7	1	6	7	0	7	129.0829	H_2^{16}O (010)	2.3/1
7	2	5	6	3	4	135.056	0.013	4 21									
7	0	7	6	1	6	138.434	-0.000	25 233	4	3	2	3	2	1	138.422	HDO	11/1
8	1	7	8	0	8	138.572	0.011	4 5.6									
7	1	7	6	0	6	139.162	-0.020	1 79									
5	2	4	4	1	3	139.405	-0.038	1 75	7	1	7	6	0	6	139.445	H_2^{17}O	5/1
8	2	7	8	1	8	140.959	-0.044	1 17									
3	3	1	2	2	0	147.309	0.003	15 159									
3	3	0	2	2	1	148.833	-0.001	8 483									
6	2	5	5	1	4	152.090	0.003	15 182	10	3	8	10	2	9	152.029	H_2^{18}O	6.8/1
									6	6	0	6	5	1	{-0.038}	H_2^{18}O	3/1
6	6	1	6	5	2	153.120	-0.015	3 6.4	8	6	3	8	5	4	153.066	H_2^{16}O	2/1

and upon higher-order centrifugal distortion constants taken from the other water isotopes. H_2^{17}O lines have been observed in rotation-vibration bands (14-16), but some assignments were uncertain, and energy levels from combination differences of these infrared lines have not been published.

A selection of H_2^{16}O lines in the spectra which are the basis of this work have been used for calibration of the spectrometer. To avoid errors in the calibration due to overlapping lines a complete assignment had to be made, also including rotational transitions of HDO and of the (010) vibrational state of H_2^{16}O . These and the ground-state lines of H_2^{16}O will be the subject of another paper.

TABLE I—Continued

H ₂ ¹⁸ O Lines										Overlapping Lines									
J'	K' _A	K' _C	J''	K'' _A	K'' _C	Observed	Calc. -	Obs.	Wt. Inten.	J'	K' _A	K' _C	J''	K'' _A	K'' _C	Pos./Sep.	Isot.	Int. Rat.	
8	0	8	7	1	7	156.933	-0.005		3 57	11	1	10	10	2	9	156.901	HDO	29/1	
8	1	8	7	0	7	157.222	0.001		12 172	8	0	8	7	1	7	157.239	H ₂ ¹⁷ O	17/1	
7	2	6	6	1	5	165.426	-0.006		8 44										
4	3	2	3	2	1	168.490	0.005		25 400										
4	2	2	3	1	3	172.900	-0.007		11 51	8	1	7	7	2	6	172.878	H ₂ ¹⁸ O	2/1	
9	0	9	8	1	8	175.263	-0.002		9 109										
9	1	9	8	0	8	175.384	-0.001		6 36										
8	2	7	7	1	6	180.218	0.004		6 87	11	4	7	10	5	6	180.2893	H ₂ ¹⁶ O	1/1	
5	3	2	5	0	5	181.680	0.007		2 6.9										
5	3	3	4	2	2	186.135	0.006		14 95										
9	1	8	8	2	7	192.852	-0.001		11 48										
9	2	8	8	1	7	196.423	-0.007		5 17										
9	2	7	8	3	6	196.505	-0.008		1 16										
6	3	4	5	2	3	200.210	0.019		15 179										
4	4	1	3	3	0	200.337	0.003		30 545	{	6	4	2	5	3	3	200.305	HDO	30/1
4	4	0	3	3	1	200.578	0.004		20 182		11	8	3	11	7	4	200.3617	H ₂ ¹⁶ O	32/1
5	3	2	4	2	3	207.113	0.000		29 286										
7	3	5	6	2	4	211.519	0.010		10 34										
11	0	11	10	1	10	211.668			10 31	11	1	11	10	0	10	(0.021)	H ₂ ¹⁸ O	3/1	
						(211.662)	-0.010												
10	1	9	9	2	8	211.813	0.014		2 8.4										
10	2	9	9	1	8	213.499	0.007		5 26										
8	3	6	7	2	5	221.255	0.003		3 54										
5	2	3	4	1	4	221.347	-0.015		4 96	8	3	5	7	2	6	221.369	HDO	17/1	
5	4	2	4	3	1	223.850	-0.002		6 129										
5	4	1	4	3	2	225.501	0.007		27 390										
12	1	12	11	0	11	229.748			6 4.5	12	0	12	11	1	11	(0.009)	H ₂ ¹⁸ O	3/1	
						(229.750)	-0.002												
11	1	10	10	2	9	230.288	-0.010		2 12	12	1	12	11	0	11	230.216	H ₂ ¹⁷ O	3.6/1	
10	3	8	9	2	7	242.089	0.005		6 12										
4	3	2	3	0	3	242.957	0.002		6 23										
6	3	3	5	2	4	244.393	0.055		1 54										
5	5	0	4	4	1	251.028			50 375	5	5	1	4	4	0	(-0.033)	H ₂ ¹⁸ O	3/1	
						(251.036)	0.011												
6	4	2	5	3	3	251.583	0.016		20 80										
7	4	4	6	3	3	263.275			28 40	4	4	1	4	1	4	263.2695	H ₂ ¹⁶ O	1/4.8	
						(263.300)	-0.006												
6	5	2	5	4	1	275.288	0.004		25 231										
6	5	1	5	4	2	275.579	0.000		14 77										
8	4	5	7	3	4	277.087	0.008		12 53										
5	3	3	4	0	4	279.351	0.017		2 11										
9	4	6	8	3	5	287.152			6 7.1	10	5	6	9	4	5	287.167	HDO	3/1	
						(287.147)	0.013												
7	3	4	6	2	5	289.100	0.005		9 76										

EXPERIMENTAL DATA

A modified Beckman/RIIC FS-720 Michelson interferometer was used for obtaining the water spectra. The instrument and the computational procedures have been described in (17, 18). Seven separate spectra were considered, each of these being the average of 4 to 11 Fourier transformations of single-sided interferograms run under identical conditions. The resolution of the spectra was in the range $0.08\text{--}0.15\text{ cm}^{-1}$, and different pressures between 0.5 and 17.8 Torr were employed in a 3 m cell at $23 \pm 2^\circ\text{C}$. All regions of the spectrum were recorded at least once at maximum resolution. The spectra were calibrated independently with H_2^{18}O frequencies known to better than $\pm 0.005\text{ cm}^{-1}$ (99% confidence interval), calculated from the levels of Ref. (5). Further details of the calibration procedure and a reproduction of the whole spectrum will be

Table I, continued

H ¹⁸ O Lines										Overlapping Lines										
J'	K' _a	K' _c	J"	K" _a	K" _c	Observed	Calc.-	Obs.	Wt.	Inten.	J'	K' _a	K' _c	J"	K" _a	K" _c	Pos./Sep.	Isot.	Int.Rat.	
10	4	7	9	3	6	294.654	0.007		3	7.8	{	7	7	1	6	6	0	298.738	HDO	4/1(Σ)
7	5	3	6	4	2	298.775			12	41		7	7	0	6	6	1			
						(298.784)	0.028													
6	6	1	5	5	0	299.514			29	176		6	6	0	5	5	1	(0.004)	H ¹⁸ ₂ O	3/1
						(299.513)	-0.009													
7	5	2	6	4	3	300.173	0.007		26	124										
12	4	9	11	3	8	308.190	0.035		1	0.88										
4	4	1	3	1	2	309.766	0.000		1	2.4	{	17	1	16	17	0	17	309.7855	H ¹⁶ ₂ O	3/1(Σ)
												2	16	17	1	17		309.7891		
8	4	4	7	3	5	313.693			15	19	{	8	7	2	7	6	1	313.714	HDO	2.7/1(Σ)
						(313.685)	0.000					8	7	1	7	6	2			
8	5	4	7	4	3	320.674	0.006		15	57										
6	3	4	5	0	5	321.341	0.002		10	33										
7	6	1	6	5	2	324.112			5	94		7	6	2	6	5	1	(-0.044)	H ¹⁸ ₂ O	3/1
						(324.122)	-0.032													
8	5	3	7	4	4	325.302	0.019		3	19										
9	5	5	8	4	4	339.574	0.016		2	7.4										
8	3	5	7	2	6	340.732	0.005		1	10	{	7	7	1	6	6	0	(-0.0006)	H ¹⁸ ₂ O	3/1
7	7	0	6	6	1	345.840			24	59		4	4	0	3	1	3	345.8556	H ¹⁶ ₂ O	(Σ)1/2.5
						(345.803)	-0.135													
8	6	3	7	5	2	348.229	-0.011		13	44										
8	6	2	7	5	3	348.472			13	15		9	6	3	9	3	6	348.4647	H ¹⁶ ₂ O	1/3.2
						(348.495)	-0.028													
6	4	3	5	1	4	352.675	0.009		5	15										
9	4	5	8	3	6	353.483	0.017		2	21										
7	3	5	6	0	6	367.923	0.000		2	7.8										
8	7	2	7	6	1	370.508			6	27		8	7	1	7	6	2	(0.006)	H ¹⁸ ₂ O	3.3/1
						(370.506)	-0.130													
9	6	4	8	5	3	371.700	-0.014		2	6.0	17	3	14	16	4	13	371.6322	H ¹⁶ ₂ O	14/1	
9	6	3	8	5	4	372.678	0.005		6	18										
10	5	5	9	4	6	381.783	-0.052		1	2.7										
8	8	1	7	7	0	389.818	-0.575		7	15		8	8	0	7	7	1	(0.0001)	H ¹⁸ ₂ O	3/1
10	6	5	9	5	4	393.841	0.003		1	6.5	{	9	7	3	8	6	2	(-0.041)	H ¹⁸ ₂ O	3/1
9	7	2	8	6	3	394.945			3	11		16	5	12	15	4	11	394.9360	H ¹⁶ ₂ O	(Σ)20/1
						(394.955)	-0.123													
9	3	6	8	2	7	397.887	0.005		1	12	{	9	8	2	8	7	1	(0.001)	H ¹⁸ ₂ O	3/1
9	8	1	8	7	2	414.680			2	6.1		8	2	6	7	1	7	414.727	H ¹⁶ ₂ O	(010)(Σ)7/1
						(414.674)	-0.568													
8	4	5	7	1	6	414.937	0.019		2	13										
11	5	6	10	4	7	416.351	0.070		1	2.4										
10	7	4	9	6	3	418.730	0.041		1	4.1										
9	9	0	8	8	1	429.792	-0.025		1	2.9		9	9	1	8	8	0	(-0.0000)	H ¹⁸ ₂ O	3/1
11	7	4	10	6	5	442.816	-0.046		1	1.3										
11	4	7	10	3	8	453.937	-0.012		1	1.9										
9	4	6	8	1	7	454.979	0.019		1	2.6		8	4	5	7	1	6	454.9714	H ¹⁶ ₂ O	(010) 1/1
9	2	7	8	1	8	457.195	0.014		2	7.2										
8	5	4	7	2	5	465.882	0.054		1	3.4										
9	3	7	8	0	8	470.281	0.000		1	2.3										
10	4	7	9	1	8	499.707	-0.005		1	3.6										
7	5	2	6	2	5	500.733	0.038		1	1.9										
11	3	8	10	2	9	520.198	-0.068		1	1.6										
10	3	8	9	0	9	524.059	-0.017		2	3.2										

presented in a succeeding report. Three short sections of the 17.8 Torr spectrum containing some H₂¹⁸O and H₂¹⁷O lines are given here as Fig. 1.

Most of the line frequencies were determined by the polynomial fit program described in (17), but all were checked manually for obvious distortions. Some partly resolved lines were separated graphically on a large scale plot. The digitizing interval of spectra extending to >400 cm⁻¹ was 0.038 cm⁻¹ and of those of the low-frequency region, 0.019 cm⁻¹. Each line in each spectrum was weighted according to its appearance as follows: 1, in noise, uncertain, ±0.04 cm⁻¹; 2, about two times the mean noise level,

$\pm 0.03 \text{ cm}^{-1}$; 3, weak but clear, $\pm 0.025 \text{ cm}^{-1}$; 5, medium, slight distortions, $\pm 0.01 \text{ cm}^{-1}$; 10, strong, undistorted, $\pm 0.005 \text{ cm}^{-1}$ (if a single line). The mean frequencies of Tables I and II were calculated accordingly from the spectra. The total weight (Wt.) of each line is given in Tables I and II, also as a rough indication of the line intensity, which was considered to be too uncertain to justify a more precise evaluation. Because the region below 150 cm^{-1} was not recorded at maximum pressure, it should be noticed that

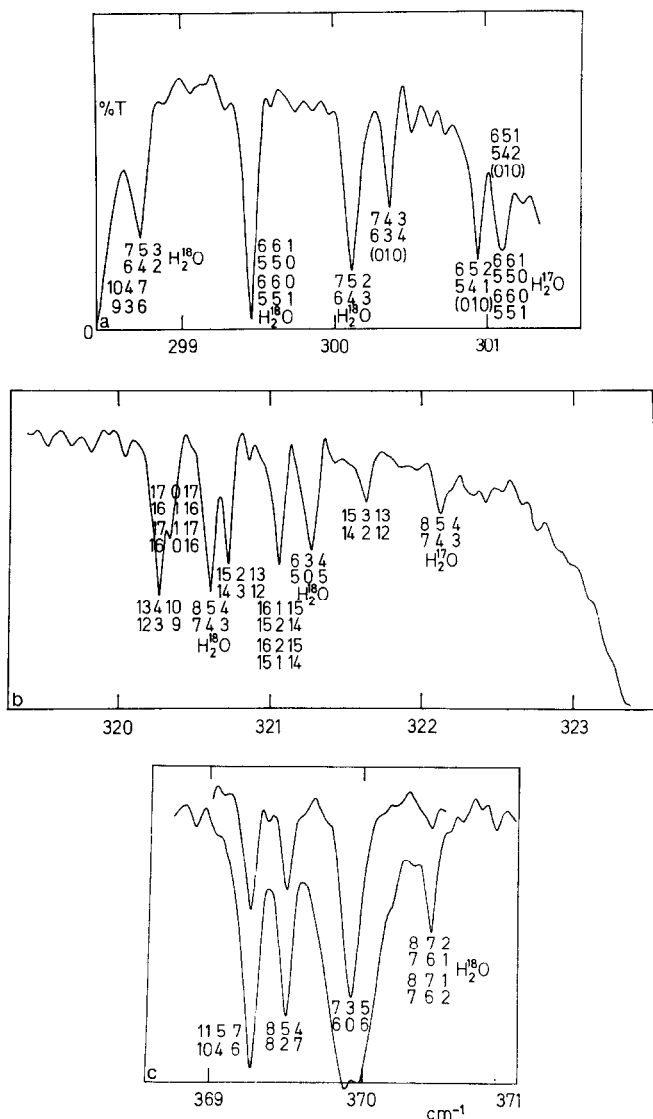


FIG. 1. Three sections of the room temperature water spectrum showing some H_2^{18}O and H_2^{17}O lines in natural abundance. The quantum numbers J, K_a, K_c are given in this order, with the upper state above the lower one. Lines without isotopic specification are due to H_2^{16}O in the ground or (010) state. Path length = 3 m, pressures 17.8 and 5 Torr, and resolution 0.08 cm^{-1} .

Table II
Assignment and Positions (cm^{-1}) of H_2^{17}O Rotational Lines

H ₂ ¹⁷ O Lines									Overlapping Lines								
J'	K' _a	K' _c	J''	K'' _a	K'' _c	Observed	Calc.- Obs.	Wt. Inten.	J'	K' _a	K' _c	J''	K'' _a	K'' _c	Pos./Sep.	Isot.	Int.Rat.
5	3	2	5	2	3	61.437	-0.051	1 5.8									
3	1	3	2	0	2	71.893	0.014	1 9.9									
4	2	3	4	1	4	75.184	-0.047	1 11									
2	2	1	1	1	0	91.958	0.001	2 37									
4	4	1	4	3	2	104.414	-0.011	1 6.6									
6	2	5	6	1	6	105.383	-0.017	2 7.8									
6	0	6	5	1	5	119.861	-0.022	2 16									
6	1	6	5	0	5	121.576	0.011	7 51	9	1	9	8	1	8	121.584	HDO	34/1
4	2	3	3	1	2	126.335	-0.003	5 45									
3	2	1	2	1	2	132.204	0.019	3 36									
7	0	7	6	1	6	138.684	0.012	4 42									
3	3	1	2	2	0	148.109	0.023	3 29									
3	3	0	2	2	1	149.623	0.003	10 88									
6	2	5	5	1	4	152.725	0.007	4 33									
7	2	6	6	1	5	165.988	0.040	2 8.0									
4	3	2	3	2	1	169.383	-0.012	4 72									
9	0	9	8	1	8	175.609	0.004	5 20									
4	3	1	3	2	2	176.705	-0.009	5 25									
8	2	7	7	1	6	180.761	0.007	3 16									
5	3	3	4	2	2	187.096		7 17	7	4	3	8	1	8	187.0743	H ₂ ¹⁶ O	2/1
						(187.107)	-0.005										
9	1	8	8	2	7	193.157	-0.010	1 8.7									
10	1	10	9	0	9	193.943		2 11	10	0	10	9	1	9	(-0.054)	H ₂ ¹⁷ O	3/1
						(193.956)	-0.018										
6	3	4	5	2	3	201.276	0.001	7 32									
4	4	1	3	3	0	201.443	0.002	14 99									
4	4	0	3	3	1	201.683	-0.005	6 33									
5	3	2	4	2	3	207.743	-0.001	10 52									
11	0	11	10	1	10	212.108		1 5.5	11	1	11	10	0	10	(0.023)	H ₂ ¹⁷ O	3/1
						(212.103)	-0.018		3	3	1	2	0	2	212.169	H ₂ ¹⁸ O	3/1
5	2	3	4	1	4	221.481	0.012	1 17									
5	4	2	4	3	1	224.995	-0.004	3 23									
5	4	1	4	3	2	226.597	-0.002	7 71	8	4	5	7	3	4	226.603	HDO	10/1
10	3	8	9	2	7	243.069	0.028	1 2.2									
6	3	3	5	2	4	244.881	-0.042	1 9.8									
6	4	3	5	3	2	246.526	0.008	11 42									
5	5	0	4	4	1	252.402		20 68	5	5	1	4	4	0	(-0.031)	H ₂ ¹⁷ O	3/1
						(252.409)	0.008										
6	4	2	5	3	3	252.627	0.006	2 15									
7	4	4	6	3	3	264.663	0.001	4 7.2									
6	5	2	5	4	1	276.682	0.006	6 42									
6	5	1	5	4	2	276.966	-0.002	3 14									
7	4	3	6	3	4	281.235	0.000	3 23	6	5	1	5	4	2	301.1834	H ₂ ¹⁶ O(010)	2.5/1
6	6	1	5	5	0	301.157		11 32	6	6	0	5	5	1	(0.004)	H ₂ ¹⁷ O	3/1
						(301.141)	0.006										
7	5	2	6	4	3	301.581		3 22	7	3	4	6	2	5	301.5940	H ₂ ¹⁶ O(010)	1.6/1
						(301.573)	-0.016										
8	4	4	7	3	5	314.334	0.009	1 3.4									
8	5	4	7	4	3	322.202	0.005	3 10	7	5	2	6	4	3	325.7850	H ₂ ¹⁶ O(010)	1/1.4
7	6	1	6	5	2	325.744	0.014	10 17	7	6	2	6	5	1	(-0.041)	H ₂ ¹⁷ O	3/1
8	5	3	7	4	4	326.637	0.015	2 3.5									
7	7	0	6	6	1	347.657	0.008	3 11	7	7	1	6	6	0	(-0.001)	H ₂ ¹⁷ O	3/1
8	7	2	7	6	1	372.370	0.015	3 5.0	8	7	1	7	6	2	(0.006)	H ₂ ¹⁷ O	3.3/1
8	8	1	7	7	0	391.874	0.021	1 2.7	8	8	0	7	7	1	(0.000)	H ₂ ¹⁷ O	3/1

the total weight of the lines here is lower than if it were corresponding to their actual intensity.

Tables I and II list all observed lines of H_2^{13}O and H_2^{17}O , respectively. Some lines are unresolved doublets or are overlapped by lines due to other isotopic species. To obtain more accurate transition frequencies from such blended lines, it was assumed that

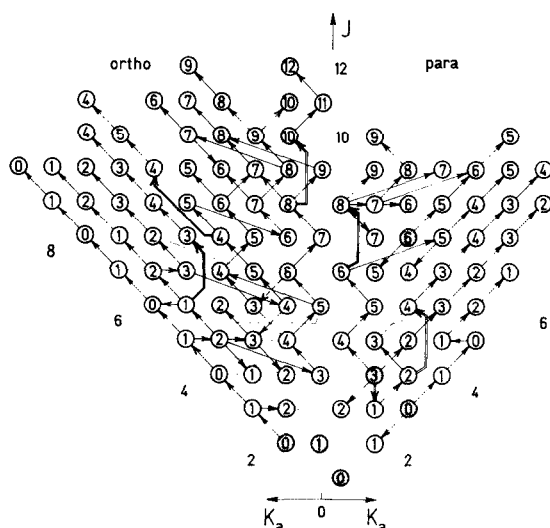


FIG. 2. Schematic representation of the ground-state rotational levels and far infrared transitions of H_2^{18}O . Each level is given as a circle with the K_c quantum number inside. J and K_a are ordinate and abscissa, respectively. Dotted lines indicate microwave measurements and double lines represent combination differences, both from Table V. All other lines in the scheme show far infrared transitions observed in this work and listed in Table I. Lines without arrowheads have not been used in the calculation of Table III. It should be noted that the arrows show the direction of calculation, not that of the transition of the molecule.

the observed line position was the mean of the frequencies of the components, weighted by their respective intensities. This has been verified for doublets recorded at spectral slit widths larger than five times the separation (21). The method is just approximate

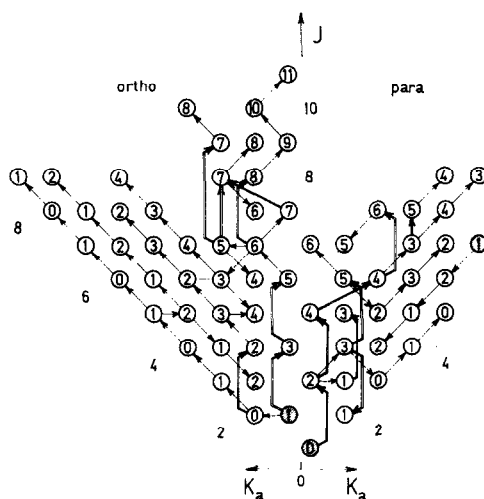


FIG. 3. Schematic representation of the ground-state rotational levels and far infrared transitions of H_2^{17}O (see caption of Fig. 2).

T a b l e I I I
Ground State Energy Levels of H_2^{16}O and H_2^{17}O in cm^{-1}

J	K _a	K _c	H_2^{16}O		H_2^{17}O		H_2^{16}O		H_2^{17}O	
			This Work	Ref. (12)	This Work	Ref. (12)	This Work	Ref. (12)	This Work	Ref. (12)
0	0	0	0.000	0.000	0.000	0.000	881.925	881.900	883.650	883.664
1	0	1	23.755	23.796	23.773	23.773	980.236	980.230	981.504	981.504
1	1	1	36.720	36.750	36.931	36.931	1001.688	1001.709	1003.791	1003.791
1	1	0	42.023	42.024	42.187	42.187	1047.340	1047.317	1048.662	1048.662
2	0	2	69.911	69.930	70.004	70.004	1116.644	1116.620	1119.484	1119.484
2	1	2	78.975	78.995	79.210	79.226	1126.455	1126.480	1128.935	1128.965
2	1	1	94.788	94.790	94.972	94.970	1246.379	1246.369	1250.499	1250.505
2	2	1	133.473	133.480	134.145	134.144	1247.167	1247.20	1251.310	1251.298
2	2	0	134.782	134.785	135.427	135.430	1399.444	1399.74		
3	0	3	136.341	136.344	136.537	136.536	1574.716		1582.226	
3	1	3	141.358	141.377	141.899	141.901	1768.823		1778.292	
3	1	2	172.885	172.888	173.109	173.109	916.265	916.250	918.090	
3	2	2	204.761	204.765	205.451	205.479	916.298	916.300		
3	2	1	210.808	210.800	211.414	211.434	1074.777	1074.779	1076.788	
3	3	1	282.093	282.093	283.536	283.559	1075.909	1075.920		
3	3	0	282.305	282.304	283.768	283.765	1198.195	1198.190	1199.982	
4	0	4	221.233	221.237	221.624	221.618	1211.195			
4	1	4	223.832	223.830	224.282	224.302	1279.821	1279.800		
4	1	3	274.811	274.793	275.130	275.128	1334.476	1334.48		
4	2	3	298.624	298.629	299.444	299.436	1355.171	1355.19		
4	2	2	314.468	314.460	315.075	315.076	1466.029			
4	3	2	379.298	379.293	380.797	380.804	1468.615			
4	3	1	380.705	380.700	382.136	382.174	1618.867			
4	4	1	482.644	482.632	485.211	485.209	1619.057			
4	4	0	482.671	482.660	485.219	485.237	1794.400			
5	0	5	324.043	324.044	324.661	324.658				
5	1	5	325.217	325.216	325.876	325.878	1989.389			
5	1	4	398.367	398.372	398.895	398.877	2198.615			
5	2	4	414.173	414.177	415.130	415.126				
5	2	3	445.160	445.160	445.765	445.792	1109.809	1109.809	1112.046	
5	3	3	500.593	500.595	502.182	502.179	1109.770	1109.770		
							1287.722	1287.727		
							1288.276	1288.264		

TABLE III—Continued

5	3	2	505.736	505.727	507.187	507.174	10	2	9	1433.061
5	4	2	604.555	604.550	607.181	607.165	10	3	8	1440.304
5	4	1	604.799	604.796	607.395	607.403	10	3	7	1443.051
5	5	1		733.673	737.629	737.629	10	4	7	1574.484
5	5	0	733.690	733.678	737.620	737.632	10	5	5	1716.259
6	0	6	444.848	444.860	445.737	445.717	10	6	5	1862.455
6	1	6	445.347	445.355	446.237	446.243	10	7	4	2037.787
6	1	5	541.182	541.183	542.048	541.996	11	0	11	1321.471
6	2	5	550.457	550.447	551.620	551.610	11	1	10	1518.564
6	2	4	501.251	601.246		501.961	11	3	8	1808.473
6	3	4	645.377	645.387	647.062	647.074	11	4	7	1894.242
6	3	3	658.565	658.623	660.010	659.988	11	5	6	1990.835
6	4	3	751.043	751.040	753.713	753.712	11	7	4	2305.272
6	4	2	752.186	752.195	754.809	754.819	12	1	12	1551.119
6	5	2	880.087	880.079	884.076	884.104	12	4	9	2116.663
6	5	1	880.134	880.121	884.112	884.140				
6	6	1	1033.203	1033.195	1038.761	1038.934				
6	6	0		1033.195		1038.934				
7	0	7	583.781	583.767	584.921	584.941				
7	1	7	593.981	593.972	595.162	595.162				
7	1	6	701.707	701.697	702.888	702.888				
7	2	6	<u>705.608</u>	706.608	708.022	708.022				
7	2	5	780.433	780.440	781.281	781.281				
7	3	5	812.770	812.753	814.630	814.616				
7	3	4	839.957	839.961	840.868	840.868				
7	4	4	921.865	921.910	924.673	924.647				
7	4	3	925.705	925.698	928.296	928.301				
7	5	3	1050.970	1050.990	1055.082	1055.082				
7	5	2	1051.216	1051.210	1055.285	1055.283				
7	6	2		1204.172						
7	6	1	1204.209	1204.186	1209.896					
7	7	1		1378.98						
7	7	0	1379.006	1378.98	1386.418					
8	0	8	740.914	740.920	742.401	742.401				
8	1	8	741.003	741.000	742.481	742.494				
8	1	7	379.486	379.481	281.108	281.108				

The underlined levels have been used as basis for the calculation, as explained in the text.

as used here because the slit width dependence was left out, but the frequency corrections of Tables I and II are so small (generally $\leq 0.01 \text{ cm}^{-1}$) that the residual errors are less than the precision of the line positions. Exceptions are the lines of H_2^{18}O $7_{44} \leftarrow 6_{33}$ (corr. 0.025 cm^{-1}) and $7_{70} \leftarrow 6_{61}$ (corr. 0.038 cm^{-1}), where the error of the correction may be as high as 50%. For H_2^{16}O , doublet separations such as, for instance, between $6_{61} \leftarrow 5_{50}$ and $6_{60} \leftarrow 5_{51}$ are well known (5). The calculated separations for the corresponding H_2^{18}O and H_2^{17}O doublets (19, 12) are almost unchanged from those of H_2^{16}O so that the correction method described above may be applied with confidence, even where the absolute level height for H_2^{18}O and H_2^{17}O is uncertain. In Tables I and II the corrected frequencies are given in brackets below the observed ones.

The calculated frequencies of Table I are from Ref. (19). For those of Table II the empirical expression $\nu_{17} = (1 - 0.529)\nu_{16} + 0.529 \nu_{18}$ given in Ref. (14) has been used. (The indices of the expression refer to the corresponding oxygen isotope.) ν_{16} 's have been obtained from the levels of Ref. (5) and ν_{18} 's were taken from this work. The empirical expression was used for calculating the line positions of Table II because it yields precise predictions and because high J levels are missing from the list of Ref. (12). Where both calculation methods were applicable, the agreement is within the experimental error, except for the $6_{61} \leftarrow 5_{50}$ transition. The line is predicted from Ref. (12) to be at 301.302 cm^{-1} , whereas the empirical expression gives 301.147 cm^{-1} , much nearer the observed line at 301.157 (corr. 301.141) cm^{-1} .

The intensities (Int.) listed in Tables I and II in units of $10^{-23} \text{ cm}^{-1}/\text{molecules} \cdot \text{cm}^{-2}$ have been obtained from the intensities of the corresponding H_2^{16}O lines as given in Ref. (20), multiplied by the isotopic ratios. For overlapping lines the positions (Pos.) have been taken from this work in the case of H_2^{18}O or H_2^{17}O lines, from Ref. (20) for HDO lines, and from Ref. (5) for H_2^{16}O or $\text{H}_2^{16}\text{O}(010)$ lines. Where instead the separations (Sep.) of narrow doublets are given in brackets, they have been obtained from Ref. (19) or (12). The sign convention for the doublet separation is: Positive, when the main component is at lower frequency than the overlapping one. The approximate intensity ratio of the line of interest and overlapping lines is given in the last column

T a b l e I V
Comparison of Energy Levels (cm^{-1}) Used for Table III

J	K _a	K _c	H_2^{18}O				H_2^{17}O
			Ref. (14)	Ref. (15)	Ref. (13)	Ref. (19)	Ref. (12)
1	0	1	23.756	23.759	23.756	<u>23.7548</u>	23.773
1	1	0	42.024	42.023	42.024	<u>42.0233</u>	
2	2	0	134.785	134.784	134.785	<u>134.7832</u>	
3	1	3	141.576	141.581	141.577	<u>141.5682</u>	
6	5	1	- difference to level 6_{52}				0.036
7	2	6	706.608	706.603	<u>706.608</u>	706.6032	

The underlined levels have been used in this work - see text.

T a b l e V
Additional Data (in cm^{-1}) Used for Table III

Isotopic Species	Upper Level				Ground State Levels						Combination Difference	Lit. Ref.
	v_1	v_2	v_3	J	K_a	K_c	J'	K'_a	K'_c	J''		
H_2^{18}O	(0 0 1)	2	1	2	3	1	3	2	1	1	46.7803	(16)
	(0 0 1)	4	2	3	5	2	4	3	2	2	209.4120	(16)
	(0 0 1)	6	4	2	7	4	3	5	4	1	320.9064	(16)
	(0 0 1)	7	0	7	8	0	8	6	0	6	296.0663	(16)
	(1 0 0)	8	4	5	9	5	4	7	3	4	629.058	(13,14)*
	(0 0 1)	9	1	9	10	1	10	8	1	8	368.8065	(16)
Microwave line:					6	1	6	5	2	3	0.1876	(11)
H_2^{17}O	(0 0 1)	1	0	1	2	0	2	0	0	0	70.006	(16)
	(0 0 1)	2	0	2	3	0	3	1	0	1	112.7640	(16)
	(0 0 1)	2	1	2	3	1	3	1	1	1	104.969	(16)
	(0 0 1)	2	1	1	3	1	2	1	1	0	130.9222	(16)
	(0 0 1)	3	0	3	4	0	4	2	0	2	151.6183	(16)
	(0 0 1)	3	1	2	4	1	3	2	1	1	180.1589	(16)
	(0 0 1)	4	0	4	5	0	5	3	0	3	188.1236	(16)
	(0 0 1)	4	1	4*	5	1	5	3	1	3	183.9777	(16)
	(0 1 0)	5	1	5	5	2	4	4	0	4	193.506	(15)**
	(0 0 1)	6	2	5	7	2	6	5	2	4	292.906	(14)
	(0 1 0)	6	2	4	7	3	5	6	3	3	154.620	(15)†
	(0 0 1)	7	1	7	8	1	8	6	1	6	296.2446	(16)††
	(0 1 0)	7	1	6	8	2	7	6	2	5	332.036	} (15) §
	(0 1 0)	7	1	6	8	2	7	7	0	7	298.684	
	(0 0 1)	8	2	6	9	2	7	7	2	5	418.601	} (14) §§
	(0 0 1)	6	2	4	7	2	5	6	2	5	229.761	
Microwave Lines:					1	1	0	1	0	1	18.4134	(12)
					2	1	1	2	0	2	24.9659	(12)
					3	1	3	2	2	0	6.4712	(12)
					5	1	5	4	2	2	10.8017	(12)
					6	1	6	5	2	3	0.4515	(12)

* Calculated from line 183 (13) and line 158 (14). Lines 960 and 1010 (14) also have (100) 8_{45} as the upper level, but seem to be misassigned.

** The assignment of the (010) $5_{13} \leftarrow (000) 5_{24}$ line has been changed from 1503.739 to 1503.533 cm^{-1} , because the combination difference differed 0.2 cm^{-1} from the calculated value (12).

† The assignment of the (010) $6_{24} \leftarrow (000) 7_{35}$ line has been changed from 1392.250 to 1392.147 cm^{-1} , because of disagreement with the calculated combination difference (12). The change is confirmed by the combination difference $7_{35} \leftarrow 6_{15}$ (from line 540 (15)).

†† The assignment of the (001) $7_{17} \leftarrow (000) 8_{18}$ line has been changed from 3581.89324 to 3581.54970 cm^{-1} because of disagreement with the calculated combination difference (12).

§ The mean of the values 883.656 and 883.605 cm^{-1} has been used for the 8_{27} level of Table III.

§§ The sum of the combination differences has been used for the 9_{27} level of Table III.

(Int. Rat.) of Tables I and II, in this order. For triplets, a (Σ) together with one of the figures indicates that the total intensity of a narrow doublet relative to the third component is given. Again, the intensities have been taken from Ref. (20).

T a b l e V I
Comparison of Energy Level Differences (in cm^{-1})
 H_2^{18}O

Upper Level $J \quad K \quad K_c$	Ground State Levels $J' \quad K' \quad K'_c$	Combination Difference	This Work - Comb. Dif.
1 0 1	2 0 2	69.9272	-0.017
1 0 1*	2 1 2	36.9649	-0.013
2 0 2	3 0 3	112.5828	0.004
2 1 2	3 1 3	104.8194	0.029
2 1 1	3 1 2	130.8596	0.003
2 1 1	3 1 2	93.8943	0.016
2 2 1	3 2 2	69.9728	0.005
2 2 0	3 2 1	77.3233	0.012
3 0 3	4 0 4	151.3068	0.016
3 1 3	4 1 4	144.8396	0.017
3 1 3	4 1 4	50.9454	0.001
3 1 2	4 1 3	180.0144	0.009
3 2 2	4 2 3	165.1446	0.006
3 2 2	4 2 3	162.2830	-0.001
3 2 2	4 2 3	87.8209	-0.006
3 2 1	4 2 2	179.6756	0.009
3 2 1	4 2 2	109.7034	0.004
3 3 1	4 3 2	96.9842	0.008
3 3 0	4 3 1	98.6068	0.006
4 0 4	5 0 5	187.7100	-0.009
4 1 4	5 1 5	183.648	0.000
4 1 3	5 1 4	225.4779	0.004
4 1 3	5 1 4	174.5318	0.004
4 2 2	5 2 3	308.8226	-0.004
4 2 2	5 2 3	234.3596	-0.008
4 2 2	5 2 3	146.5303	-0.002
4 3 2	5 3 3	218.5040	0.006
4 3 1	5 3 2	223.4214	0.009
4 3 1	5 3 2	126.4369	0.001

RESULTS

The number of observed rotational lines of H_2^{18}O and H_2^{17}O was in both cases almost equal to the number of energy levels to be determined. Thus the levels listed in Table III were built up as illustrated in Figs. 2 and 3 from the line frequencies of this work and a minimum number of additional data of comparable accuracy from the literature. In Table IV different values of the levels which were used as starting points in the calculation are compared. These levels are hatched in Figs. 2 and 3 and the direction of arrows shows how the energy level scheme was developed. For H_2^{17}O it was appropriate to connect two series of transitions by the calculated separation of levels 6_{51} and 6_{52} (12). It should be noted that generally the values in Tables I to VII have been calculated to four figures after the decimal point, but just three are given in the Tables. This may lead to insignificant numerical discrepancies.

Table V contains the additional information used for calculating Table III, mainly ground-state combination differences from Ref. (16). Some lines from H_2^{17}O rotation-vibration bands had to be reassigned as specified in the footnotes of Table V. No systematic search for dubious assignments has been made, but the number of reassigned

TABLE VI—*Continued*

4 4 0	5 4 1	4 4 1	122.1482	0.007
5 0 5	6 0 6	4 0 4	225.6131	0.002
5 1 5	6 1 6	4 1 4	221.5181	-0.003
5 2 4	6 2 5	4 2 3	251.8309	} 0.002
5 3 2*	6 2 5	4 2 3	251.8314	
5 2 4	6 2 5	5 0 5	226.4074	0.007
5 2 4	6 2 5	5 2 3	105.2942	0.003
5 2 3	6 2 4	4 2 2	286.7783	0.005
5 3 3	6 3 4	4 3 2	266.0911	-0.013
5 3 3	6 3 4	5 3 2	139.6531	} -0.013
6 3 3	6 3 4	5 3 2	139.6542	
5 4 2	6 4 3	4 4 1	260.3916	0.008
5 4 2	6 4 3	5 4 1	146.2429	0.001
6 5 1	6 5 2	5 5 0	146.3740	0.023
6 0 6	7 0 7	5 0 5	259.7313	0.007
6 1 5	7 1 6	5 1 4	303.3339	0.006
6 2 5	7 2 6	5 2 4	292.4290	0.006
6 2 4	7 2 5	5 2 3	335.2946	-0.021
6 2 4	7 2 5	6 2 5	230.0002	-6.025
7 4 4	7 4 3	6 4 3	174.6668	-0.004
7 1 7	8 1 8	6 1 6	295.6525	0.003
7 2 6	8 2 7	6 2 5	331.4632	0.005
8 0 8	9 0 9	7 0 7	332.4789	0.006
Microwave Lines				
	2 1 1	2 0 2	24.8612	0.016
	4 1 4	3 2 1	13.0293	-0.006
	4 2 3	3 3 0	16.3131	0.004
	5 1 5	4 2 2	10.7563	-0.008
	5 3 3	4 4 0	17.9237	0.009
	5 3 2	4 4 1	23.0853	0.007
	6 2 4	7 1 7	17.2513	0.018
	6 4 3	5 5 0	17.3499	0.002
FIR Lines				
	1 1 1	0 0 0	36.7450	-0.026
	3 1 2	2 2 1	39.4683	0.004
	3 2 1	3 1 2	37.9127	0.010
	4 2 2	4 1 3	39.5518	0.005

Near infrared data from Ref. (16), microwave lines from Ref. (11), and far infrared lines from Ref. (10).

* Upper-level (100) vibrational state, all other upper levels (001) state.

H₂¹⁷O lines indicates that the rotation-vibration bands of this species should be reinvestigated.

To test the accuracy and internal consistency of the energy levels obtained in this work a comparison was made between combination differences calculated from the levels of Table III and ground-state combination differences from such lines of Ref. (16), which have not been evaluated so far. The results are given as Tables VI and VII, also including some comparisons with microwave and far infrared lines which have not been used in the energy level calculation. It is seen that most of the deviations for H₂¹⁸O are less than 0.01 cm⁻¹ (rms deviation = 0.010 cm⁻¹), whereas for H₂¹⁷O the scattering is larger, corresponding to the weakness of the lines of this isotopic species. It is also interesting to note that the average value of a selection of 46 deviations less than 0.01 cm⁻¹ from Table VI is slightly positive: 0.0028 cm⁻¹. This is equivalent to 1/27 of the resolution used in this work and indicates that errors due to unsymmetrical line profiles have not been eliminated completely.

T a b l e V I I
Comparison of Energy Level Differences (in cm^{-1})

H_2^{17}O									
Upper Level			Ground State Levels			Combination		This Work	
J	K _a	K _c	J'	K' _a	K' _c	J''	K'' _a	Difference	- Comb. Dif.
1	1	1	2	1	2	1	1	0	37.0420
2	1	1	3	1	2	2	1	2	93.8831
3	1	3	3	1	2	2	1	2	93.8830
3	1	3	4	1	4	2	1	2	145.0767
3	1	3	4	1	4	3	1	2	51.1937
3	2	2	4	2	3	2	2	1	165.2935
3	2	2	4	2	3	3	2	1	88.0043
4	3	1	4	3	2	3	3	0	97.0388
4	1	3	5	1	4	3	1	2	225.7687
4	2	2	5	2	3	3	2	1	234.3572
4	2	2	5	2	3	4	2	3	146.354
6	0	6	7	0	7	5	0	5	260.2797
Microwave Lines									
			4	1	4	3	2	1	12.8685
			4	2	3	3	3	0	15.6712
FIR Line									
			3	2	1	3	1	2	38.323

Near infrared data from Ref. (16), all upper levels are from the (001) vibrational state. Microwave lines from Ref. (12). The far infrared line is from (10).

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