

Spectrum of H_2^{18}O and H_2^{17}O in the 5030 to 5640 cm^{-1} Region¹

ROBERT A. TOTH

Jet Propulsion Laboratory, California Institute of Technology, Pasadena, California 91103

AND

J. M. FLAUD AND C. CAMY-PEYRET

*Laboratoire de Physique Moléculaire et d'Optique Atmosphérique, Bâtiment 221,
Campus d'Orsay, 91405, Orsay, France*

Measurements of line center positions of H_2^{18}O and H_2^{17}O in the 5030 to 5640 cm^{-1} region have been made at high resolution. This region contains absorptions of the (011) and (110) bands of H_2^{18}O and H_2^{17}O . Values of the ground state levels of H_2^{18}O and H_2^{17}O as well as those for the states of (011) and (110) of H_2^{18}O and H_2^{17}O were determined from the data. Strength measurements of 23 strong absorbing lines of H_2^{18}O were also included and compared to values obtained for H_2^{16}O multiplied by the $\text{O}^{18}/\text{O}^{16}$ isotopic abundance ratio. These results indicate that H_2^{18}O line strengths can be calculated, to good approximation, from the H_2^{16}O values for the (011) and (110) bands.

INTRODUCTION

The present study was undertaken to obtain a knowledge of the line positions of H_2^{18}O and H_2^{17}O in the 1.9 μm spectral region which, in turn, is important for the interpretation of atmospheric spectra. The H_2^{18}O data were obtained in the same manner as those presented for the 2900 to 3400 cm^{-1} region published earlier by Toth and Margolis (1). Additional spectra of H_2^{16}O , which contained absorption of H_2^{18}O and H_2^{17}O , were also used for purposes of line position measurements (2) and line strength measurements (3).

This work encompasses measurements of the line positions in the (011) and (110) bands of H_2^{18}O and H_2^{17}O as well as obtaining values for the ground state and upper state levels. Also strength measurements of 23 H_2^{18}O lines are presented.

EXPERIMENTAL DETAILS

The data were obtained with a 1.8 m Jarrel-Ash grating spectrometer having a 300 line/mm grating blazed at 5.7 μm in first order. Observations were made in third order at a spectral resolution of $\sim 0.05 \text{ cm}^{-1}$. The source was a 1000 W quartz-iodine projection lamp and the detector was a PbS element which was cooled to approximately -60°C . A multitraversal cell of 2 m base length was used for all measurements. The

¹ This paper presents the results of one phase of research carried out at the Jet Propulsion Laboratory, California Institute of Technology, under Contract NAS7-100, sponsored by the National Aeronautics and Space Administration.

TABLE 1. GROUND STATE ENERGY LEVELS (CM⁻¹) OF H₂¹⁸O AND H₂¹⁷O

H ₂ ¹⁸ O			H ₂ ¹⁷ O			H ₂ ¹⁸ O			H ₂ ¹⁷ O		
JK ₁ -K ₁	This Work	Garing and McClatchey ^a	Toth and Margolis ^b	This Work	Garing and McClatchey ^a	JK ₁ -K ₁	This Work	Garing and McClatchey ^a	Toth and Margolis ^b	This Work	Garing and McClatchey ^a
0 0 0	•000	•000	•000	•000	•000	8 0 8	740.918	740.913	740.930	742.414	742.403
1 0 1	23.756	23.755	23.756	23.775	23.773	8 1 8	740.999	741.000	741.000	742.505	742.496
1 1 1	36.750	36.749	36.750	36.940	36.932	8 1 7	879.493	879.495	879.481	881.103	881.098
1 1 0	42.024	42.024	42.024	42.190	42.188	8 2 7	881.917	881.914	881.900	883.659	883.656
2 0 2	69.928	69.927	69.930	70.010	70.004	8 2 6	980.229	980.227	980.230	981.504	981.497
2 1 2	78.990	78.989	78.995	79.236	79.228	8 3 6	1001.711	1001.711	1001.709	1003.602	1003.796
2 1 1	94.789	94.788	94.790	94.973	94.970	8 3 5	1047.331	1047.336	1047.317	1048.672	1048.672
2 2 1	133.477	133.476	133.480	134.141	134.149	8 4 4	1116.639	1116.646	1116.620	1119.516	1119.516
2 2 0	134.785	134.783	134.785	135.425	135.435	8 4 4	1126.443	1126.449	1126.480	1128.971	1128.971
3 0 3	136.338	136.336	136.344	136.540	136.536	8 5 3	1246.380	1246.381	1246.369	1250.940	1250.940
3 1 3	141.568	141.568	141.577	141.908	141.902	8 5 3	1247.210	1247.218	1247.200	1251.333	1251.333
3 1 2	172.884	172.882	172.888	173.107	173.109	8 6 3	1399.437	1399.434	1399.740	1405.199	1405.199
3 2 2	204.758	204.756	204.763	205.477	205.485	8 6 2	1574.658	1574.630	1574.630	1582.245	1582.245
3 2 1	210.800	210.799	210.800	211.435	211.439	8 7 2	1574.658	1574.630	1574.630	1582.244	1582.244
3 3 1	282.095	282.095	282.095	283.570	283.570	8 8 1	1768.750	1768.750	1768.750	1778.395	1778.395
3 3 0	282.306	282.307	282.304	283.776	283.776	8 8 0	1768.750	1768.750	1768.750	1778.395	1778.395
4 0 4	221.234	221.234	221.237	221.623	221.619	9 0 9	916.258	916.261	916.250	918.102	918.111
4 1 4	223.829	223.828	223.830	224.305	224.304	9 1 9	916.297	916.297	916.300	918.157	918.150
4 1 3	274.803	274.803	274.793	275.116	275.129	9 1 8	1074.763	1074.761	1074.779	1076.823	1076.802
4 2 3	298.621	298.620	298.629	299.442	299.442	9 2 8	1075.911	1075.907	1075.920	1198.964	1198.964
4 2 2	314.461	314.460	314.460	315.068	315.082	9 2 7	1198.210	1198.207	1198.190	1213.379	1213.379
4 2 1	379.291	379.293	379.293	380.815	380.815	9 3 7	1211.191	1211.193	1211.193	1337.522	1337.522
4 3 1	380.703	380.704	380.700	382.180	382.186	9 3 6	1279.800	1279.809	1279.800	1357.593	1357.593
4 3 0	482.646	482.646	482.632	485.238	485.238	9 4 6	1334.485	1334.494	1334.680	1470.274	1470.274
4 4 0	482.676	482.674	482.660	485.260	485.266	9 4 5	1355.221	1355.214	1355.190	1472.737	1472.737
5 0 5	324.047	324.047	324.044	324.660	324.660	9 5 5	1466.029	1466.040	1466.040	1624.756	1624.756
						9 6 4	1468.616	1468.634	1618.920	1624.756	1624.756
						9 6 3	1619.071	1619.079			

SPECTRUM OF H₂¹⁸O AND H₂¹⁷O

5	1	5	325.216	325.216	325.861	9	7	3	1794.360	1802.047
5	2	4	398.361	398.372	398.877	9	7	2	1794.360	1822.049
5	2	4	414.170	414.177	415.124	9	8	2	1989.305	1999.034
5	2	3	445.159	445.160	445.787	9	8	1	1989.305	1999.034
5	3	3	500.597	500.598	502.180	9	9	1	2200.430	2212.332
5	3	2	505.730	505.727	507.185	9	9	0	2200.430	2212.332
5	4	2	604.547	604.548	607.187	10	0	0	1109.785	1112.037
5	4	1	604.794	604.796	607.422	10	1	0	1109.601	1112.054
5	5	1	733.681	733.684	737.652	10	1	9	1287.727	1290.230
5	5	0	733.685	733.678	737.656	10	1	9	1287.727	1290.230
6	0	6	444.848	444.860	445.729	10	2	9	1288.271	1288.262
6	1	6	445.346	445.355	446.250	10	2	8	1433.030	1433.061
6	1	5	541.179	541.181	541.994	10	3	8	1440.291	1440.295
6	2	5	550.453	550.447	551.605	10	3	7	1534.385	1534.368
6	2	4	601.236	601.240	601.964	10	4	7	1574.458	1574.475
6	3	4	645.385	645.387	647.081	10	4	6	1611.673	1613.941
6	3	3	658.009	658.614	660.001	10	5	6	1709.575	1709.580
6	4	3	751.036	751.038	753.726	10	5	6	1716.240	1720.252
6	4	2	752.191	752.192	754.838	10	5	5	1862.510	1868.929
6	5	2	880.080	880.079	884.109	10	6	4	1863.050	1868.929
6	5	1	880.115	880.119	884.146	10	6	4	2037.980	2045.732
6	6	1	1033.195	1033.194	1036.815	11	0	11	1321.457	1324.137
6	6	0	1033.195	1033.195	1036.815	11	0	11	1321.456	1324.137
7	0	7	583.782	583.780	584.944	11	1	11	1321.464	1324.144
7	1	7	583.985	583.989	585.157	11	1	10	1518.541	1518.534
7	1	6	701.695	701.695	702.887	11	2	10	1518.798	1518.779
7	2	6	706.599	706.599	708.884	11	2	9	1644.437	1644.462
7	2	5	780.456	780.454	781.025	11	3	8	1808.380	1808.410
7	3	5	812.766	812.767	814.618	11	5	6	1990.880	1990.880
7	3	4	839.555	839.555	840.684	12	0	12	1551.195	1551.200
7	4	4	921.900	921.910	924.666	12	0	12	1551.195	1551.203
7	4	3	925.704	925.707	926.698	12	1	11	1767.211	1767.218
7	5	3	1050.994	1050.991	1055.086	12	2	11	1767.535	1767.531
7	5	2	1051.207	1051.204	1055.267	12	2	11	1767.535	1767.531
7	6	2	1204.170	1204.169	1203.861	12	0	12	1551.195	1551.190
7	6	1	1204.177	1204.174	1203.866	12	0	12	1551.195	1551.190
7	7	1	1378.970	1378.961	1386.486	13	0	13	1767.211	1767.218
7	7	0	1378.970	1378.961	1386.486	13	0	13	1767.535	1767.531

TABLE 3. MEASUREMENTS AND ASSIGNMENTS OF H₂¹⁸O AND H₂¹⁷O IN THE 5030 TO 5640 CM⁻¹ SPECTRAL REGION

Line No.	Frequency (CM ⁻¹)		Upper State JK ₋₁ K ₁	Lower State JK ₋₁ K ₁	Band	Mol.	Line No.	Frequency (CM ⁻¹)		Upper State JK ₋₁ K ₁	Lower State JK ₋₁ K ₁	Band	Mol.	
	(Obs.)	(Calc.)						(Obs.)	(Calc.)					
1	5036.893	5036.892	3 3 0	4 4 1	110	1A	48	5108.010	5108.008	2 1 2	3 3 1	11	1A	
2	5039.005	5039.005	3 3 1	4 4 0	110	1B	49	5108.209	5108.208	6 1 6	6 2 5	110	1B	
3	5041.173	5041.171	3 2 2	4 4 1	11	1A	50	5108.414	5108.415	8 3 6	9 3 7	11	1B	
3	5041.173	5041.173	10 4 6	11 4 7	11	1B	51	5109.977	5109.976	8 4 5	9 4 6	11	1B	
4	5041.323	5041.324	4 2 2	5 3 3	110	1B	52	5110.036	5110.039	7 3 4	8 3 5	11	1B	
5	5045.072	5045.058	10 3 7	11 3 8	11	1B	53	5113.151	5113.095	2 1 1	3 2 2	110	1B	
5	5045.072	5045.050	3 2 1	4 4 0	11	1B	53	5113.151	5113.152	8 1 7	9 1 8	11	1B	
6	5047.834	5047.837	3 1 3	4 2 2	110	1B	54	5114.910	5114.909	8 2 7	9 2 8	11	1B	
7	5051.361	5051.364	5 1 4	6 3 3	11	1B	55	5115.495	5115.489	7 2 5	8 2 6	11	1B	
8	5051.654	5051.654	11 1 10	12 1 11	11	1B	56	5116.671	5116.704	6 0 6	6 1 5	110	1B	
9	5051.839	5051.839	11 2 10	12 2 11	11	1B	57	5117.943	5117.943	8 5 3	9 5 4	11	1B	
10	5052.412	5052.413	7 0 7	8 1 8	110	1B	58	5118.547	5118.549	4 1 4	5 0 5	110	1B	
11	5052.846	5052.830	7 1 7	8 0 8	110	1B	59	5119.743	5119.743	8 5 4	9 5 5	11	1B	
11	5052.846	5052.848	3 2 2	4 3 1	110	1B	60	5121.903A	5121.903	8 0 8	9 0 9	11	1B	
12	5053.896	5053.886	11 0 11	12 0 12	11	1A	61	5121.975A	5121.971	8 1 8	9 1 9	11	1A	
12	5053.896	5053.894	11 1 11	12 1 12	11	1A	62	5122.711	5122.713	5 2 4	5 4 1	11	1B	
13	5059.967	5059.966	3 2 1	4 3 2	110	1B	63	5124.232	5124.234	7 4 4	8 3 5	5	110	1B
14	5061.025	5061.039	10 5 5	11 5 6	11	1B	64	5124.638	5124.648	2 1 1	3 3 0	11	1B	
15	5064.956	5061.956	10 2 8	11 2 9	11	1B	65	5125.200	5125.200	4 1 1	2 2 2	110	1B	
16	5065.802	5065.802	9 3 6	10 3 7	11	1B	66	5127.344	5127.344	5 1 5	5 2 4	110	1B	
17	5068.724	5068.724	9 4 5	10 4 6	11	1B	67	5128.515	5128.515	7 4 3	8 4 4	11	1B	
18	5068.842	5068.843	5 1 4	6 2 5	110	1B	68	5128.762	5128.763	6 2 5	6 3 4	110	1B	
19	5072.391	5072.389	3 1 3	4 3 2	11	1B	69	5129.007	5129.004	6 3 4	6 4 3	110	1B	
20	5072.444	5072.453	6 2 5	7 1 6	110	1B	70	5130.334	5130.334	7 3 5	8 3 6	11	1B	
21	5072.742	5072.742	10 1 9	11 1 10	11	1B	71	5130.957	5130.957	3 0 3	4 2 2	11	1B	
22	5073.132	5073.132	10 2 9	11 2 10	11	1B	72	5131.727	5131.730	3 0 3	4 1 4	110	1B	
23	5073.899	5073.898	6 0 6	7 1 7	110	1B	73	5132.054	5132.065	1 1 0	2 2 1	110	1B	
24	5074.877	5074.879	6 1 6	7 0 7	110	1B	73	5132.054	5132.042	8 6 2	9 6 3	11	1B	
25	5077.019	5077.019	10 0 10	11 0 11	11	1B	74	5132.204A	5132.187	8 6 3	9 6 4	11	1B	
25	5077.019	5077.024	10 1 10	11 1 11	11	1B	74	5132.204A	5132.205	7 1 6	8 1 7	11	1B	
26	5077.382	5077.385	8 4 5	9 3 6	110	1B	75	5132.623	5132.625	5 3 3	5 4 2	110	1B	
27	5079.496	5079.496	9 2 7	10 2 8	11	1B	76	5134.686	5134.686	7 4 4	8 4 5	11	1B	
28	5081.009	5081.008	4 3 5	5 2 6	11	1B	77	5140.870	5140.869	4 3 2	4 4 1	110	1B	
29	5081.066	5081.064	2 2 1	3 3 0	110	1B	78	5135.505A	5135.507	7 2 6	8 2 7	11	1B	
30	5082.302	5082.302	4 1 3	5 2 4	110	1B	79	5136.382A	5136.380	6 2 4	7 2 5	11	1B	
31	5082.510	5082.511	2 2 0	3 3 1	110	1B	80	5138.215	5138.212	5 3 2	5 4 1	110	1B	
32	5083.928	5083.928	4 0 4	5 2 3	11	1B	81	5139.207	5139.213	5 2 4	5 3 3	110	1B	
33	5086.100	5086.098	9 4 5	10 4 7	11	1B	82	5139.337	5139.336	6 3 3	7 3 4	11	1B	
34	5087.150	5087.149	9 1 7	10 2 8	110	1B	83	5140.334	5140.338	7 1 6	7 2 5	110	1B	
35	5087.300	5087.300	9 3 7	10 3 8	11	1B	84	5141.066	5141.064	3 1 3	4 0 4	110	1B	
36	5090.421	5090.417	2 1 2	3 2 1	110	1B	85	5141.457	5141.456	5 0 5	5 1 4	110	1B	
37	5090.716	5090.713	8 3 5	9 3 6	11	1B	86	5143.602A	5143.603	7 0 7	8 0 8	11	1B	
38	5091.719	5091.717	7 0 7	7 1 6	110	1B	87	5143.773A	5143.773	7 1 7	8 1 8	11	1B	
38	5091.719	5091.720	7 2 6	7 4 3	11	1B	88	5143.982	5143.982	8 7 1	9 7 2	11	1B	
39	5093.303	5093.299	9 1 8	10 1 9	11	1B	88	5143.982	5143.982	8 7 2	9 7 3	11	1B	
40	5094.018	5094.017	9 5 5	10 5 6	11	1B	89	5144.973	5144.971	7 3 2	8 3 3	11	1B	
41	5094.270	5094.274	9 2 8	10 2 9	11	1B	90	5145.608	5145.607	7 5 3	8 5 4	11	1B	
42	5094.468	5094.471	5 0 5	6 1 6	110	1B	91	5146.703	5146.703	4 2 3	4 3 2	110	1B	
43	5096.673	5096.661	5 1 5	6 0 6	110	1B	92	5148.240	5148.235	4 2 2	4 4 1	11	1B	
44	5096.902	5096.902	8 2 6	9 2 7	11	1B	93	5148.561	5148.562	2 0 2	3 1 3	110	1B	
45	5098.633	5098.631	5 2 4	6 1 5	110	1B	94	5150.513A	5150.512	6 1 5	7 1 6	11	1B	
45	5098.633	5098.616	8 4 4	9 4 5	11	1B	95	5151.456	5151.456	3 2 2	3 3 1	110	1B	
46	5099.735A	5099.705	9 0 9	10 0 10	11	1B	96	5151.962						
46	5099.735A	5099.735	9 1 9	10 1 10	11	1B	97	5153.244	5153.244	6 3 4	7 3 5	11	1B	
47	5104.280	5104.278	3 1 2	4 3 1	11	1B	98	5155.703						
							99	5156.156A	5156.158	6 2 5	7 2 6	11	1B	

foreoptics, source, and detector were housed in a vacuum box which was connected to the vacuum spectrometer and the absorption cell.

The H₂¹⁸O sample was supplied by the Bio-Rad Corp. The stated sample mixture ratio of H₂¹⁸O to H₂¹⁶O was 1-to-1 and the H₂¹⁷O impurity was 0.1% of the sample. The path lengths ranged from 8 to 48 m and the sample pressure and temperature were 2.3 Torr and 296 K, respectively. The H₂¹⁸O and H₂¹⁷O line positions were calibrated from the known H₂¹⁶O absorptions (2). The line strength measurements were made from spectra of H₂¹⁶O (3).

DATA ANALYSIS

The identifications of the majority of the H₂¹⁷O absorptions were made by superimposing spectra of H₂¹⁸O/H₂¹⁶O with those of H₂¹⁶O in which the optical density of

TABLE 3—Continued

Line No.	Frequency (CM ⁻¹)		Upper State JK ₁ K ₂	Lower State JK ₁ K ₂	Band	Mot.	Line No.	Frequency (CM ⁻¹)		Upper State JK ₁ K ₂	Lower State JK ₁ K ₂	Band	Mot.	
	(Obs.)	(Calc.)						(Obs.)	(Calc.)					
100	5156.248	5156.244	8 0 8	8 2 7	11	18	148	5188.375N	5188.377	4 1 3	5 1 4	11	18	
101	5157.282	5157.280	6 4 2	7 4 3	11	18	149	5188.785	5188.784	7 3 4	8 2 7	110	18	
102	5157.539	5157.540	3 1 3	3 2 2	110	18	150	5190.055	5190.057	1 1 1	2 0 2	110	18	
103	5157.789	5157.789	7 6 1	8 6 2	11	18	151	5192.552	5192.553	5 3 2	6 2 5	110	18	
104	5157.823	5157.823	7 6 2	8 6 3	11	18	152	5194.759	5194.753	5 0 5	6 0 6	11	17	
105	5158.775	5158.748	3 2 2	4 1 3	110	18	153	5195.341	5195.296	5 4 2	6 4 3	11	17	
106	5158.775	5158.775	6 1 8	8 1 7	11	18	154	5195.341	5195.341	2 0 2	2 1 1	110	18	
106	5159.901	5159.887	5 2 3	6 2 4	11	18	154	5195.501N	5195.502	4 2 2	5 2 3	11	17	
107	5160.121	5160.121	6 1 5	6 2 4	110	18	155	5195.619N	5195.616	5 1 5	6 1 6	11	17	
108	5161.215	5161.218	4 2 2	4 3 1	110	18	156	5196.985	5196.989	5 5 0	6 5 1	11	18	
109	5162.230	5162.231	6 4 3	7 3 4	110	18	157	5197.024	5197.024	5 5 1	6 5 2	11	18	
110	5164.061		7 2 5	7 3 4	110	18	158	5197.934N	5197.999	4 1 3	5 1 4	11	17	
111	5164.240	5164.240	4 0 4	4 1 3	110	18	158	5197.934N	5197.934	4 3 1	5 3 2	11	18	
112	5164.748N	5164.744	6 0 6	6 7 0	7	11	18	159	5198.069	5198.073	6 0 6	6 2 5	11	18
113	5164.879	5164.879	2 1 2	3 0 3	110	18	160	5198.469	5198.472	4 3 1	5 2 4	110	18	
114	5165.140N	5165.145	5 2 3	5 3 2	110	18	161	5198.932	5198.932	7 1 6	7 3 5	11	18	
114	5165.140N	5165.141	6 1 6	7 1 7	11	18	162	5199.235	5199.238	1 0 1	2 2 0	11	18	
115	5165.642	5165.650	1 0 1	2 1 2	110	18	163	5201.675N	5201.675	4 3 2	5 3 3	11	18	
116	5165.796	5165.800	6 2 4	6 4 3	11	18	164	5202.619	5202.616	1 0 1	1 1 0	110	18	
117	5166.915	5166.915	6 0 4	6 3 3	110	18	165	5202.806	5202.806	4 3 1	5 3 2	11	18	
118	5167.737	5167.740	2 1 2	2 2 1	110	18	166	5205.039N	5205.040	4 0 4	5 0 5	11	18	
119	5168.793N	5168.794	5 1 4	6 1 5	11	18	167	5206.367	5206.362	4 2 3	5 2 4	11	18	
120	5168.908	5168.907	5 3 2	6 3 3	11	18	168	5206.655	5206.654	4 1 4	5 1 5	11	18	
121	5168.969	5168.969	2 0 2	3 2 1	11	18	169	5206.827	5206.827	5 5 1	6 5 2	11	17	
122	5169.373	5169.374	3 1 3	3 3 0	11	18	169	5206.827	5206.822	6 1 5	6 3 4	11	18	
122	5169.373	5169.348	7 7 1	8 7 2	11	18	170	5207.222	5207.220	3 3 1	4 2 2	110	18	
122	5169.373	5169.348	7 7 0	8 7 1	11	18	171	5207.448	5207.447	4 1 3	4 3 2	11	18	
123	5171.137	5171.125	7 4 3	8 3 6	110	18	172	5207.520	5207.520	4 3 1	5 2 4	110	17	
124	5171.230	5171.230	6 5 1	7 5 2	11	18	173	5207.747	5207.740	4 3 1	5 3 2	11	17	
125	5171.407	5171.407	6 5 2	7 5 3	11	18	174	5207.949	5207.947	6 1 6	6 1 5	11	18	
126	5172.354	5172.356	4 3 2	5 2 3	110	18	175	5209.377	5209.376	5 1 4	5 3 3	11	18	
127	5174.137	5174.137	5 1 4	5 2 3	110	18	176	5210.178N	5210.178	3 1 2	4 1 3	11	18	
128	5174.196N	5174.197	6 0 6	7 0 7	11	17	177	5210.587	5210.591	6 2 7	8 2 6	11	18	
129	5174.608	5174.603	6 1 6	7 1 7	11	17	178	5210.943N	5210.943	4 0 0	5 4 1	11	18	
130	5177.073	5177.054	5 2 4	6 2 5	11	18	179	5211.165	5211.165	4 4 1	5 4 2	11	18	
130	5177.073	5177.088	5 3 3	6 3 4	11	18	180	5211.379	5211.380	4 3 2	5 3 3	11	17	
131	5177.924	5177.922	7 0 7	7 2 6	11	18	181	5211.708	5211.708	5 4 1	6 3 4	110	18	
132	5178.471	5178.473	8 2 6	8 4 5	11	18	182	5213.265N	5213.265	3 2 1	4 2 2	11	18	
133	5179.941		5 3 2	6 3 3	11	17	183	5214.582N	5214.584	4 0 4	5 0 5	11	17	
134	5182.017		4 1 3	4 2 2	110	18	184	5215.427	5215.428	3 2 1	4 1 4	110	18	
135	5182.674		3 0 3	3 1 2	110	18	185	5215.617	5215.617	4 1 4	5 1 5	11	17	
136	5183.073	5183.068	2 1 1	2 2 0	110	18	186	5215.936	5215.935	5 0 5	5 2 4	11	18	
136	5183.073	5183.077	7 1 7	7 1 6	11	18	187	5216.704	5216.705	4 2 2	5 1 5	110	18	
137	5183.300	5183.301	6 6 1	7 6 2	11	18	188	5219.886N	5219.891	3 1 2	4 1 3	11	17	
137	5183.300	5183.297	6 6 0	7 6 1	11	18	189	5220.916N	5220.917	3 3 0	4 2 3	110	18	
138	5184.341	5184.335	9 2 8	9 2 7	11	18	190	5222.952	5222.950	3 2 1	4 2 2	11	17	
139	5184.494	5184.498	3 1 2	3 2 1	110	18	191	5224.184N	5224.184	3 0 3	4 0 4	11	18	
139	5184.494	5184.490	0 0 0	1 1 1	110	18	192	5225.195	5225.196	3 2 2	4 2 3	11	18	
140	5184.655	5184.655	5 4 1	6 4 2	11	18	193	5225.491	5225.490	3 3 0	4 3 1	11	18	
141	5185.245	5185.252	5 0 5	6 0 6	11	18	194	5226.690N	5226.690	3 3 1	4 3 2	11	18	
142	5185.570N	5185.570	5 4 2	6 4 3	11	18	195	5227.851	5227.851	3 1 3	4 1 4	11	18	
143	5185.721N	5185.722	4 4 2	5 2 3	11	18	196	5229.495	5229.493	3 3 0	4 2 3	110	17	
144	5186.115N	5186.115	5 1 5	6 1 6	11	18	197	5230.202	5230.466N	4 0 4	4 2 3	11	18	
145	5186.210	5186.252	6 3 3	7 2 6	110	18	198	5230.466N	5230.466	5 1 5	5 1 4	11	18	
145	5186.210	5186.204	8 1 7	8 3 6	11	18	199	5233.101N	5233.100	3 2 2	4 2 3	11	17	
146	5186.385N	5186.385	5 2 4	6 2 5	11	17	200	5233.653N	5233.651	3 2 2	4 2 3	11	17	
147	5186.715	5186.712	5 3 3	6 3 4	11	17	201	5233.780	5233.783	3 0 3	4 0 4	11	17	

H₂¹⁷O was identical in the two spectra. Therefore in comparing the two sets of spectra the peak absorptions of H₂¹⁶O lines were much smaller in the H₂¹⁸O/H₂¹⁶O data than in the H₂¹⁶O spectra, while the opposite effect was observed for the H₂¹⁸O lines. The quantum assignments of the measured H₂¹⁸O lines were determined by noting that: (a) the frequency difference between identical transitions of H₂¹⁶O and H₂¹⁸O are approximately 20 and 10 cm⁻¹ for the (011) and (110) bands, respectively, and (b) the peak absorptions for identical transitions in H₂¹⁶O and H₂¹⁸O were approximately equal in the H₂¹⁸O/H₂¹⁶O spectra. The H₂¹⁷O assignments were obtained by noting that their transition frequencies lie in a region midway between the positions of identical transitions in H₂¹⁶O and H₂¹⁸O.

TABLE 3—Continued

Line No.	Frequency (CM ⁻¹)		Upper State JK ₋₁ K ₁	Lower State JK ₋₁ K ₁	Band	Mol.	Line No.	Frequency (CM ⁻¹)		Upper State JK ₋₁ K ₁	Lower State JK ₋₁ K ₁	Band	Mol.
	(Obs.)	(Calc.)						(Obs.)	(Calc.)				
202	5234.070A	5234.070	2 1 1	3 1 2	11	1A	253	5306.836	5306.834	4 2 2	5 0 5	11	1A
203	5236.447N	5236.447	3 3 1	4 3 2	11	17	254	5307.397	5307.401	6 3 4	6 3 3	11	1A
204	5236.967	5236.968	7 2 6	7 2 5	11	18	255	5308.381					
205	5237.471N	5237.472	3 1 3	4 1 4	11	17	256	5308.739	5308.738	3 3 0	3 2 1	11	1A
206	5240.645H	5240.660	3 0 3	3 2 2	11	18	257	5310.631	5310.633	3 0 3	2 2 0	11	1A
206	5240.645H	5240.642	2 2 0	3 2 1	11	18	258	5313.017H	5313.017	3 2 2	3 2 1	11	1A
207	5241.786	5241.786	1 1 0	1 0 1	11	18	259	5315.104	5315.104	2 2 0	3 0 3	11	1A
208	5243.431H	5243.431	2 0 2	3 0 3	11	18	260	5315.318H	5315.318	2 2 1	2 2 0	11	1A
209	5245.345H	5245.345	2 2 1	3 2 2	11	1A	261	5315.990	5315.988	5 0 5	4 1 4	11	1A
210	5247.925	5247.925	2 1 1	2 0 2	11	18	262	5316.356	5316.357	1 1 1	1 1 0	11	17
211	5248.536H	5248.535	2 1 2	3 1 3	11	18	263	5316.743	5316.743	5 3 3	5 3 2	11	1A
212	5250.456H	5250.454	2 2 0	3 2 1	11	17	264	5316.921	5316.923	3 3 1	3 2 2	11	1A
213	5253.097H	5253.097	2 0 2	3 0 3	11	17	265	5317.453H	5317.492	3 3 0	3 2 1	11	17
214	5253.380	5253.380	2 0 2	1 1 1	11	18	265	5317.453H	5317.451	1 1 0	1 1 1	11	18
215	5255.089	5255.091	2 2 1	3 2 2	11	17	266	5317.974	5318.018	8 4 5	8 4 4	11	18
216	5257.066	5257.067	4 1 4	4 1 3	11	18	266	5317.974	5317.965	2 2 0	2 2 1	11	18
217	5258.222	5258.222	2 1 2	3 1 3	11	17	267	5318.287	5318.287	4 0 4	3 2 1	11	1A
217	5258.222	5258.210	6 2 5	5 4 2	11	18	268	5318.893	5318.894	4 3 2	4 2 3	11	1A
218	5258.960	5258.960	3 1 2	3 0 3	11	18	269	5320.280	5320.280	5 1 4	4 0 4	11	18
219	5259.410H	5259.412	1 1 1	2 1 1	11	18	270	5320.680	5320.675	5 1 4	4 2 3	11	18
220	5259.986	5259.985	1 1 1	0 0 0	11	18	271	5321.344	5321.346	2 2 1	1 1 0	11	18
221	5261.829	5261.821	3 1 2	2 2 1	11	18	272	5321.570N	5321.569	4 3 2	4 3 1	11	18
222	5264.096N	5264.095	1 0 1	2 0 2	11	18	273	5321.645	5321.650	3 2 2	3 2 1	11	17
223	5266.372	5266.381	4 1 4	4 1 3	11	17	274	5322.968H	5323.002	5 3 3	5 2 4	11	18
223	5266.372	5266.373	3 2 1	3 1 2	11	18	274	5322.968H	5322.968	3 2 1	3 2 2	11	18
224	5267.124	5267.118	4 2 2	4 1 3	11	18	275	5323.674H	5323.675	3 3 1	3 3 0	11	18
225	5269.578H	5269.578	1 1 1	2 1 2	11	18	276	5324.097H	5324.098	3 3 0	3 3 1	11	18
226	5269.818	5269.817	2 2 0	2 1 1	11	18	277	5324.373H	5324.373	4 3 1	4 3 2	11	18
227	5272.278	5272.275	8 3 6	8 3 5	11	1A	278	5325.145	5325.143	2 1 2	2 2 0	11	17
228	5272.514	5272.514	5 2 3	5 1 4	11	18	279	5325.622	5325.621	7 4 4	7 4 3	11	1A
229	5273.823	5273.824	1 0 1	2 0 2	11	17	280	5326.919	5326.919	5 3 2	5 3 3	11	18
230	5275.232	5275.238	4 1 3	4 0 4	11	18	281	5327.214	5327.214	1 1 0	1 1 1	11	17
231	5276.575	5276.569	3 0 3	2 1 2	11	18	282	5327.747H	5327.748	2 2 0	2 2 1	11	17
232	5277.190	5277.196	6 3 3	7 1 6	11	18	283	5327.857	5327.856	2 2 0	1 1 1	11	18
233	5278.796H	5278.796	3 1 3	3 1 2	11	18	284	5327.964	5327.964	2 1 1	2 1 2	11	18
234	5279.312H	5279.311	1 1 1	2 1 2	11	17	285	5328.600	5328.602	6 2 5	6 1 6	11	18
235	5284.382	5284.345	6 2 4	6 1 5	11	18	286	5329.691H	5329.690	6 4 5	6 4 2	11	1A
235	5284.382	5284.380	2 2 1	2 1 2	11	18	287	5331.380	5331.380	4 3 2	4 3 1	11	17
236	5286.712	5286.712	0 0 0	1 0 1	11	18	287	5331.380	5331.390	3 3 0	4 1 3	11	18
237	5288.656H	5288.670	3 1 3	3 1 2	11	17	288	5331.612H	5331.612	5 4 2	5 4 1	11	18
238	5290.199	5290.245	7 3 4	7 2 5	11	18	289	5331.947	5331.948	6 4 2	6 4 3	11	1A
239	5291.611	5291.613	6 3 3	6 2 4	11	18	290	5332.270	5332.269	5 4 1	5 4 2	11	18
239	5291.611	5291.584	5 2 3	4 3 2	11	18	290	5332.270	5332.260	4 2 2	4 2 3	11	18
240	5292.372	5292.370	3 1 3	2 0 2	11	18	291	5332.540	5332.541	3 2 1	3 2 2	11	17
241	5292.488	5292.490	7 3 5	7 3 4	11	18	292	5332.666	5332.667	6 0 6	5 1 5	11	18
242	5295.250	5295.249	5 1 4	5 0 5	11	18	293	5333.064	5333.036	4 4 1	4 4 0	11	18
243	5295.314H	5295.314	2 1 2	2 1 1	11	18	293	5333.064	5333.091	4 4 0	4 4 1	11	18
244	5296.487	5296.487	0 0 0	1 0 1	11	17	293	5333.064	5333.058	7 4 3	7 4 4	11	18
245	5297.476	5297.475	4 0 4	3 1 3	11	18	294	5333.280	5333.281	7 4 5	7 3 4	11	18
246	5297.848	5297.847	5 3 2	5 2 3	11	19	295	5333.500	5333.465	3 3 1	3 3 0	11	17
247	5298.182	5298.181	4 3 1	4 2 2	11	18	295	5333.500	5333.506	6 3 3	6 3 4	11	18
248	5302.165	5302.165	4 2 3	4 1 4	11	18	296	5333.897H	5333.897	3 3 0	3 3 1	11	17
249	5305.314	5305.303	4 3 1	5 1 4	11	18	297	5334.024	5334.023	1 0 1	0 0 0	11	18
249	5305.314	5305.335	9 4 6	9 4 5	11	18	298	5334.107H	5334.110	4 3 1	4 3 2	11	17
250	5306.092	5306.091	4 2 3	4 2 2	11	18	299	5334.622	5334.614	6 1 6	5 0 5	11	18
251	5306.259	5306.258	4 1 4	3 0 3	11	18	300	5334.974	5334.976	9 5 5	9 5 4	11	18
252	5306.544H	5306.544	1 1 1	1 1 0	11	18	301	5337.199	5337.198	8 4 4	8 4 5	11	18

The H₂¹⁸O line strength measurements were made from previously recorded spectra of H₂¹⁶O (3) in which only the strongest lines of H₂¹⁸O were observed. The line strengths were determined by the method of equivalent widths. The procedure for determining line strengths of water vapor from the equivalent widths has been discussed in detail in earlier reports (3, 4).

RESULTS

Values of the ground state levels of H₂¹⁸O and H₂¹⁷O are presented in Table 1. The Table lists the rotational quantum numbers ($JK_{-1}K_1$) and three sets of values for H₂¹⁸O

TABLE 3—Continued

Line No.	Frequency (CM ⁻¹)		Upper State JK ₁ K ₂	Lower State JK ₁ K ₂	Band	Mol.	Line No.	Frequency (CM ⁻¹)		Upper State JK ₁ K ₂	Lower State JK ₁ K ₂	Band	Mol.
	(Obs.)	(Calc.)						(Obs.)	(Calc.)				
302	5339.242						351	5362.068					
303	5340.181	5340.179	8 5 3	8 5 4	11 18		352	5362.547N	5362.541	3 1 3	2 1 2	11 17	
304	5340.776	5340.780	7 5 3	7 5 2	11 18		353	5363.201	5363.200	3 3 0	3 0 3	110 18	
305	5341.186	5341.187	7 5 2	7 5 3	11 18		354	5364.757	5364.757	5 1 4	5 1 5	11 18	
306	5341.606H	5341.607	5 4 2	5 4 1	11 17		355	5365.399N	5365.396	3 0 3	2 0 2	11 17	
307	5341.854H	5341.855	4 2 2	4 2 3	11 17		356	5366.060H	5366.061	3 3 0	2 2 1	110 18	
308	5342.286	5342.286	6 5 2	6 5 1	11 18		357	5366.899	5366.896	3 3 1	2 2 0	110 18	
309	5342.371	5342.357	6 5 1	6 5 2	11 18		358	5367.481N	5367.479	3 2 2	3 0 3	11 18	
310	5342.893N	5342.880	4 4 1	4 4 0	11 17		359	5369.118	5369.119	7 2 5	7 2 6	11 18	
310	5342.893N	5342.902	4 4 0	4 4 1	11 17		360	5369.180	5369.182	10 1 10	9 0 9	110 18	
311	5343.695	5343.695	6 4 2	6 3 3	110 18		361	5390.192N	5390.192	3 1 2	2 1 1	11 18	
312	5343.834H	5343.834	1 0 1	0 0 0	11 17		362	5390.302N	5390.340	3 2 2	2 2 1	11 18	
313	5344.596	5344.604	7 3 4	7 3 5	11 18		362	5390.302N	5390.302	4 1 4	3 1 3	11 18	
314	5346.483	5346.477	6 1 5	5 3 2	11 18		363	5391.410	5391.408	4 3 1	4 0 4	110 18	
315	5346.955	5346.955	5 2 3	5 2 4	11 18		364	5392.749N	5392.749	4 0 4	3 0 3	11 18	
316	5347.175	5347.189	6 1 5	5 2 4	110 18		365	5392.941N	5392.941	3 2 1	2 2 0	11 18	
317	5348.070	5348.066	7 0 7	6 1 6	110 18		366	5394.500					
318	5348.891	5348.895	7 1 7	6 0 6	110 18		367	5394.785N	5394.786	3 3 0	2 2 1	110 17	
319	5349.775	5349.776	9 2 6	8 3 5	11 18		368	5399.509					
320	5351.362	5351.363	5 4 1	5 3 2	110 18		369	5399.589					
321	5351.611						370	5398.466					
322	5351.681	5351.626	8 6 3	8 6 2	11 18		371	5398.945N	5398.944	3 2 2	2 2 1	11 17	
322	5351.681	5351.676	8 6 2	8 6 3	11 18		372	5399.050					
323	5353.352	5353.353	2 1 2	1 1 1	11 18		373	5399.320	5399.318	4 2 3	4 0 4	11 18	
324	5354.277	5354.276	6 6 1	6 6 0	11 18		374	5400.038N	5400.034	3 1 2	2 1 1	11 17	
324	5354.277	5354.279	6 6 0	6 6 1	11 18		375	5400.358	5400.353	4 2 2	3 1 3	110 18	
325	5355.489	5355.474	8 4 5	8 3 6	110 18		376	5401.399					
326	5355.751	5355.757	4 4 0	4 3 1	110 18		377	5402.593	5402.593	3 2 1	2 2 0	11 17	
327	5356.013H	5356.013	2 0 2	1 0 1	11 18		378	5402.705N	5402.704	4 0 4	3 0 3	11 17	
328	5356.398	5356.401	6 4 3	6 3 4	110 18		379	5402.928					
328	5356.398	5356.366	5 4 2	5 3 3	110 18		380	5403.282					
329	5357.260						381	5403.461H	5403.460	5 2 4	5 0 5	11 18	
330	5358.800	5358.799	7 4 4	7 3 5	110 18		382	5404.450					
331	5360.268	5360.267	3 2 1	2 1 2	110 18		383	5406.716N	5406.715	4 3 2	3 2 1	110 18	
332	5362.262		9 7 3	9 7 2	11 18		384	5406.860	5406.861	6 1 5	6 1 6	11 18	
333	5362.831	5362.828	8 1 8	7 0 7	110 18		385	5407.632N	5407.632	5 1 5	4 1 4	11 18	
334	5362.908N	5362.909	4 1 3	4 1 4	11 18		386	5407.885N	5407.884	4 3 1	3 2 2	110 18	
335	5363.712	5363.712	8 7 2	8 7 1	11 18		387	5408.312					
335	5363.712	5363.712	8 7 1	8 7 2	11 18		388	5408.671N	5408.671	5 0 5	4 0 4	11 18	
336	5364.930N	5364.930	2 1 1	1 1 0	11 18		389	5413.195	5413.147	10 3 7	10 3 8	11 18	
337	5365.011	5365.036	7 7 0	7 7 1	11 18		389	5413.195	5413.195	8 2 6	8 2 7	11 18	
337	5365.011	5365.036	7 7 1	7 7 0	11 18		390	5413.850	5413.854	4 1 3	3 1 2	11 18	
337	5365.011	5365.007	5 2 4	4 1 3	110 18		391	5414.092					
338	5365.862H	5365.862	2 0 2	1 0 1	11 17		392	5415.792	5415.794	4 2 3	3 2 2	11 18	
339	5366.384N	5366.383	6 2 4	6 2 5	11 18		393	5417.143H	5417.176	4 3 1	3 2 2	110 17	
340	5368.800	5368.802	8 3 5	8 3 6	11 18		394	5417.143H	5417.143	5 1 5	4 1 6	11 17	
341	5370.345	5370.341	7 1 6	6 2 6	110 18		394	5417.559N	5417.561	5 1 5	4 1 4	11 17	
342	5372.568	5372.568	4 1 3	4 1 4	11 17		395	5417.904	5417.904	6 2 5	6 0 6	11 18	
343	5372.690N	5372.690	3 1 3	2 1 2	11 18		396	5418.853N	5418.859	5 0 5	4 0 4	11 17	
344	5374.589		8 8 1	8 8 0	11 18		397	5418.956	5418.959	5 3 2	5 0 5	110 18	
344	5374.589		8 8 0	8 8 1	11 18		398	5420.080N	5420.081	4 2 2	3 2 1	11 18	
345	5374.751H	5374.751	2 1 1	1 1 0	11 17		399	5420.177H	5420.177	4 3 2	3 3 1	11 18	
346	5375.491N	5375.490	5 0 3	2 0 2	11 18		400	5421.357H	5421.358	4 3 1	3 3 0	11 18	
347	5375.788	5375.787	6 2 5	5 1 4	110 18		401	5422.712	5422.711	5 3 3	4 2 2	110 18	
348	5376.090	5376.090	9 0 9	8 1 8	110 18		402	5423.766N	5423.766	4 1 3	3 1 2	11 17	
349	5376.304	5376.304	9 1 9	8 0 8	110 18		402	5423.766N	5423.766	4 2 3	3 2 2	11 17	
350	5380.175	5380.175	2 2 1	2 0 2	11 18		403	5423.908N	5423.910	6 1 6	5 1 5	11 18	

and two sets for H₂¹⁷O. The three sets of H₂¹⁸O values are from (a) the present work, (b) those given by Garing and McClatchey (5), and (c) the values obtained in an analysis of the (020) band by Toth and Margolis (1). The H₂¹⁷O levels are from (a) the present study and (b) Garing and McClatchey (5). Values given for this work were determined by the method of combination differences using the measured transition frequencies in the (011) and (110) bands for H₂¹⁸O and in the (011) band for H₂¹⁷O. This procedure requires that values of the levels (101), (110), (111), are known from other sources or they are derived from the measured positions by obtaining an initial set of upper state levels and iterating between upper and lower state levels.

TABLE 3—Continued

Line No.	Frequency (CM ⁻¹)		Upper State JK ₁ K ₂	Lower State JK ₁ K ₂	Band	Mol.	Line No.	Frequency (CM ⁻¹)		Upper State JK ₁ K ₂	Lower State JK ₁ K ₂	Band	Mol.
	(Obs.)	(Calc.)						(Obs.)	(Calc.)				
404	5424.111	5424.112	5 3 3	5 1 4	11	18	457	5477.334	5477.334	6 4 3	5 4 2	11	18
405	5424.479N	5424.479	6 0 6	5 0 5	11	18	458	5477.603	5477.602	5 4 1	4 3 2	11	18
406	5424.833	5424.831	6 3 4	6 1 5	11	18	459	5478.190N	5478.190	6 4 2	5 4 1	11	18
407	5427.698	5427.713	7 1 6	7 1 7	11	18	460	5478.558		9 1 9	8 1 8	11	17
407	5427.698	5427.686	2 2 0	1 0 1	11	18	461	5478.681	5478.681	6 3 3	5 2 4	11	18
408	5428.886	5428.886	5 2 4	4 2 3	11	18	462	5479.841	5479.835	4 3 1	4 1 4	11	18
409	5429.858	5429.854	4 2 2	3 2 1	11	17	463	5481.178	5481.177	4 3 2	3 0 3	11	18
410	5429.980	5429.990	4 3 2	3 3 1	11	17	464	5481.628		6 2 4	5 2 3	11	17
411	5430.353	5430.350	7 3 5	7 1 6	11	18	465	5482.191N	5482.191	10 1 10	9 1 9	11	18
412	5431.145N	5431.148	4 3 1	3 3 0	11	17	466	5482.213N	5482.218	10 0 10	9 0 9	11	18
413	5433.643	5433.642	7 2 6	7 0 7	11	18	467	5482.742N		6 3 3	5 3 2	11	17
414	5433.875	5433.880	6 1 6	5 1 5	11	17	468	5484.222	5484.221	8 2 7	7 2 6	11	18
415	5434.475N	5434.474	6 0 6	5 0 5	11	17	469	5486.660N	5486.660	7 3 5	6 3 4	11	18
416	5435.171	5435.170	5 1 4	4 1 3	11	18	470	5488.720N	5488.720	6 5 2	5 5 1	11	18
417	5438.556N	5438.556	5 2 4	4 2 3	11	17	471	5488.752N	5488.752	6 5 1	5 5 0	11	18
418	5439.426N	5439.426	7 1 7	6 1 6	11	18	472	5491.448	5491.447	5 4 2	5 2 3	11	18
419	5439.669N	5439.668	7 0 7	6 0 6	11	18	473	5494.476N	5494.480	7 2 5	6 2 4	11	18
420	5440.118	5440.113	8 3 6	8 1 7	11	18	474	5494.542N	5494.543	4 2 2	3 0 3	11	18
421	5443.182N	5443.182	5 3 3	4 3 2	11	18	475	5495.287N	5495.288	11 1 11	10 1 10	11	18
422	5444.069						476	5496.057	5496.056	6 4 3	5 3 2	11	18
423	5444.384	5444.385	5 3 2	4 2 3	11	18	477	5496.386N		8 1 7	7 1 6	11	17
424	5446.663N	5446.664	5 3 2	4 3 1	11	18	478	5496.491N		7 3 5	6 3 4	11	17
425	5446.812N	5446.813	5 3 2	4 3 1	11	18	479	5498.595		6 5 1	5 5 0	11	17
426	5446.916	5446.916	8 1 7	8 1 8	11	18	480	5498.762	5498.761	7 3 4	6 3 3	11	18
427	5448.588	5448.587	6 2 5	5 2 4	11	18	481	5500.289N	5500.289	6 2 4	5 1 5	11	18
428	5449.401	5449.401	7 1 7	6 1 6	11	17	482	5500.626N	5500.628	9 2 8	8 2 7	11	18
428	5449.401	5449.401	8 3 6	7 2 5	11	18	483	5501.534	5501.533	9 1 8	8 1 7	11	18
429	5449.667	5449.667	7 0 7	6 0 6	11	17	484	5501.706	5501.707	6 4 2	5 3 3	11	18
430	5451.546						485	5502.767	5502.767	7 4 3	6 4 2	11	18
431	5451.749	5451.753	3 3 1	2 0 2	11	18	486	5506.839N	5506.840	6 3 6	7 3 5	11	18
432	5452.824	5452.828	9 3 7	9 1 8	11	18	487	5511.407	5511.404	3 3 0	2 1 1	11	18
433	5452.978	5452.980	5 3 3	4 3 2	11	17	488	5511.911	5511.907	7 5 3	6 5 2	11	18
434	5453.846	5453.846	6 1 5	5 1 4	11	18	489	5512.065	5512.066	7 5 2	6 5 1	11	18
435	5453.959	5453.960	5 4 2	4 4 1	11	18	490	5512.956	5512.956	7 4 4	6 3 3	11	18
436	5454.170	5454.144	4 4 1	3 3 0	11	18	491	5514.656N	5514.656	8 2 6	7 2 5	11	18
437	5454.280	5454.283	8 1 8	7 1 7	11	18	492	5516.019	5516.019	10 2 9	9 2 8	11	18
438	5454.376	5454.365	4 4 0	3 3 1	11	18	493	5516.520	5516.520	10 1 9	9 1 8	11	18
438	5454.376	5454.379	8 0 8	7 0 7	11	18	494	5517.115	5517.116	4 4 0	4 2 3	11	18
439	5457.799	5457.798	3 2 1	2 0 2	11	18	495	5517.633	5517.630	8 4 5	7 3 4	11	18
440	5462.351	5462.346	9 4 6	9 2 7	11	18	496	5519.812	5519.812	13 1 13	12 1 12	11	18
441	5463.792	5463.791	5 4 2	4 4 1	11	17	497	5519.812	5519.812	13 0 13	12 0 12	11	18
442	5464.262	5464.232	8 4 5	8 2 6	11	18	498	5520.247	5520.248	7 4 6	6 2 5	11	18
443	5464.396N	5464.396	8 0 8	7 0 7	11	17	499	5522.559	5522.561	6 4 5	7 4 4	11	18
444	5464.624	5464.625	3 3 0	3 1 3	11	18	500	5524.071	5524.065	7 6 1	6 6 0	11	18
445	5464.724	5464.729	9 1 8	7 1 9	11	18	501	5524.071	5524.065	7 6 2	6 6 1	11	18
446	5465.412	5465.413	6 3 4	5 3 3	11	18	502	5524.852		8 2 6	7 2 5	11	17
447	5466.283	5466.287	9 2 8	9 0 9	11	18	503	5525.880	5525.880	9 3 7	8 3 6	11	18
448	5466.972	5466.971	7 2 6	6 2 5	11	18	504	5526.991	5526.991	3 3 1	2 1 2	11	18
449	5468.537N	5468.537	9 1 9	8 1 8	11	18	505	5527.451	5527.451	7 4 3	6 3 4	11	18
450	5468.572N	5468.572	9 0 9	8 0 8	11	18	506	5528.132N	5528.133	6 4 4	7 4 3	11	18
451	5470.520	5470.519	7 1 6	6 1 5	11	18	507	5530.783	5530.780	4 3 1	3 1 2	11	18
452	5470.871	5470.869	9 1 4	7 2 5	11	18	508	5530.903	5530.903	11 2 10	10 2 9	11	18
453	5471.674N	5471.677	6 2 4	5 2 3	11	18	509	5530.958	5530.958	8 3 5	7 3 4	11	18
454	5473.160N	5473.161	6 3 3	5 3 2	11	18	510	5531.138	5531.138	11 1 10	10 1 9	11	18
455	5476.260	5476.260	5 4 2	4 3 1	11	18							
456	5476.944N		7 2 6	6 2 5	11	17							

Table 2 lists values of the levels obtained from the present data on the (011), (110) states of H₂¹⁸O and H₂¹⁷O. They were obtained from the ground state levels given for this work in Table 1 and the measured line frequencies. The (011) band is much stronger than the (110) band for respective isotopic species and therefore the majority of the H₂¹⁸O and H₂¹⁷O observations were of lines in the (011) band. The two entries given in Table 2 for the (110) state of H₂¹⁷O were derived from transitions for which the line strengths are enhanced due to Coriolis mixing of the (110) state levels with levels in the (011) state. In fact many levels in the (011) state are affected by Coriolis interactions with levels in the (110) state for either molecule. The magnitude of the mixing which these levels exhibit in H₂¹⁶O is presented in a paper by Camy-Peyret and Flaud (6).

TABLE 3—Continued

Line No.	Frequency (CM ⁻¹)		Upper State		Lower State		Band No.
	(Obs.)	(Calc.)	JK ₋₁ K ₁				
509	5531.376						
510	5532.295	5532.295	9 2 7	8 2 6	11 18		
511	5532.536	5532.531	6 4 2	6 2 5	11 18		
512	5533.553	5533.545	6 3 3	6 1 6	11 18		
513	5534.778	5534.778	8 5 4	7 5 3	11 18		
514	5535.350	5535.352	8 5 3	7 5 2	11 18		
515	5539.891	5539.891	5 2 3	4 0 4	11 18		
516	5543.783	5543.783	10 3 8	9 3 7	11 18		
517	5543.915	5543.917	9 4 6	8 4 5	11 18		
518	5545.193	5545.193	12 2 11	11 2 10	11 18		
519	5545.345	5545.345	12 1 11	11 1 10	11 18		
520	5546.935	5546.927	8 6 3	7 6 2	11 18		
520	5546.935	5546.936	8 6 2	7 6 1	11 18		
521	5546.183	5546.183	10 2 8	9 2 7	11 18		
522	5552.714	5552.713	5 3 2	4 1 3	11 18		
523	5552.856	5552.856	9 3 6	8 3 5	11 18		
524	5553.954	5553.954	9 4 5	8 4 4	11 18		
525	5557.213	5557.212	9 5 5	8 5 4	11 18		
526	5558.775	5558.775	9 5 4	8 5 3	11 18		
527	5559.012	5559.012	13 2 12	12 2 11	11 18		
528	5559.130	5559.130	13 1 12	12 1 11	11 18		
529	5559.400	5559.400	8 7 2	7 7 1	11 18		
529	5559.400	5559.400	8 7 1	7 7 0	11 18		
530	5560.608	5560.608	11 3 9	10 3 8	11 18		
531	5560.703	5560.704	4 3 2	3 1 3	11 18		
532	5563.253	5563.253	11 2 9	10 2 8	11 18		
533	5564.210	5564.210	10 4 7	9 4 6	11 18		
534	5569.410	5569.410	9 6 4	8 6 3	11 18		
535	5569.530	5569.530	9 6 3	8 6 2	11 18		
536	5573.638	5573.638	10 3 7	9 3 6	11 18		
537	5578.983	5578.983	10 4 6	9 4 5	11 18		
538	5580.535	5580.530	6 3 3	5 1 4	11 18		
539	5582.595	5582.595	10 5 5	9 5 4	11 18		
540	5583.369	5583.369	11 4 8	10 4 7	11 18		
541	5583.444						
542	5592.789	5592.789	6 2 4	5 0 5	11 18		
543	5598.644	5598.644	5 3 3	4 1 4	11 18		
544	5604.938	5604.937	4 4 0	3 2 1	11 18		
545	5616.184	5616.191	7 3 4	6 1 5	11 18		
546	5620.812						
547	5637.826	5637.825	6 4 2	5 2 3	11 18		
548	5637.984	5637.985	5 4 2	4 2 3	11 18		

The magnitude of the mixing of the levels in H₂¹⁸O and H₂¹⁷O for these states is comparable to that in H₂¹⁶O.

Table 3 is a listing of the line numbers, measured and calculated line center frequencies, upper and lower state rotational quantum numbers, band, and isotopic species. The line numbers pertain to the numbering of the absorptions shown in Fig. 1. Absorptions not numbered in Fig. 1 are due to H₂¹⁶O. The calculated frequencies given in Table 3 were determined from the upper state levels given in Table 2 and the ground state values, of this study, given in Table 1. The measured frequencies denoted by an "N" in Table 3 refer to measurements taken from very high resolution interferometric spectra of H₂¹⁶O (2). A few of the measured lines given in Table 3 do not have quantum assignments (for example, line numbers 96, 98, 110, 133, etc.) and they may be due to yet unidentified transitions in the (011) and (110) bands of H₂¹⁸O and H₂¹⁷O or "hot" band transitions of H₂¹⁸O, mainly in the (021)-(010) band. Also quantum assignments were made for a few lines for which the calculated frequencies were not determined (for example, line numbers 3, 14, 88, 110, etc.) because either or both the upper and lower state levels could not be obtained from the data.

TABLE 4

Positions (cm⁻¹) of H₂¹⁸O/H₂¹⁷O and H₂¹⁶O lines which were blended in the spectral data.

Position	Upper Level		Lower Level		Band	Mol.	Position	Upper Level		Lower Level		Band	Mol.
	J	K ₋₁ K ₁	J	K ₋₁ K ₁				J	K ₋₁ K ₁	J	K ₋₁ K ₁		
5113.827	4	0 4	5	1 5	110	18	5338.762	3	2 2	2	1 1	110	18
5113.872	9	2 8	10	2 9	011	16	5338.755	2	2 0	2	2 1	011	16
5125.731	5	1 5	5	3 2	011	13	5343.413	3	1 2	3	1 3	011	13
5125.709	2	1 1	3	2 2	110	16	5343.419	5	5 1	5	5 0	011	18
							5343.423	5	5 0	5	5 1	011	18
5134.773	9	1 9	9	1 8	011	18	5343.403	3	2 1	3	2 2	011	16
5134.779	8	2 7	9	2 8	011	16							
							5353.083	7	6 2	7	6 1	011	18
5156.951	3	2 1	3	3 0	110	18	5353.090	7	6 1	7	6 2	011	18
5157.007	4	1 4	4	2 3	110	16	5353.110	4	2 3	3	1 2	110	18
							5353.072	5	4 1	5	4 2	011	16
5159.981	6	4 3	7	4 4	011	18							
5159.937	6	3 3	7	3 4	011	16	5357.159	4	4 1	4	3 2	110	18
							5357.136	8	4 4	8	4 5	011	16
5243.834	2	1 1	3	1 2	011	17							
5243.889	3	2 2	4	2 3	011	16	5363.190	2	1 2	1	1 1	011	17
							5363.143	7	3 4	7	3 5	011	16
5246.292	2	0 2	2	2 1	011	13							
5246.303	3	3 0	4	3 1	011	16	5399.589	4	1 4	3	1 3	011	17
							5399.632	4	3 2	5	1 5	011	16
5261.519	6	2 5	6	2 4	011	18							
5261.501	2	2 0	3	2 1	011	16	5427.469	4	3 2	4	1 3	011	18
							5427.496	6	1 5	6	1 6	011	16
5269.181	1	1 0	2	1 1	011	17							
5269.136	2	1 2	3	1 3	011	16	5433.097	3	3 1	3	1 2	011	18
							5433.119	8	2 6	8	2 7	011	16
5277.461	2	1 2	1	0 1	110	18							
5277.462	4	1 4	4	1 3	011	16	5434.884	6	3 4	5	2 3	110	18
							5434.922	4	1 3	3	1 2	011	16
5282.348	5	2 4	5	2 3	011	18							
5282.355	4	2 2	4	1 3	110	16	5447.046	5	2 3	4	1 4	110	18
							5447.056	5	3 3	5	1 4	011	16
5291.714	4	1 3	3	2 2	110	18							
5291.706	2	1 2	1	0 1	110	16	5449.902	8	2 7	8	0 8	011	13
							5449.863	2	2 0	1	0 1	011	16
5314.175	4	2 3	4	2 2	011	17							
5314.148	7	3 5	7	3 4	011	16	5486.220	8	1 7	7	1 6	011	18
							5486.214	6	3 4	5	3 3	011	16
5314.594	5	2 4	5	1 5	110	18							
5314.518	5	3 2	5	2 3	110	16	5507.838	12	0 12	11	0 11	011	13
							5507.831	12	1 12	11	1 11	011	18
5326.610	5	3 3	5	3 2	011	17	5507.828	8	1 7	7	1 6	011	16
5326.646	3	3 0	3	2 1	110	16							
5337.705	2	1 1	2	1 2	011	17							
5337.702	5	3 3	5	3 2	011	16							
5338.562	8	5 4	8	5 3	011	18							
5338.517	4	0 4	3	2 1	011	16							
5338.534	5	3 3	5	2 4	110	16							

Several H₂¹⁸O and H₂¹⁷O lines were blended with H₂¹⁶O lines in the spectra to an extent that measurements of their line positions could not be made with any accuracy. Table 4 is a listing of these lines along with their quantum and band assignments. The H₂¹⁶O frequencies are those given in Ref. (2) and the H₂¹⁸O and H₂¹⁷O values were calculated from the ground state levels, of this work, given in Table 1 and the upper state levels listed in Table 2.

Table 5 lists the measured line strengths of H₂¹⁸O. For comparison purposes, those transitions for which measurements of H₂¹⁶O line strengths were made (3) are also given in terms of the product with the ¹⁸O/¹⁶O ratio; 0.00209. Inspection of the H₂¹⁸O and H₂¹⁶O results indicate that, to good approximation, H₂¹⁸O line strengths can be calcu-

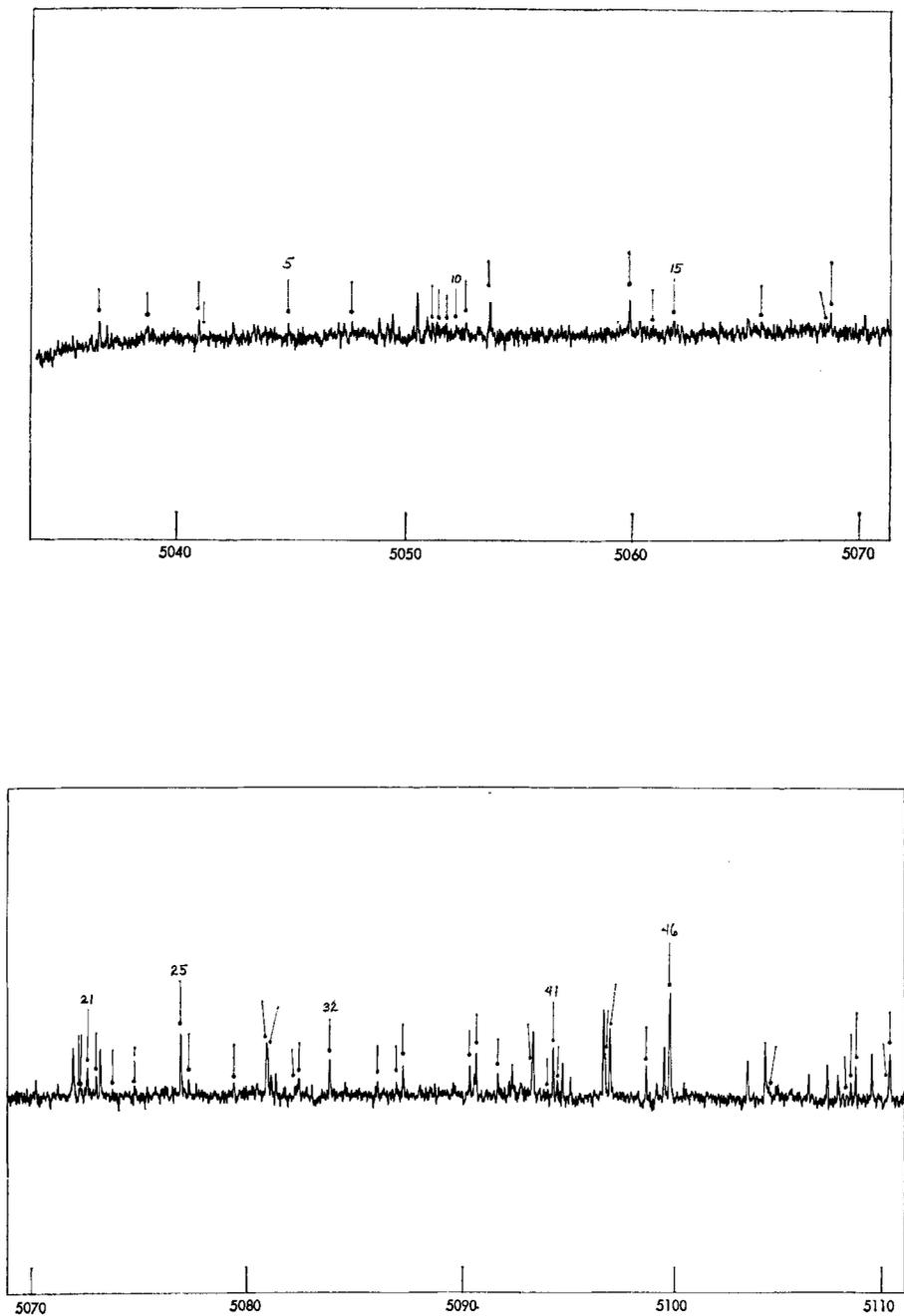


FIG. 1. Scan of the 5030 to 5640 cm^{-1} region of H_2^{18}O , H_2^{16}O , and H_2^{17}O . The sample pressure was 2.3 Torr and the path was 48 m. The isotopic ratio of the water vapor sample was 50, 50, and 0.1% for H_2^{18}O , H_2^{16}O , and H_2^{17}O , respectively. The stroke marks denote the H_2^{18}O and H_2^{17}O lines and only these are numbered.

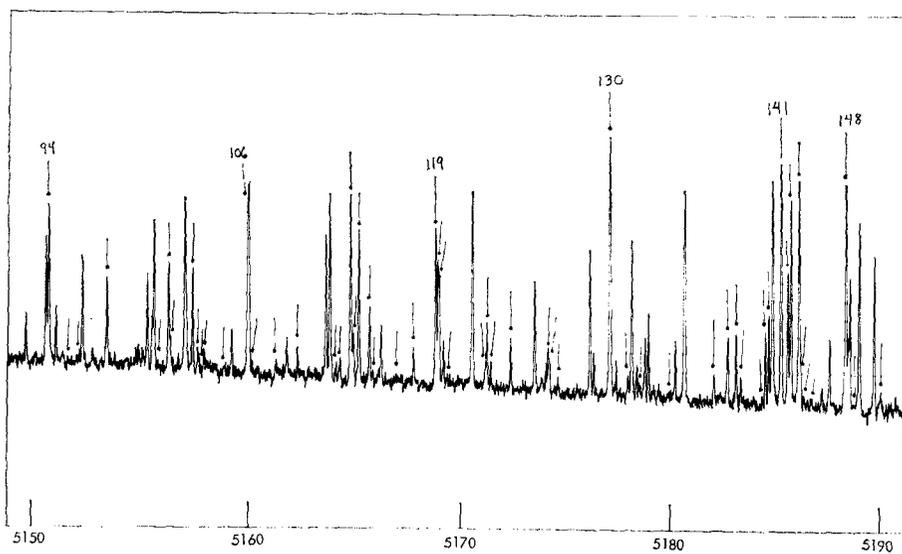
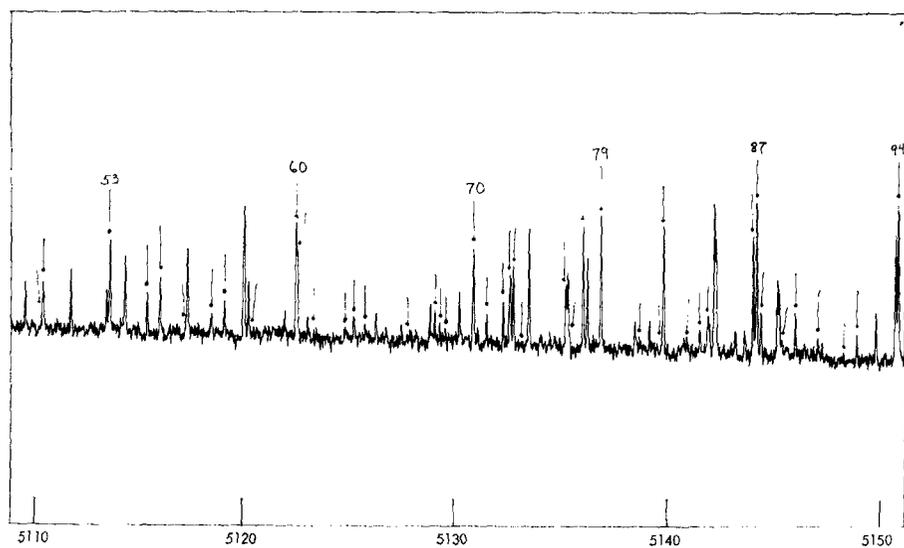


FIG. 1—Continued

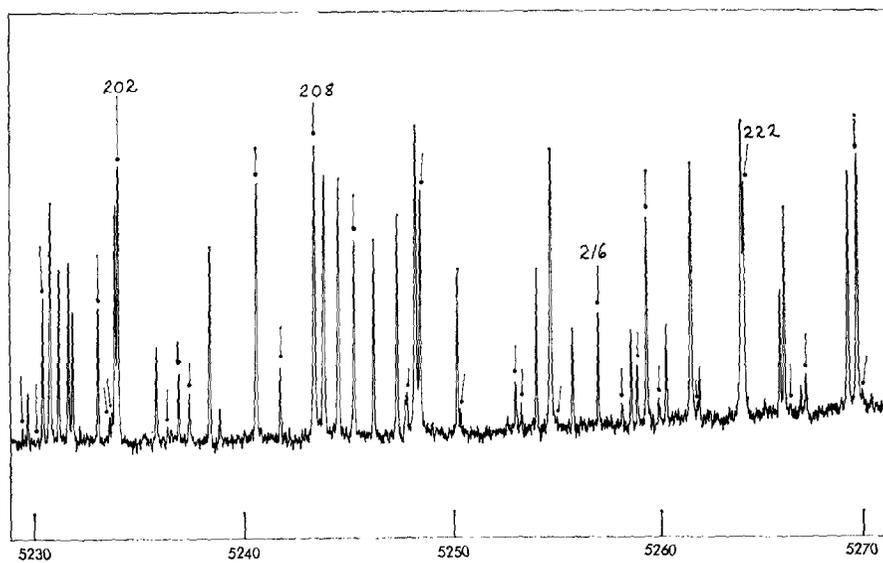
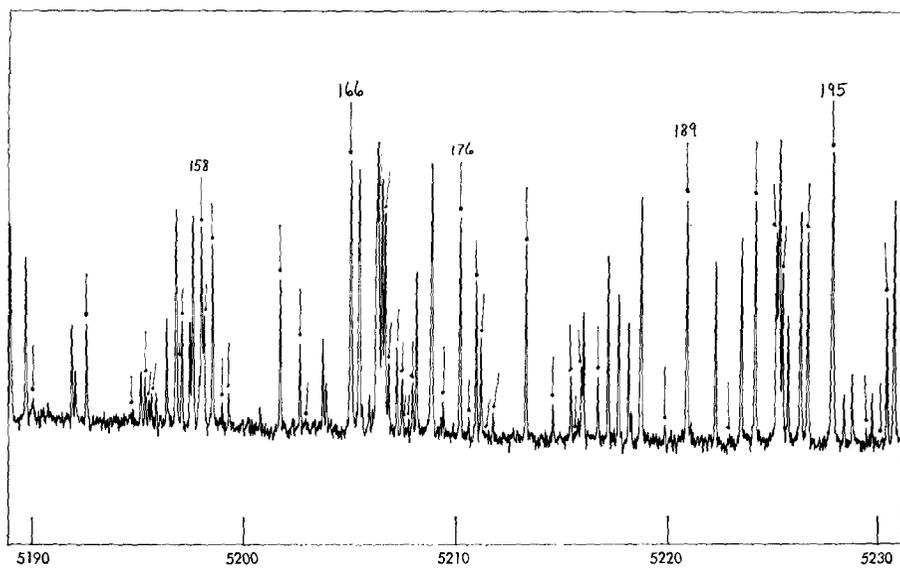


FIG. 1—Continued

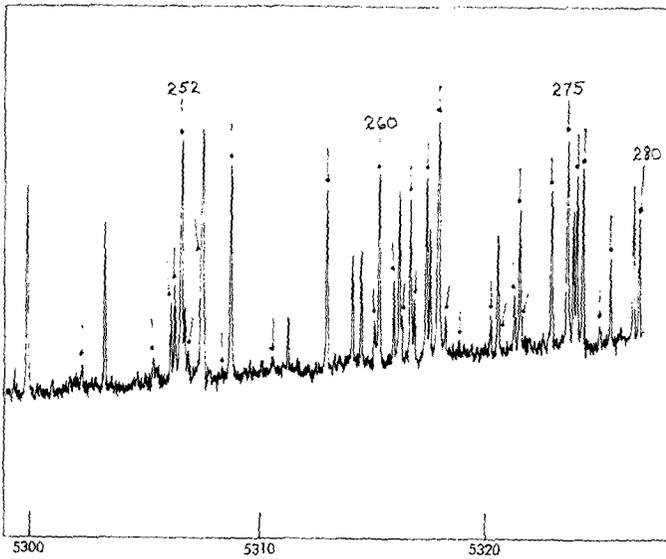
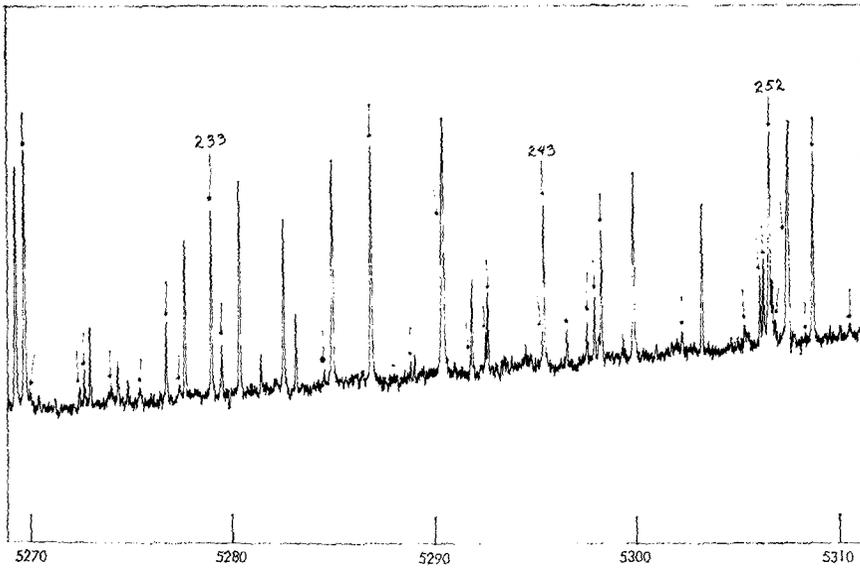


FIG. 1—Continued

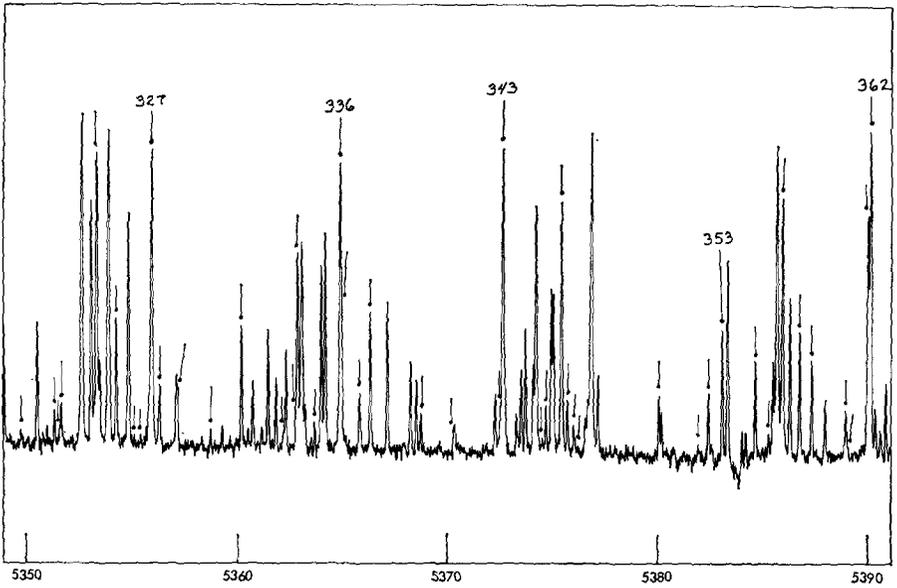
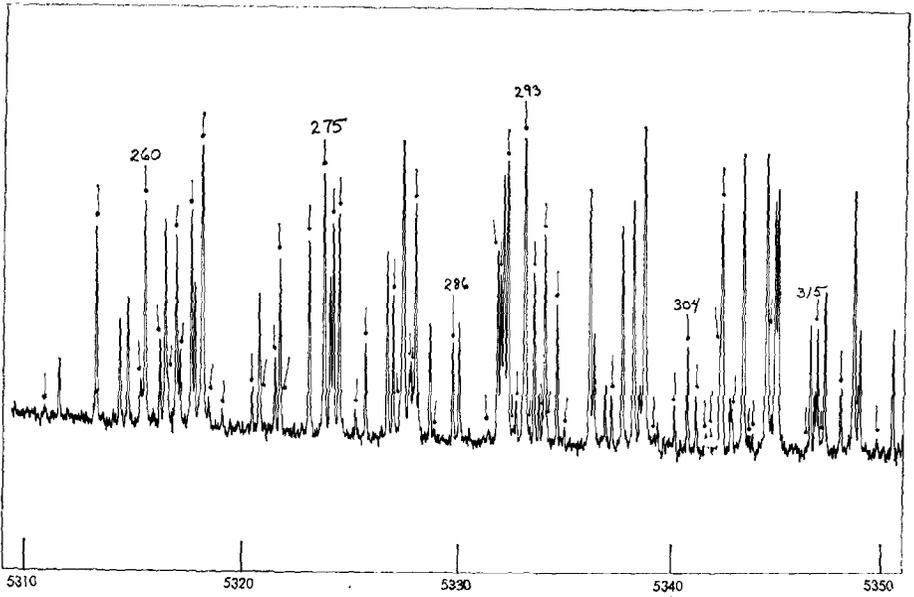


FIG. 1—Continued

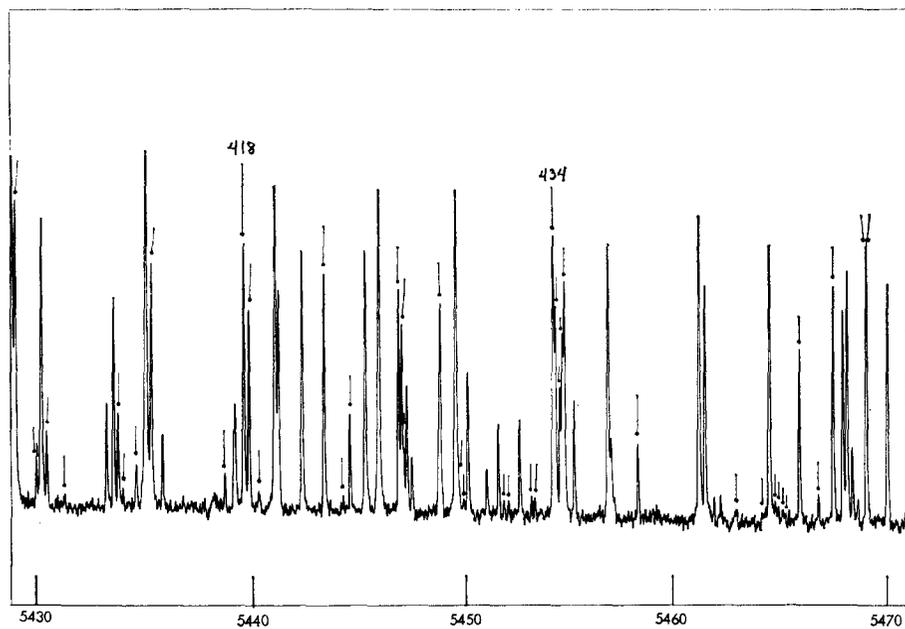
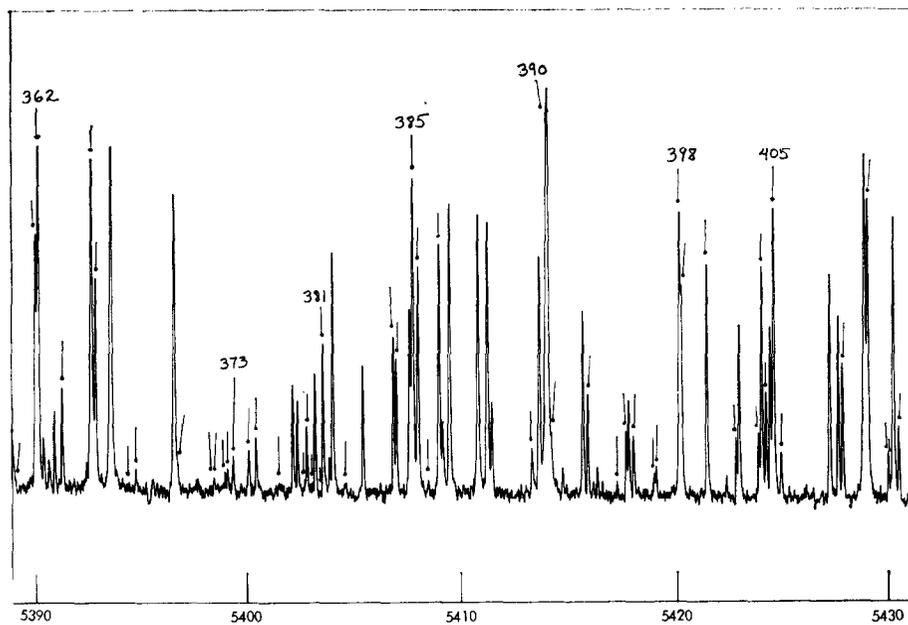


FIG. 1—Continued

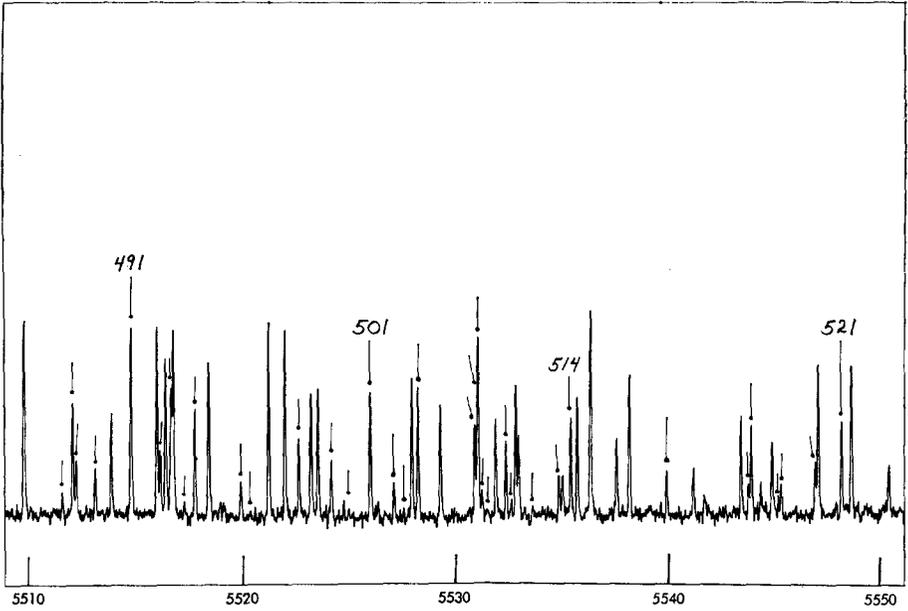
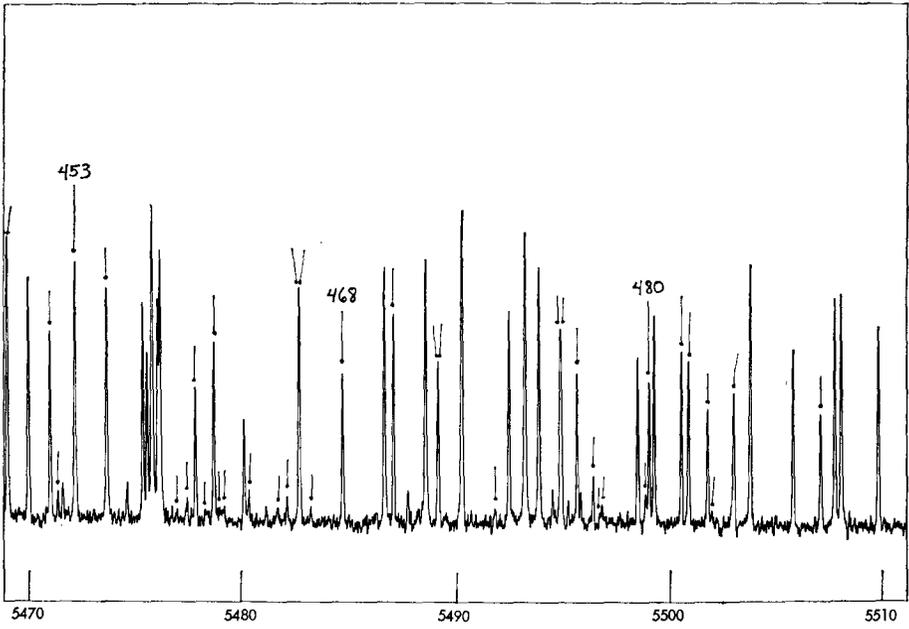


FIG. 1—Continued

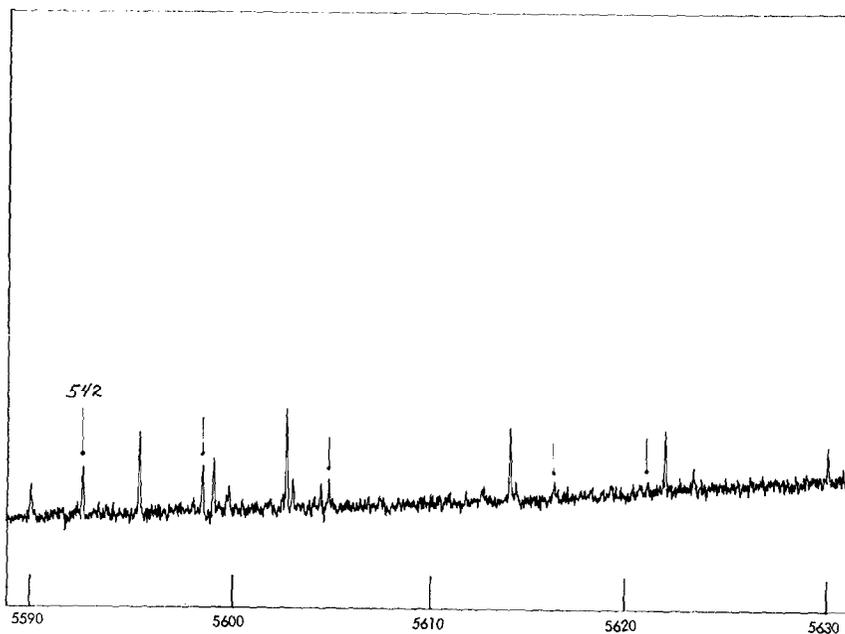
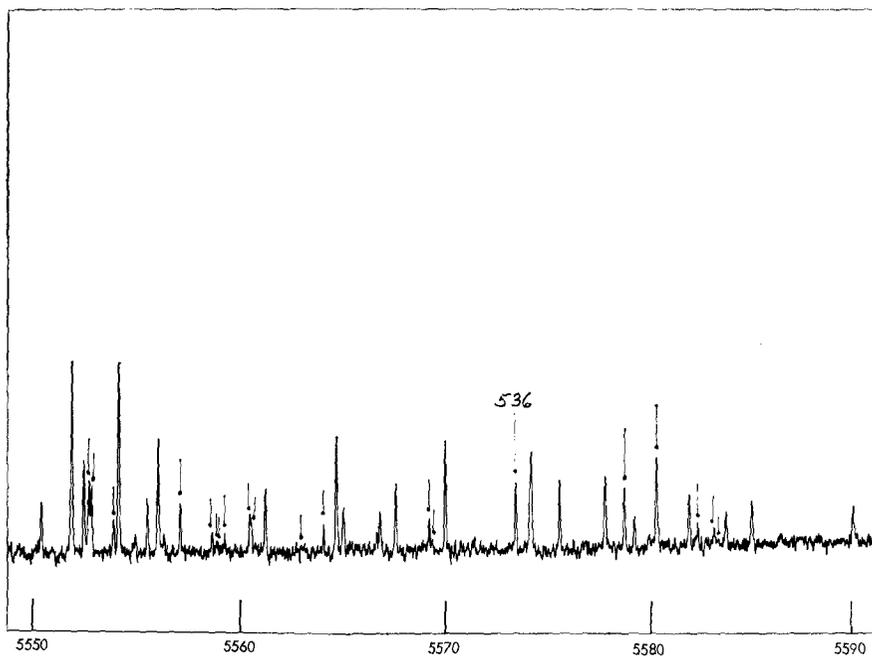


FIG. 1—Continued

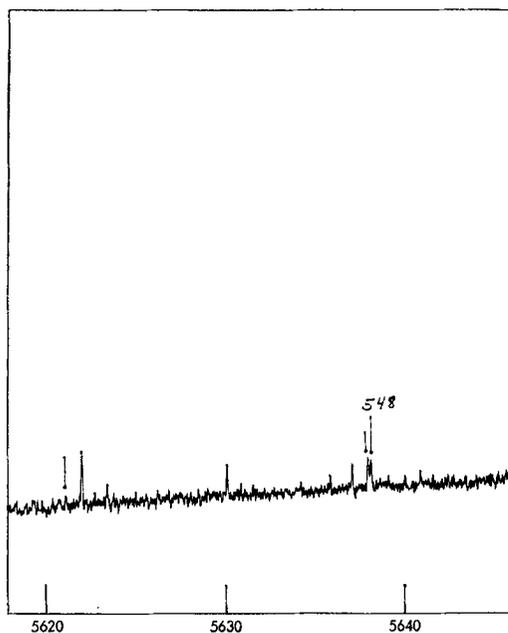


FIG. 1—Continued

TABLE 5

Line strengths of H_2^{18}O in the (011) band measured from spectra of H_2^{16}O . Strength in $\text{cm}^{-2} \text{ atm}^{-1}$ at 296K.

Frequency (cm^{-1})	Upper State J K ₋₁ K ₁	Lower State J K ₋₁ K ₁	Strength		Upper State Level Perturbed ^b
			H_2^{18}O	I X H_2^{16}O	
5186.115	5 1 5	6 1 6	4.76-4	5.46-4	NO
5205.039	4 0 4	5 0 5	6.10-4		NO
5210.178	3 1 2	4 1 3	2.30-4	2.53-4	NO
5243.431	2 0 2	3 0 3	9.35-4		NO
5269.578	1 1 1	2 1 2	6.29-4		NO
5306.544	1 1 1	1 1 0	8.02-4	8.17-4	NO
5317.974	2 2 0	2 2 1	1.08-3		NO
5322.968	3 2 1	3 2 2	1.88-4	1.96-4	YES
5323.674	3 3 1	3 3 0	7.20-4		NO
5324.373	4 3 1	4 3 2	3.65-4	4.63-4	YES
5327.964	2 1 1	2 1 2	6.30-4		NO
5356.013	2 0 2	1 0 1	1.16-3		NO
5372.690	3 1 3	2 1 2	1.12-3		NO
5375.491	3 0 3	2 0 2	4.25-4		NO
5392.749	4 0 4	3 0 3	1.37-3		NO
5407.885 ^c	4 3 1	3 2 2	2.02-4	1.43-4	YES
5408.871	5 0 5	4 0 4	2.45-4		NO
5421.357	4 3 1	3 3 0	2.90-4	2.71-4	YES
5439.669	7 0 7	6 0 6	1.96-4	1.59-4	NO
5443.182	5 3 3	4 3 2	3.14-4	3.14-4	YES
5453.846	6 1 5	5 1 4	5.07-4		NO
5453.959	5 4 2	4 4 1	1.29-4	9.69-5	NO
5478.190	6 4 2	5 4 1	1.14-4	1.13-4	YES

a. $I = \text{H}_2^{18}\text{O}/\text{H}_2^{16}\text{O} = 0.00209$. H_2^{16}O strengths from ref (3)

b. From H_2^{16}O mixing coefficients given in ref (6).

c. (110) band.

lated from the H₂¹⁶O values. The last column in Table 5 indicates if the upper state level is perturbed or not as inferred from the mixing coefficients given by Camy-Peyret and Flaud (δ) for the states (030), (011), and (110) of H₂¹⁶O. The uncertainty in the H₂¹⁸O measured line strength values is $\pm 10\%$.

RECEIVED: February 7, 1977

REFERENCES

1. R. A. TOTH AND J. S. MARGOLIS, *J. Mol. Spectrosc.* **57**, 236 (1975).
2. C. CAMY-PEYRET, J. M. FLAUD, AND R. A. TOTH, *J. Mol. Spectrosc.* **67**, 117-131 (1977).
3. R. A. TOTH, C. CAMY-PEYRET, AND J. M. FLAUD, *J. Quant. Spectrosc. Radiat. Transfer*, in preparation.
4. R. A. TOTH, *J. Quant. Spectrosc. Radiat. Transfer* **13**, 1127 (1973).
5. J. S. GARING AND R. A. McCLATCHEY, *Appl. Opt.* **12**, 2545 (1973).
6. C. CAMY-PEYRET AND J. M. FLAUD, *J. Mol. Spectrosc.* **59**, 327 (1976).