

Line Positions and Intensities for the $\nu_1 + \nu_2$ and $\nu_2 + \nu_3$ Bands of H_2^{18}O

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The intensities of about 90 lines of the $\nu_1 + \nu_2$ and $\nu_2 + \nu_3$ bands of H_2^{18}O have been measured using a Fourier transform spectrum of natural water vapor. The constants involved in the rotational expansion of the transformed transition moment operators corresponding to these bands have been determined through a fit of these line intensities. The constants obtained are used to compute the whole spectrum of the $\nu_1 + \nu_2$ and $\nu_2 + \nu_3$ bands of H_2^{18}O providing reliable line positions and intensities. For lines involving perturbed levels a comparison is given with the results obtained for H_2^{16}O and it is shown that the results for one isotopic species cannot be transferred directly to another one.

INTRODUCTION

High-resolution studies demand both precise line positions and intensities. To calculate the H_2^{18}O spectrum, in addition to the isotopic shift of the vibration-rotation lines, one has to decide how to compute their intensities. Usually, to obtain the absorption of an isotopic substituted molecule, it is assumed that the transition moment does not change very much from one isotopic species to another. Then the absorption of the less abundant isotope is derived from the absorption of the main isotope by multiplying by the isotopic ratio. For example in the case of water, one is used to multiply the absorption of H_2^{16}O by 0.00204 (natural abundance) to get the absorption of H_2^{18}O (1, 2). However, we know from the study of the rotational energy levels of the (110) and (011) vibrational states of H_2^{18}O (3) that the influence of the resonances is very sensitive to the isotopic substitution. Consequently, it is quite evident that the intensities of lines involving perturbed levels cannot be correctly estimated by the method described above. Under these conditions, we have decided to calculate directly the H_2^{18}O spectrum in the 1.9- μm region using realistic vibration-rotation wavefunctions and an appropriate transition moment operator expansion.

The line positions have been derived from a new set of energy levels (4) and line

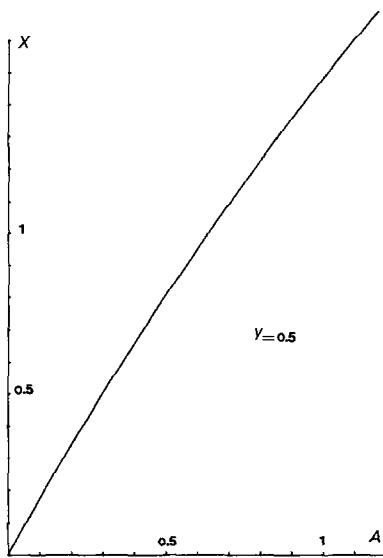


FIG. 1. Curve of growth for a Voigt profile.

TABLE I

Agreement between Observed and Calculated Line Intensities for the 1.9- μm Bands of $\text{H}_2^{18}\text{O}^*$

N°	$\sigma(\text{cm}^{-1})$	$J' K_a K_c'$	$J K_a K_c$	$k_g^P(\text{obs})$ ($\text{cm}^{-2} \text{atm}^{-1}$)	$k_g^P(\text{calc})$ ($\text{cm}^{-2} \text{atm}^{-1}$)	%
70	5121.903	8 0 8	9 0 9	0.6618×10^{-4}	0.7060×10^{-4}	6.7
70	5136.380	6 2 4	7 2 5	0.1040×10^{-3}	0.9661×10^{-4}	7.2
70	5143.603	7 0 7	8 0 8	0.5143×10^{-4}	0.4849×10^{-4}	5.7
70	5143.773	7 1 7	8 1 8	0.1540×10^{-3}	0.1454×10^{-3}	5.6
70	5156.158	6 2 5	7 2 6	0.4472×10^{-4}	0.4482×10^{-4}	0.2
70	5164.744	6 0 6	7 0 7	0.2823×10^{-3}	0.2699×10^{-3}	4.4
70	5165.141	6 1 6	7 1 7	0.8919×10^{-4}	0.9300×10^{-4}	4.3
60	5165.145	5 2 3	5 3 2		0.8880×10^{-4}	8.0
70	5168.794	5 1 4	6 1 5	0.8225×10^{-4}	0.5816×10^{-4}	1.7
70	5185.570	5 4 2	6 4 3	0.5716×10^{-4}	0.3154×10^{-3}	0.3
70	5185.722	4 2 2	5 2 3	0.3164×10^{-3}	0.4449×10^{-3}	1.3
70	5186.115	5 1 5	6 1 6	0.4392×10^{-3}	0.4439×10^{-3}	5.7
70	5188.377	4 1 3	5 1 4	0.4200×10^{-3}	0.1814×10^{-3}	15.5
70	5197.934	4 3 1	5 3 2	0.1571×10^{-3}	0.1023×10^{-3}	0.4
70	5198.472	4 2 3	5 2 4	0.1028×10^{-3}	0.6204×10^{-4}	15.5
70	5201.675	4 3 2	5 3 3	0.7344×10^{-4}	0.6620×10^{-3}	4.9
70	5205.040	4 0 4	5 0 5	0.6312×10^{-3}	0.2134×10^{-3}	0.4
70	5210.178	3 1 2	4 1 3	0.2144×10^{-3}	0.6393×10^{-4}	8.6
70	5210.943	4 4 0	5 4 1	0.6997×10^{-4}	0.1306×10^{-3}	6.6
70	5213.265	3 2 1	4 2 2	0.1399×10^{-3}	0.3182×10^{-3}	4.9
70	5220.917	3 2 2	4 2 3	0.3346×10^{-3}	0.2876×10^{-3}	3.2
70	5224.184	3 0 3	4 0 4	0.2970×10^{-3}	0.1846×10^{-3}	17.5
70	5226.690	3 3 1	4 3 2	0.1571×10^{-3}	0.8238×10^{-3}	7.3
70	5227.851	3 1 3	4 1 4	0.7677×10^{-3}	0.4041×10^{-4}	15.7
70	5233.100	5 1 5	5 1 4	0.3495×10^{-4}	0.9784×10^{-3}	0.6
70	5243.431	2 0 2	3 0 3	0.9839×10^{-3}	0.1354×10^{-3}	6.6
70	5245.345	2 2 1	3 2 2	0.1448×10^{-3}	0.2062×10^{-3}	5.3
70	5259.412	1 1 0	2 1 1	0.2177×10^{-3}	0.3648×10^{-3}	1.3
70	5269.578	1 1 1	2 1 2	0.6375×10^{-3}	0.1135×10^{-3}	4.9
70	5295.314	2 1 2	2 1 1	0.1116×10^{-3}	0.5424×10^{-4}	1.7
70	5298.181	4 2 3	4 2 2	0.5198×10^{-4}	0.1301×10^{-3}	7.3
70	5306.544	1 1 1	1 1 0	0.7634×10^{-3}	0.1768×10^{-3}	4.0
60	5313.017	3 3 0	3 2 1	0.1842×10^{-3}	0.1093×10^{-2}	5.8
70	5315.318	2 2 1	2 2 0	0.3694×10^{-3}	0.1588×10^{-3}	11.8
70	5317.965	2 2 0	2 2 1	0.1042×10^{-2}	0.1301×10^{-3}	7.3
70	5321.569	4 3 2	4 3 1	0.1212×10^{-3}		
70	5322.968	3 2 1	3 2 2	0.1420×10^{-3}		

TABLE I—Continued

N°	$\sigma(\text{cm}^{-1})$	J' K _a ' K _c '	J K _a K _c	$k_{\sigma}^P(\text{obs})$	$k_{\sigma}^P(\text{calc})$	%	(cm ⁻² atm ⁻¹)	
70	5323.675	3 3 1	3 3 0	0.8131 $\times 10^{-3}$	0.8525 $\times 10^{-3}$	4.8		
70	5324.373	4 3 1	4 3 2	0.3852 $\times 10^{-3}$	0.3935 $\times 10^{-3}$	2.2		
70	5325.621	7 4 4	7 4 3	0.3543 $\times 10^{-4}$	0.2734 $\times 10^{-4}$	22.9		
70	5327.564	2 1 1	2 1 2	0.3727 $\times 10^{-3}$	0.3678 $\times 10^{-3}$	1.3		
70	5334.023	1 0 1	0 0 0	0.2293 $\times 10^{-3}$	0.2126 $\times 10^{-3}$	7.3		
70	5340.780	7 5 3	7 5 2	0.2793 $\times 10^{-4}$	0.2526 $\times 10^{-4}$	9.5		
70	5356.013	2 0 2	1 0 1	0.1050 $\times 10^{-2}$	0.1105 $\times 10^{-2}$	5.2		
60	5360.267	3 2 1	2 1 2	0.3529 $\times 10^{-4}$	0.3654 $\times 10^{-4}$	3.5		
70	5366.383	6 2 4	6 2 5	0.4692 $\times 10^{-4}$	0.4337 $\times 10^{-4}$	7.6		
70	5372.690	3 1 3	2 1 2	0.1026 $\times 10^{-2}$	0.1101 $\times 10^{-2}$	7.3		
70	5375.490	3 0 3	2 0 2	0.4274 $\times 10^{-3}$	0.4286 $\times 10^{-3}$	0.3		
60	5386.896	3 3 1	2 2 0	0.3027 $\times 10^{-4}$	0.3435 $\times 10^{-4}$	13.5		
70	5390.192	3 1 2	2 1 1	0.3517 $\times 10^{-3}$	0.3478 $\times 10^{-3}$	1.1		
70	5392.749	4 0 4	3 0 3	0.1142 $\times 10^{-2}$	0.1220 $\times 10^{-2}$	6.9		
70	5403.460	5 2 4	5 0 5	0.4349 $\times 10^{-4}$	0.4863 $\times 10^{-4}$	11.8		
60	5406.715	4 3 2	3 2 1	0.5783 $\times 10^{-4}$	0.5492 $\times 10^{-4}$	5.0		
70	5407.884	4 2 3	3 2 2	0.2023 $\times 10^{-3}$	0.2064 $\times 10^{-3}$	2.0		
70	5408.871	5 0 5	4 0 4	0.3164 $\times 10^{-3}$	0.3339 $\times 10^{-3}$	5.5		
70	5420.081	4 2 2	3 2 1	0.6195 $\times 10^{-3}$	0.6318 $\times 10^{-3}$	2.0		
70	5420.177	4 3 2	3 3 1	0.8682 $\times 10^{-4}$	0.9411 $\times 10^{-4}$	8.4		
70	5421.358	4 3 1	3 3 0	0.3211 $\times 10^{-3}$	0.2823 $\times 10^{-3}$	12.0		
70	5423.910	6 1 6	5 1 5	0.2358 $\times 10^{-3}$	0.2401 $\times 10^{-3}$	1.8		
70	5439.426	7 1 7	6 1 6	0.4670 $\times 10^{-3}$	0.4668 $\times 10^{-3}$	0.0		
70	5439.668	7 0 7	6 0 6	0.1586 $\times 10^{-3}$	0.1560 $\times 10^{-3}$	1.6		
70	5443.182	5 3 3	4 3 2	0.2964 $\times 10^{-3}$	0.3164 $\times 10^{-3}$	6.7		
70	5446.664	5 2 3	4 2 2	0.1916 $\times 10^{-3}$	0.1840 $\times 10^{-3}$	4.0		
70	5446.813	5 3 2	4 3 1	0.1074 $\times 10^{-3}$	0.1050 $\times 10^{-3}$	2.2		
70	5453.846	6 1 5	5 1 4	0.4743 $\times 10^{-3}$	0.4851 $\times 10^{-3}$	2.3		
70	5453.960	5 4 2	4 4 1	0.1081 $\times 10^{-3}$	0.1088 $\times 10^{-3}$	0.6		
70	5454.170	5 4 1	4 4 0	0.3460 $\times 10^{-4}$	0.3625 $\times 10^{-4}$	4.8		
70	5454.379	8 0 8	7 0 7	0.2821 $\times 10^{-3}$	0.2705 $\times 10^{-3}$	4.1		
70	5465.413	6 3 4	5 3 3	0.8694 $\times 10^{-4}$	0.8074 $\times 10^{-4}$	7.1		
70	5466.971	7 2 6	6 2 5	0.2781 $\times 10^{-3}$	0.2623 $\times 10^{-3}$	5.7		
70	5470.519	7 1 6	6 1 5	0.9925 $\times 10^{-4}$	0.9325 $\times 10^{-4}$	6.0		
70	5471.677	6 2 4	5 2 3	0.3572 $\times 10^{-3}$	0.3709 $\times 10^{-3}$	3.9		
70	5473.161	6 3 3	5 3 2	0.2536 $\times 10^{-3}$	0.2348 $\times 10^{-3}$	7.4		
70	5477.334	6 4 3	5 4 2	0.3672 $\times 10^{-4}$	0.3668 $\times 10^{-4}$	0.1		
70	5478.190	6 4 2	5 4 1	0.1093 $\times 10^{-3}$	0.1100 $\times 10^{-3}$	0.6		
70	5494.480	7 2 5	6 2 4	0.7018 $\times 10^{-4}$	0.6865 $\times 10^{-4}$	2.2		
70	5494.543	4 2 2	3 0 3	0.3490 $\times 10^{-4}$	0.3176 $\times 10^{-4}$	9.0		
70	5495.288	11 1 11	10 1 10	{ 0.1568 $\times 10^{-4}$	0.3754 $\times 10^{-4}$	17.8		
70	5495.296	11 0 11	10 0 10					
70	5500.289	7 4 4	6 4 3	0.7691 $\times 10^{-4}$	0.7538 $\times 10^{-4}$	2.0		
70	5500.628	9 2 8	8 2 7	0.5936 $\times 10^{-4}$	0.6781 $\times 10^{-4}$	14.2		
70	5501.533	9 1 8	8 1 7	0.2740 $\times 10^{-4}$	0.2293 $\times 10^{-4}$	16.3		
70	5506.840	8 3 6	7 3 5	0.3070 $\times 10^{-4}$	0.2623 $\times 10^{-4}$	14.6		
70	5514.656	8 2 6	7 2 5	0.9135 $\times 10^{-4}$	0.9814 $\times 10^{-4}$	7.4		
60	5517.630	8 4 5	7 3 4	0.2479 $\times 10^{-4}$	0.2295 $\times 10^{-4}$	7.4		
70	5525.880	9 3 7	8 3 6	0.3403 $\times 10^{-4}$	0.3605 $\times 10^{-4}$	6.0		
70	5528.133	8 4 4	7 4 3	0.3703 $\times 10^{-4}$	0.4068 $\times 10^{-4}$	9.9		
70	5530.958	8 3 5	7 3 4	0.4080 $\times 10^{-4}$	0.4910 $\times 10^{-4}$	20.4		

* The meaning of the different columns is:

N°, vibrational assignment of the transition consisting of two numbers which correspond to the upper and lower states with the abbreviation 0 for (000), 6 for (110), and 7 for (011);

σ , computed wavenumber (cm⁻¹) of the transition;

J'K_a'K_c' and JK_aK_c, rotational quantum numbers, respectively, of the upper and lower levels;

k_{σ}^P (obs) k_{σ}^P (calc), observed and calculated values of the line intensity in cm⁻² atm⁻¹;

% = 100 × | k_{σ}^P (obs) - k_{σ}^P (calc)| / k_{σ}^P (obs).

The temperature is equal to 296 K.

intensities have been deduced from a Fourier transform spectrum of water vapor where the abundance of H_2^{18}O was the natural one. From these experimental data on line intensities, the constants appearing in the transformed transition moment operator have been deduced following the theoretical formulation proposed in Ref. (5). Finally, using these constants, we have calculated the whole spectrum of the $\nu_1 + \nu_2$ and $\nu_2 + \nu_3$ bands of H_2^{18}O .

EXPERIMENTAL LINE INTENSITIES

The spectrum of natural water (0.204% H_2^{18}O , 0.037% H_2^{17}O) has been recorded with a Fourier transform interferometer (resolution 0.015 cm^{-1}) with a pressure¹ $P \simeq 1.77 \times 10^{-2}$ atm at 296 K. At this pressure both collisions and Doppler effects contribute to the line broadening. In these conditions, the absorption coefficient $\alpha(\sigma)$ (in cm^{-1}) at the wavenumber σ is related to the reduced Voigt profile as (7):

$$\frac{\alpha(\sigma)}{\alpha_0} = K(x, y) = \frac{y}{\pi} \int_{-\infty}^{+\infty} \frac{\exp(-t^2)}{y^2 + (x - t)^2} dt,$$

$$\alpha_0 = \frac{k_\sigma P}{\gamma_D} \left(\frac{\log 2}{\pi} \right)^{\frac{1}{2}},$$

$$y = \frac{\gamma_L}{\gamma_D} (\log 2)^{\frac{1}{2}}, \quad x = \frac{\sigma - \sigma_0}{\gamma_D} (\log 2)^{\frac{1}{2}}, \quad \gamma_L = \gamma_L^0 P,$$

where $k_\sigma P$ is the line strength or integrated absorption coefficient at unit pressure (usually in $\text{cm}^{-2} \text{ atm}^{-1}$), γ_D is the Doppler half-width, γ_L is the Lorentz half-width at pressure P , and σ_0 is the wavenumber at the line center.

TABLE II

Constants Involved in the Expansion of the Transformed Transition Moment Operator $\nu \mu' z$ for H_2^{18}O

Band	j	ν_{A_j}	$\nu_{\mu'_j}$
$\nu=6$ $(\nu_1 + \nu_2)$	1	φ_x	$-0.00219_8 \pm 0.00031$
	4	$\{i \varphi_y, J_z\}$	$0.119^a \times 10^{-3}$
	5	$\{\varphi_z, i J_y\}$	$0.646^a \times 10^{-4}$
$\nu=7$ $(\nu_2 + \nu_3)$	1	φ_z	$0.01830_6 \pm 0.00022$
	4	$\frac{1}{2} [\{\varphi_x, i J_y\} - \{i \varphi_y, J_x\}]$	$(-0.116_6 \pm 0.041) \times 10^{-3}$

All the results are in D. The errors quoted are 99% statistical confidence intervals.

a Constrained to the H_2^{16}O value given in Ref. [6].

¹ A slight readjustment of the measured pressure has been done to improve the absolute precision of the results using the intensities of some H_2^{16}O lines computed in Ref. (6).

TABLE III
Rotational Constants for the (000), (110), and (011) States of $\text{H}_2^{18}\text{O}^a$

	(000)	(110)	(011)
E_v	0.0000		
A_v	$27.531075_8 \pm 0.000098$	$29.857_7 \pm 0.011$	$29.1757_1 \pm 0.0039$
B_v	$14.522139_8 \pm 0.000056$	$14.4818_7 \pm 0.0026$	$14.6088_7 \pm 0.0018$
C_v	$9.237985_8 \pm 0.000022$	$8.9121_3 \pm 0.0016$	$8.9577_8 \pm 0.0012$
Δ_k^v	$(0.31596_6 \pm 0.00013) \times 10^{-1}$	$(0.520_7 \pm 0.018) \times 10^{-1}$	$(0.4886_5 \pm 0.0040) \times 10^{-1}$
Δ_{JK}^v	$(-0.5692_3 \pm 0.0017) \times 10^{-2}$	$(-0.748_5 \pm 0.048) \times 10^{-2}$	$(-0.767_3 \pm 0.011) \times 10^{-2}$
Δ_J^v	$(0.12531_1 \pm 0.00015) \times 10^{-2}$	$(0.1372_3 \pm 0.0030) \times 10^{-2}$	$(0.1456_3 \pm 0.0017) \times 10^{-2}$
δ_k^v	$(0.1288_4 \pm 0.0015) \times 10^{-2}$	$(0.276_3 \pm 0.028) \times 10^{-2}$	$(0.335_0 \pm 0.011) \times 10^{-2}$
δ_j^v	$(0.51014_6 \pm 0.00044) \times 10^{-3}$	$(0.584_1 \pm 0.016) \times 10^{-3}$	$(0.606_1 \pm 0.013) \times 10^{-3}$
δ_k^v	$(0.1156_0 \pm 0.0012) \times 10^{-3}$	$(0.327_5 \pm 0.089) \times 10^{-3}$	$(0.262_6 \pm 0.012) \times 10^{-3}$
δ_{KJ}^v	$(-0.171_9 \pm 0.016) \times 10^{-4}$	$(-0.59_0 \pm 0.27) \times 10^{-4}$	$(-0.369_6 \pm 0.037) \times 10^{-4}$
δ_{JK}^v	$(-0.128_6 \pm 0.018) \times 10^{-7}$		
δ_j^v	$(0.525_7 \pm 0.013) \times 10^{-6}$		$(0.70_0 \pm 0.11) \times 10^{-6}$
δ_k^v	$(0.271_0 \pm 0.016) \times 10^{-4}$		$(0.691_9 \pm 0.094) \times 10^{-4}$
δ_j^v	$(0.2676_6 \pm 0.0051) \times 10^{-6}$		$(0.325_4 \pm 0.076) \times 10^{-6}$
δ_k^v	$(-0.405_1 \pm 0.024) \times 10^{-6}$		$(-0.1012_1 \pm 0.0096) \times 10^{-5}$
δ_{KKJ}^v	$(0.48_5 \pm 0.26) \times 10^{-7}$		
δ_k^v	$(-0.108_8 \pm 0.024) \times 10^{-6}$		$(-0.42_0 \pm 0.15) \times 10^{-6}$
<u>COUPLING CONSTANT</u>			
$k_{76} = -0.3029_2 \pm 0.0013$			

^a All the results are given in cm^{-1} and the quoted errors are 99% confidence intervals.

The curve of growth method (8, 9) has been used for the computation of the line strength k_σ^P . For that we have calculated the function $X(A, y)$ defined as follows:

$$X(A, y) = \int_{-\infty}^{+\infty} \{1 - \exp[-AK(x, y)]\} dx,$$

where

$$A = \alpha_0 l \quad \text{and} \quad X = \frac{W}{\gamma_D} (\log 2)^{\frac{1}{2}}.$$

l is the absorption path and W the equivalent width (i.e., the area under the absorption line).

For H_2O self-broadening, γ_L^0 can appreciably vary from one line to another around the mean value of $0.3 \text{ cm}^{-1} \text{ atm}^{-1}$ and this yields to variation of y between approximatively 0.4 and 0.6. On account of the maximum value measured for W in the studied spectrum, the corresponding values of X are less than 1.7. Thus using Jansson and Korb's tables (10) we can see that the maximum error on A due to an uncertainty on y does not exceed 11%. Since we are close enough to the linear region of the curve of growth y is not a very sensitive parameter. We have precisely sampled the curve of growth (Fig. 1) for the mean value 0.5 of y using a program (11) based on Gautschi's algorithm (12)

TABLE IV
Calculated Rotational Energy Levels for the Vibrational States (000), (110), and (011)*

J	k_a	k_c	(000)	(110)	% (011)	(011)	% (110)
0	0	0	0.000	a 5221.251	0.0	a 5310.464	0.0
1	0	1	23.755	a 5244.638	0.0	a 5334.026	0.0
1	1	1	36.748	a 5259.985	0.0	a 5348.568	0.0
1	1	0	42.024	a 5265.540	0.0	a 5354.200	0.0
2	0	2	69.928	a 5290.128	0.0	a 5379.770	0.0
2	1	2	78.983	a 5301.212	0.0	a 5380.164	0.1
2	1	1	94.789	a 5317.848	0.0	a 5406.955	0.0
2	2	1	133.476	a 5363.369	0.0	a 5450.102	0.0
2	2	0	134.783	a 5364.605	0.1	a 5451.446	0.0
3	0	3	136.398	a 5355.559	0.0	a 5445.420	0.0
3	1	3	140.560	a 5362.000	0.0	a 5451.681	1.4
3	2	2	172.584	a 5355.301	0.0	a 5484.983	0.1
3	2	2	204.156	a 5431.542	0.1	a 5519.538	30.2
3	2	1	210.800	a 5439.258	1.4	a 5527.727	10.9
3	3	1	282.096	a 5521.687	10.9	a 5605.984	0.0
3	3	0	282.308	a 5523.825	30.2	a 5606.195	0.0
4	0	4	221.235	a 5439.042	0.0	a 5529.089	0.0
4	1	4	252.822	a 5442.595	0.0	a 5531.868	3.4
4	1	3	274.806	a 5496.480	0.0	a 5586.738	0.1
4	2	3	298.621	a 5521.992	0.1	a 5612.642	16.2
4	2	2	314.461	b 5541.938	3.4	a 5630.880	4.0
4	3	2	379.292	a 5617.514	4.0	a 5702.278	0.9
4	3	1	380.703	a 5620.550	16.2	a 5703.668	1.0
4	4	1	482.645	c 5736.603	0.9	a 5815.705	0.0
4	4	0	482.674	c 5736.618	0.9	a 5815.733	0.0
5	0	5	324.049	a 5539.815	0.0	a 5630.107	0.0
5	1	5	325.217	a 5541.520	0.0	a 5631.461	0.3
5	1	4	398.364	a 5619.298	0.0	a 5709.979	0.0
5	2	4	414.170	a 5639.843	0.1	a 5727.510	5.9
5	2	3	445.161	b 5670.891	0.3	a 5761.120	1.6
5	3	3	500.598	a 5737.168	1.7	a 5822.481	1.9
5	3	2	505.730	a 5743.000	6.0	a 5827.517	2.4
5	4	2	604.545	b 5856.945	2.4	a 5936.608	0.5
5	4	1	604.793	b 5857.067	1.9	a 5936.846	0.5
5	5	1	733.681	a 6003.525	0.5	a 6077.102	0.0
5	5	0	733.684	a 6003.527	0.5	a 6077.105	0.0
6	0	6	444.849	a 5657.882	0.0	a 5748.529	0.0
6	1	6	445.349	a 5658.660	0.0	a 5749.126	0.1
6	1	5	541.184	a 5761.345	0.0	a 5852.216	0.0
6	2	5	550.454	a 5774.149	0.1	a 5862.751	2.1
6	2	4	601.241	a 5880.233	0.2	a 5916.125	2.7
6	3	4	645.395	a 5880.233	0.6	a 5966.013	2.6
6	3	3	658.112	a 5898.847	2.1	a 5978.889	5.7
6	4	3	751.034	a 6001.747	5.8	a 6081.888	1.1
6	4	2	752.189	b 6002.278	2.7	a 6082.990	1.1
6	5	2	880.077	b 6147.797	1.1	a 6222.399	0.5
6	5	1	880.115	b 6147.819	1.1	a 6222.435	0.5
6	6	1	1033.199	b 6321.545	0.4	b 6387.488	0.1
6	6	0	1033.199	b 6321.545	0.4	b 6387.489	0.1

leading to a quick computation of the Voigt profile. Then the reciprocal curve of growth $A(X)$ has been obtained by a least-squares fit expanding A with respect to X as a polynomial of degree 7 for X varying from 0 to 1.7 with a rms of 0.6×10^{-7} . Using this polynomial we have then calculated the line strengths k_a^P from the measured values of the equivalent width W . However, since the automatic determination of the equivalent width W (13) does not provide us with the true value of W , wings and baseline corrections are necessary (14). Being close to the linear region of the curve of growth, the general correction formula reduces to Eq. (12) of Ref. (15). The final results concerning the intensities appear in Table I. The precision on the measurements is about 10%.

CALCULATED LINE INTENSITIES

The method for calculating vibration-rotation line intensities has been described elsewhere (5) and used successfully for H_2^{18}O (6, 16, 17). It takes into account simultaneously the influence of the centrifugal distortion, of the variation of the rotational constants

TABLE IV—Continued

J	k _a	k _c	(000)	(110)	% (011)	(011)	% (110)
7	0	7	583.781	a 5793.402	0.0	a 5884.518	0.0
7	1	7	583.990	a 5797.750	0.0	a 5884.773	0.1
7	1	6	701.698	b 5920.772	0.0	a 6011.709	0.1
7	2	6	706.601	5928.014	0.0	a 6017.430	0.8
7	2	5	780.456	6003.577	0.1	b 6095.700	0.3
7	3	5	812.765	6045.281	0.5	a 6132.055	2.8
7	3	4	839.552	c 6070.651	0.9	a 6157.371	20.0
7	4	4	921.899	b 6171.581	20.3	a 6251.334	1.7
7	4	3	925.03	c 6172.867	2.9	a 6254.961	1.8
7	5	3	1020.993	6316.269	1.9	a 6397.984	1.0
7	5	2	1051.200	6316.391	1.8	a 6392.182	1.0
7	6	2	1204.172	6489.701	0.9	a 6557.265	0.5
7	6	1	1204.178	6489.704	0.9	a 6557.270	0.5
7	7	1	1378.969			a 6744.000	0.1
7	7	0	1378.969			a 6744.000	0.1
8	0	8	740.915	5946.478	0.0	a 6038.163	0.0
8	1	8	741.002	b 5946.633	0.0	a 6038.271	0.0
8	1	7	879.498	6096.862	0.0	a 6187.327	0.1
8	2	7	881.917	6100.691	0.0	a 6190.817	0.4
8	2	6	980.225	6202.507	0.1	b 6295.089	0.2
8	3	6	1001.709	6231.921	0.3	a 6319.615	2.4
8	3	5	1047.332	6275.518	0.5	a 6370.527	25.0
8	4	5	1116.641	d 6357.323	25.2	a 6444.470	2.3
8	4	4	1126.446	6369.751	2.5	a 6453.825	2.8
8	5	4	1246.376	6508.918	3.0	a 6585.786	1.5
8	5	3	1247.214	6509.396	2.4	a 6586.564	1.5
8	6	3	1399.436			a 6751.088	1.0
8	6	2	1399.471			a 6751.120	1.0
8	7	2	1574.658			a 6938.358	0.5
8	8	1	1574.658			a 6938.359	0.5
8	8	1	1586.562			a 7143.353	0.1
8	8	0	1768.562			a 7143.353	0.1
9	0	9	916.261	c 6117.141	0.0	a 6209.487	0.0
9	1	9	916.297	a 6117.211	0.0	a 6209.533	0.0
9	1	8	1074.765	6289.673	0.0	a 6381.031	0.1
9	2	8	1075.111	6289.623	0.0	d 6382.431	0.2
9	2	7	1198.202	6419.840	0/1	b 6512.500	0.1
9	3	7	1211.188	6438.923	0.2	a 6600.801	1.6
9	3	6	1279.803	6505.493	0.4	a 6600.801	1.6
9	4	6	1324.486	6574.291	4.8	a 6660.562	4.6
9	4	5	1355.212			b 6680.369	4.6
9	5	5	1466.032			a 6803.603	2.1
9	5	4	1468.629			b 6806.006	2.1
9	6	4	1618.914			a 6968.855	1.5
9	6	3	1619.074			a 6969.001	1.5
9	7	3	1794.362			7156.500	1.0
9	7	2	1794.368			7156.505	1.0
9	8	2	1989.108			7362.329	0.6
9	8	1	1989.108			7362.329	0.6
10	0	10	1109.789	d 6305.380	0.0	a 6398.474	0.0
10	1	10	1109.804	b 6305.411	0.0	a 6398.494	0.0
10	1	9	1287.734	6499.477	0.0	a 6591.290	0.1
10	2	9	1288.266	6500.455	0.0	b 6591.954	0.1
10	2	8	1433.032	6653.808	0.1	b 6746.365	0.1
10	3	8	1440.291	6665.331	0.2	a 6754.977	1.0
10	3	7	1534.379			a 6853.529	1.3
10	4	7	1534.379			a 6853.529	2.8
10	4	6	1611.678			c 6934.151	2.4
10	5	6	1709.561			7045.044	2.6
10	5	5	1716.232			a 7051.211	2.8
10	6	5	1862.487			a 7210.420	2.0
10	6	4	1863.055			7210.935	2.0
10	7	4	2037.939			7398.250	1.5
10	7	3	2037.967			7398.275	1.5

with the vibrational quantum numbers, and of the resonances. For that purpose, the matrix elements of the transformed dipole moment operator are taken between realistic vibration-rotation wavefunctions of the lower and upper states of the transition. More precisely, the relevant part of this operator can be expanded as:

$$\mu' z = \sum_{v=6,7} |0\rangle^* \mu' z \langle v| \quad \text{with } v = \begin{cases} 6 & \text{for (110),} \\ 7 & \text{for (011).} \end{cases}$$

${}^*\mu' z$ is the transformed transition moment operator which can be written as:

$${}^*\mu' z = \sum_j {}^*\mu'_j {}^*A_j,$$

where the *A_j 's are rotational operators (see Table II) and the ${}^*\mu'_j$ are constants to be

TABLE IV—Continued

J	k_a	k_c	(000)	(110)	% (011)	(011)	% (110)
11	0	11	1321.452	6511.158	0.0	a 6605.089	0.0
11	1	11	1321.459	6511.173	0.0	a 6605.097	0.0
11	1	10	1518.539	6726.437	0.0	a 6818.879	0.1
11	2	10	1518.784	6726.928	0.0	b 6819.191	0.1
11	2	9	1684.446			b 6996.263	0.1
11	3	9	1688.291			b 7000.916	0.6
11	3	8	1808.389			a 7126.660	0.5
11	4	8	1835.000			a 7257.856	2.5
11	4	7	1834.246			b 7257.859	2.3
11	5	7	1976.331			7309.502	3.1
11	5	6	1990.920			7323.026	3.6
12	0	12	1551.194	6734.424	0.0	a 6829.285	0.0
12	1	12	1551.197	6734.431	0.0	a 6829.289	0.0
12	1	11	1767.219	6970.636	0.0	a 7061.873	0.1
12	2	11	1767.332	6970.886	0.0	b 7064.020	0.1
12	2	10	1952.696			b 7262.418	0.2
12	3	10	1954.669			7264.849	0.4
12	3	9	2099.608			7417.000	0.3
13	0	13	1798.948	6975.118	0.0	a 7071.011	0.0
13	1	13	1798.949	6975.122	0.0	a 7071.013	0.0
13	1	12	2033.761			c 7326.292	0.1
13	2	12	2033.813			a 7326.361	0.1
13	2	11	2238.075			7545.167	0.2
13	3	11	2239.070			7546.414	0.3
14	0	14	2064.641			7330.213	0.0
14	1	14	2064.642			7330.214	0.0
14	1	13	2318.118			7606.120	0.1
14	2	13	2318.142			7606.153	0.1
15	0	15	2348.196			7606.834	0.0
15	1	15	2348.196			7606.835	0.0

^a A letter in front of an energy level means that the experimental value E^{obs} has been used in the least-squares fit. Moreover, the value of $\delta = |E^{\text{obs}} - E^{\text{calc}}|$ (in cm^{-1}) is such that: for a , $0 \times 10^{-3} < \delta \leq 20 \times 10^{-3}$ for b , $20 \times 10^{-3} < \delta \leq 40 \times 10^{-3}$ for c , $40 \times 10^{-3} < \delta \leq 80 \times 10^{-3}$ for d , $80 \times 10^{-3} < \delta \leq 140 \times 10^{-3}$. For the ground state, the levels introduced in the fit are those of Ref. (7) leading to a rms of $3.5 \times 10^{-3} \text{ cm}^{-1}$.

determined to give the best agreement with the experimental line intensities. The wavefunctions used to calculate the matrix elements of ${}^9\mu'z$ have been taken from Ref. (3).

In this reference the wavefunctions were calculated using the rotational constants listed in Table III and the corresponding energy levels are gathered in Table IV.

As suggested by the Editor, these two tables are reproduced here because they have come out very badly in Ref. (3).

As can be seen in Table I which lists the experimental intensities, there are relatively few lines belonging to the $\nu_1 + \nu_2$ band. This should be expected since this band is weaker² than $\nu_2 + \nu_3$ roughly by a factor 30. Thus, practically, the only lines of $\nu_1 + \nu_2$ band actually seen in the spectrum are those whose intensity is enhanced by perturbation. Under these conditions we have been compelled to fix the value of the constants appearing in the rotational correcting terms of the expansion of ${}^9\mu'z$ to the value obtained for H_2^{16}O (6). So, in addition to the constants ${}^7\mu'$, relative to the $\nu_2 + \nu_3$ band the only constant allowed to vary for the $\nu_1 + \nu_2$ band is the vibrational part ${}^6\mu'_1$ of the transformed transition moment operator corresponding to this band. The final values of the ${}^9\mu'$, introduced in the fit appear in Table II.

It is satisfying to see that the constants ${}^9\mu'$, for H_2^{18}O are comparable to those of H_2^{16}O . A statistical analysis of the relative differences between observed and calculated

² Moreover, we may recall that the spectrum used for measuring H_2^{18}O intensities is a spectrum of natural water vapor in which the amount of H_2^{18}O is 0.204%.

TABLE V

Computed List of the Lines of the $\nu_1 + \nu_2$ and $\nu_2 + \nu_3$ Bands of H_2^{18}O at 296 K^a

NO	SIGMA	J'KA'KC'	J	KA	KC	E	INTENSITY	NO	SIGMA	J'KA'KC'	J	KA	KC	E	INTENSITY
70	4976.471	5 2 4	6 4	3		751.036	0.680D-25	70	5157.280	6 4 2	7	4	3	925.704	0.142D-23
70	4980.076	5 1 5	6 3 4			645.385	0.555D-25	60	5157.540	3 1 3	3	2	2	204.758	0.849D-25
60	5002.865	4 4 1	5 5 0			733.685	0.873D-25	70	5157.823	7 6 2	8	6	3	1399.437	0.107D-24
70	5012.652	6 1 5	7 5 0			839.555	0.112D-24	70	5158.775	8 1 8	8	1	7	879.493	0.529D-25
60	5012.721	4 3 2	5 4 1			604.794	0.158D-24	70	5159.887	5 2 3	6	2	4	601.238	0.258D-23
60	5020.264	4 2 3	5 3 2			505.730	0.121D-24	70	5159.981	6 4 3	7	4	4	921.900	0.481D-24
60	5025.490	5 2 3	6 3 4			645.385	0.583D-25	60	5162.231	6 4 3	7	3	4	839.555	0.170D-24
70	5030.347	12 0 12	13 0 13			1798.948	0.586D-25	60	5164.022	7 2 5	7	3	4	839.555	0.503D-25
70	5036.892	3 2 2	4 4 1			482.646	0.181D-24	60	5164.240	4 0 4	4	1	3	274.803	0.101D-24
60	5039.005	3 3 1	4 4 0			482.676	0.855D-25	70	5164.744	6 0 6	7	0	7	583.782	0.109D-22
60	5041.171	3 3 0	4 4 1			482.646	0.750D-25	60	5164.879	2 0 2	3	0	3	136.338	0.203D-24
70	5051.364	5 1 4	6 3 3			658.609	0.821D-25	70	5165.141	6 1 6	7	1	7	563.985	0.362D-23
70	5051.839	11 1 20	12 2 11			1767.335	0.599D-25	60	5165.145	5 2 3	5	2	2	505.730	0.132D-24
60	5052.818	3 2 0	4 4 3			380.743	0.746D-25	60	5165.650	1 0 2	4	6	4	78.990	0.365D-24
70	5053.866	11 1 11	12 1 12			1551.195	0.595D-25	70	5165.800	6 2 2	2	2	2	751.036	0.527D-25
70	5053.894	11 1 11	12 1 12			1551.195	0.179D-24	60	5167.740	1 2 2	2	2	1	133.477	0.212D-24
60	5059.966	3 2 1	4 4 2			379.291	0.254D-24	70	5168.794	5 1 4	2	6	1	541.179	0.358D-23
70	5061.956	10 2 8	11 2 9			1684.437	0.779D-25	70	5168.907	5 3 2	6	3	3	658.609	0.162D-23
60	5068.843	5 1 4	6 2 5			550.453	0.733D-25	70	5166.969	2 0 2	3	2	1	210.800	0.956D-24
70	5072.389	3 1 3	4 4 3			379.291	0.808D-25	70	5171.230	6 5 1	7	5	2	1051.207	0.565D-24
70	5072.742	16 1 9	11 1 10			1518.541	0.182D-24	70	5171.407	6 5 2	7	5	3	1050.994	0.189D-24
70	5073.132	10 2 9	11 2 10			1518.798	0.640D-25	60	5172.356	4 3 2	5	2	3	445.159	0.421D-24
70	5077.019	10 0 10	11 0 11			1321.457	0.496D-24	60	5174.137	5 1 4	5	2	3	445.159	0.244D-24
70	5077.024	10 1 10	11 1 11			1321.464	0.165D-24	70	5177.054	5 2 4	6	2	5	550.453	0.917D-23
60	5077.385	8 4 5	9 3 6			1279.800	0.102D-24	70	5177.088	5 3 3	6	3	4	645.385	0.518D-23
70	5079.494	9 2 7	10 2 8			1433.030	0.793D-25	70	5177.922	7 0 7	7	2	6	706.599	0.119D-23
70	5081.008	4 1 3	5 3 2			505.738	0.338D-24	60	5182.011	4 1 3	4	2	2	314.461	0.122D-24
60	5081.064	2 2 1	3 3 0			282.306	0.348D-24	60	5182.675	3 0 3	3	1	2	172.884	0.463D-24
60	5082.511	2 2 0	3 3 1			282.095	0.118D-24	60	5183.068	2 1 1	2	2	0	134.785	0.960D-25
70	5083.928	4 0 4	5 2 2			445.155	0.329D-24	70	5183.077	7 1 7	7	1	6	701.695	0.373D-24
70	5086.098	9 4 6	10 4 7			1574.458	0.107D-24	70	5183.297	6 0 6	7	6	1	1204.177	0.146D-24
70	5087.300	9 3 7	10 3 8			1404.291	0.222D-24	70	5184.335	9 2 8	9	2	7	1198.210	0.588D-25
60	5090.417	2 1 2	3 2 1			210.800	0.186D-24	60	5184.490	0 0 0	1	1	1	36.750	0.107D-24
70	5090.713	8 3 5	9 3 6			1279.800	0.340D-24	60	5184.498	3 1 2	3	2	1	210.800	0.407D-24
70	5093.299	9 1 8	10 1 9			1267.727	0.167D-24	70	5184.655	5 4 1	6	4	2	752.191	0.777D-24
70	5094.017	9 5 6	10 5 6			1709.575	0.503D-25	70	5185.252	5 0 5	6	0	6	444.853	0.602D-23
70	5094.274	9 2 8	10 2 9			1288.271	0.498D-24	70	5185.570	4 4 2	6	4	3	751.036	0.235D-23
60	5094.471	5 0 5	6 1 6			445.346	0.105D-24	70	5185.722	4 2 2	5	2	3	445.159	0.127D-22
60	5096.577	3 3 1	2 4 2			298.621	0.204D-24	70	5186.115	1 5 1	5	1	6	445.346	0.179D-22
70	5096.902	8 2 6	9 2 7			1198.210	0.665D-24	70	5186.204	8 1 7	8	3	6	1001.711	0.128D-24
70	5098.816	8 4 4	9 4 5			1355.221	0.267D-24	70	5188.377	4 1 3	5	1	4	398.361	0.179D-22
70	5099.705	9 0 9	10 0 10			1109.785	0.416D-24	60	5192.553	3 3 2	6	2	5	550.453	0.646D-24
70	5099.735	9 1 9	10 1 10			1109.801	0.125D-23	60	5195.341	2 0 2	2	1	1	94.789	0.192D-24
70	5104.276	3 1 2	4 3 1			380.703	0.956D-25	70	5196.989	5 5 0	6	5	1	880.115	0.231D-24
60	5106.208	6 1 6	6 2 5			550.452	0.904D-25	70	5197.024	5 5 1	6	5	2	880.080	0.693D-24
70	5108.415	8 3 6	9 3 7			1211.191	0.196D-24	70	5197.934	4 3 1	5	3	2	505.730	0.732D-23
70	5109.976	8 4 5	9 4 6			1334.485	0.987D-25	70	5198.073	6 0 6	6	2	5	550.453	0.723D-24
70	5110.039	7 3 4	8 3 5			1047.331	0.307D-24	70	5198.472	4 2 3	5	2	4	414.170	0.413D-23
60	5113.095	2 1 1	3 2 2			204.768	0.949D-25	70	5198.932	7 1 6	7	3	5	812.766	0.872D-25
70	5113.152	8 1 7	9 1 8			1074.763	0.125D-23	70	5199.238	1 0 1	1	1	0	134.785	0.181D-24
60	5113.827	4 0 4	5 1 5			325.216	0.609D-25	70	5201.675	4 3 2	5	3	3	500.597	0.250D-23
70	5114.909	8 2 7	9 2 8			1075.911	0.411D-24	60	5202.616	1 0 1	1	1	1	42.024	0.503D-24
70	5115.489	7 2 5	8 2 6			980.229	0.567D-24	70	5205.040	4 0 4	5	2	4	324.047	0.267D-22
70	5117.943	8 5 3	9 5 4			1468.616	0.135D-24	60	5206.382	4 3 1	5	2	4	114.170	0.863D-24
60	5118.549	4 1 4	5 0 5			324.047	0.154D-24	70	5206.654	4 1 4	5	1	5	325.216	0.848D-23
70	5121.903	8 0 8	9 0 9			916.258	0.285D-23	70	5206.822	6 1 5	6	3	4	645.385	0.412D-24
70	5121.971	8 1 8	9 1 9			916.297	0.949D-24	60	5207.220	3 3 1	4	2	2	314.461	0.568D-24
60	5124.234	7 4 4	8 3 5			1047.331	0.847D-25	70	5207.447	4 3 2	4	3	2	379.291	0.336D-24
70	5124.648	2 1 1	3 2 3			282.308	0.143D-24	70	5207.947	6 1 6	6	5	1	541.179	0.269D-24
60	5125.200	1 1 1	2 2 2			134.785	0.163D-24	70	5209.376	5 2 4	5	3	3	500.597	0.153D-24
70	5127.344	5 1 5	5 2 4			414.170	0.501D-25	70	5210.178	3 1 2	2	2	3	274.803	0.861D-23
70	5128.515	7 4 3	8 4 4			1126.443	0.226D-24	70	5210.591	2 0 7	3	2	2	980.229	0.564D-25
70	5128.763	6 2 5	6 3 4			645.385	0.817D-25	70	5224.184	3 0 3	4	0	4	604.794	0.258D-23
70	5130.334	7 3 3	8 3 6			1001.711	0.139D-23	70	5211.165	4 1 1	5	4	2	604.547	0.861D-24
70	5130.957	3 0 3	4 1 4			314.461	0.243D-24	60	5211.708	5 4 1	6	3	4	645.385	0.113D-24
60	5131.730	3 0 3	4 1 4			223.829	0.273D-24	70	5213.265	3 2 1	4	2	2	314.461	0.527D-23
70	5132.042	8 6 2	9 6 3			1619.071	0.524D-25	60	5215.428	3 2 1	4	1	4	223.829	0.373D-24
60	5132.065	1 1 0	2 2 1			133.477	0.364D-24	70	5215.935	5 0 5	5	2	5	414.170	0.421D-24
70	5132.205	7 1 6	8 1 7			879.493	0.940D-24	60	5216.705	4 2 2	5	1	5	325.216	0.320D-24
70	5134.686	7 4 4	8 4 5			1116.639	0.711D-24	70	5220.917	3 2 2	4	0	4	298.621	0.128D-22
70	5134.773	9 1 9	9 1 8			1074.763	0.617D-25	70	5224.184	3 0 3	4	0	4	221.234	0.116D-22
60	5134.869	4 3 2	4 4 1			482.646	0.624D-25	60	5225.196	3 3 0	4	3	2	298.621	0.589D-23
70	5135.507	7 2 6	8 2 7			881.917	0.275D-23	70	5225.490	3 3 0	4	3	2	380.703	0.246D-23
70	5136.380	6 3 3	7 3 4			780.456	0.39D-23	60	5225.529	3 2 3	4	1	6	445.346	0.595D-25
70	5139.336	6 3 3	7 3 4			839.555	0.249D-23	70	5226.690	3 3 1	4	2	2	379.291	0.745D-23
60	5140.338	7 1 6	7 2 5			780.456	0.586D-25	70	5227.851	3 1 3	4	1	4	223.829	0.332D-22
60	5141.064	3 1 3	4 0 4			221.234	0.676D-25								

TABLE V—Continued

NO	SIGMA	J'KA'KC'	J	KA	KC	E	INTENSITY	NO	SIGMA	J'KA'KC'	J	KA	KC	E	INTENSITY				
60	5253.380	2	0	2	1	1	1	36.750	0.911D-25	60	5347.189	6	1	5	5	2	2	414.170	0.511D-25
70	5257.067	4	1	4	4	1	3	274.803	0.977D-24	60	5348.066	7	0	7	6	1	6	445.346	0.483D-24
60	5258.960	3	1	2	3	0	3	136.338	0.303D-24	60	5348.895	7	1	7	6	0	6	444.853	0.163D-24
70	5259.412	1	1	0	2	1	1	94.789	0.832D-23	70	5349.776	9	6	4	9	0	3	1619.071	0.729D-25
60	5259.985	1	1	1	0	0	0	0.0	0.126D-24	60	5351.363	5	4	1	5	3	2	505.730	0.233D-24
70	5261.519	6	2	5	6	2	4	601.238	0.403D-24	70	5351.626	8	6	3	8	6	2	1399.471	0.805D-25
60	5261.821	3	1	2	2	2	1	133.477	0.638D-25	70	5351.576	8	6	2	8	6	3	1399.437	0.241D-24
70	5264.095	1	0	1	2	0	2	69.928	0.123D-22	70	5353.083	7	6	2	7	6	1	1204.177	0.717D-24
60	5267.116	4	2	2	4	1	3	274.803	0.221D-24	70	5353.090	7	6	1	7	6	2	1204.170	0.239D-24
70	5269.578	1	1	1	2	1	2	78.990	0.270D-22	60	5353.110	4	2	3	3	1	2	172.884	0.521D-24
70	5272.275	6	3	6	8	3	5	1047.331	0.722D-25	70	5353.353	2	1	2	1	1	1	36.750	0.106D-22
60	5272.514	5	2	3	5	1	4	398.361	0.202D-24	70	5354.276	6	6	1	6	0	0	1033.195	0.640D-24
60	5275.238	4	1	3	4	0	4	221.234	0.536D-25	70	5354.279	6	6	0	6	6	1	1033.195	0.192D-23
60	5276.569	3	0	3	2	1	2	78.990	0.556D-25	60	5355.862	4	4	0	4	3	1	380.703	0.836D-25
70	5277.196	6	3	7	1	6	6	701.695	0.641D-25	70	5356.013	2	0	2	1	0	1	23.756	0.446D-22
60	5277.461	2	1	1	0	1	2	23.756	0.590D-24	60	5356.366	5	4	2	5	3	3	500.597	0.967D-25
70	5278.796	3	1	3	3	1	2	172.884	0.690D-23	60	5356.401	6	4	3	6	3	4	645.385	0.253D-24
70	5282.348	5	2	4	5	2	3	445.159	0.288D-23	60	5357.259	4	1	4	3	3	2	379.291	0.266D-24
70	5284.380	2	2	4	1	2	1	78.990	0.700D-25	60	5358.799	7	4	4	7	3	5	812.766	0.871D-25
70	5286.112	0	0	0	1	0	1	23.756	0.233D-22	60	5360.267	3	2	1	2	1	2	78.990	0.147D-23
60	5290.245	7	3	4	7	2	5	780.456	0.501D-25	60	5362.493	8	0	8	7	1	7	583.395	0.105D-24
60	5292.370	3	1	3	2	0	2	69.928	0.256D-24	60	5362.828	8	1	3	4	1	4	223.829	0.316D-24
70	5292.490	7	3	5	7	3	4	839.555	0.739D-24	70	5362.909	4	1	3	4	1	4	1574.658	0.143D-24
60	5295.249	5	1	4	5	0	5	324.047	0.727D-25	70	5363.712	8	7	1	8	7	2	42.024	0.311D-22
70	5295.314	2	1	2	2	1	1	94.789	0.458D-23	70	5364.930	2	1	1	1	0	0	1378.970	0.423D-24
60	5297.475	4	0	4	3	1	3	141.568	0.246D-24	60	5365.007	5	2	4	4	1	3	274.803	0.140D-24
60	5297.847	5	3	2	5	2	3	445.159	0.505D-24	70	5365.036	7	7	1	7	7	0	1378.970	0.141D-24
70	5298.181	4	2	3	4	2	2	314.461	0.219D-23	70	5365.036	7	7	0	7	7	1	1378.970	0.141D-24
60	5302.165	4	2	3	4	1	4	223.829	0.624D-25	70	5366.383	6	2	4	6	2	5	550.453	0.175D-23
70	5305.303	4	3	1	5	1	4	398.361	0.133D-24	70	5368.802	8	3	5	8	3	6	1001.711	0.255D-24
70	5305.335	9	4	6	9	4	5	1355.221	0.881D-25	60	5370.341	7	1	6	6	2	5	550.453	0.148D-24
60	5306.091	4	3	1	4	2	2	314.461	0.686D-24	70	5372.690	7	2	2	1	2	2	78.990	0.444D-22
70	5306.250	4	1	4	3	0	3	136.338	0.854D-24	70	5374.777	8	8	0	8	8	1	1768.562	0.734D-25
70	5306.492	3	2	1	4	0	4	221.234	0.947D-25	70	5375.490	3	2	2	3	0	2	59.928	0.173D-22
70	5306.544	1	1	1	1	1	0	42.024	0.317D-22	60	5375.787	6	2	5	5	1	4	398.361	0.306D-24
70	5306.834	4	2	2	5	0	5	324.047	0.167D-24	60	5376.090	9	0	9	8	1	8	740.999	0.182D-24
70	5307.401	6	3	4	2	6	3	658.609	0.773D-24	60	5376.304	9	1	9	8	0	8	740.918	0.609D-25
70	5308.738	3	2	2	3	2	2	210.800	0.130D-22	70	5380.175	2	2	2	3	0	3	69.928	0.416D-24
70	5310.633	3	3	0	3	2	2	134.785	0.600D-25	70	5382.200	2	2	2	3	0	3	136.338	0.175D-23
60	5313.017	8	4	6	8	4	5	1126.443	0.112D-24	70	5390.302	4	1	4	3	1	3	141.568	0.153D-22
70	5315.104	2	2	2	3	0	3	210.800	0.196D-24	60	5390.340	3	3	0	2	2	1	133.477	0.466D-23
70	5315.318	2	2	1	3	0	3	136.338	0.147D-24	70	5391.808	3	2	2	5	2	1	541.179	0.671D-25
60	5315.988	5	0	5	4	1	4	223.829	0.762D-24	60	5386.896	3	3	1	2	2	0	134.785	0.139D-23
70	5316.743	5	3	3	5	3	2	505.730	0.646D-23	70	5387.479	3	3	0	5	0	3	136.338	0.607D-24
60	5316.923	3	3	1	3	2	1	204.758	0.539D-24	70	5389.119	7	2	5	7	2	6	706.599	0.244D-24
70	5317.451	1	1	0	2	1	1	36.750	0.109D-22	70	5389.182	10	1	10	9	0	9	916.258	0.948D-25
70	5317.965	2	2	2	2	1	1	133.477	0.445D-25	70	5390.192	7	1	2	2	1	1	94.789	0.140D-22
70	5318.018	8	4	5	8	4	4	1126.443	0.112D-24	70	5390.302	4	1	4	3	1	3	141.568	0.153D-22
70	5318.287	4	0	3	2	3	2	210.800	0.196D-24	60	5390.340	3	3	0	2	2	1	133.477	0.466D-23
60	5318.894	4	3	2	3	0	3	298.621	0.122D-24	70	5391.408	4	2	3	4	0	4	221.238	0.758D-22
60	5320.280	5	1	5	4	0	8	221.234	0.271D-24	70	5392.749	4	0	4	3	0	3	136.338	0.492D-22
60	5320.675	5	1	4	4	2	0	298.621	0.170D-24	70	5392.941	2	1	2	2	2	0	134.785	0.608D-23
60	5321.346	2	2	2	1	1	0	42.024	0.424D-24	60	5395.761	6	5	2	6	4	3	751.036	0.561D-25
70	5321.569	4	3	2	1	4	3	380.703	0.525D-23	60	5398.733	5	0	5	4	1	1	604.794	0.619D-25
70	5322.968	3	2	1	3	2	2	204.758	0.640D-23	60	5398.996	8	2	7	7	1	6	701.695	0.120D-24
70	5323.075	3	3	1	3	3	0	282.306	0.344D-22	60	5399.318	4	3	1	4	0	4	221.238	0.161D-24
70	5323.098	3	3	3	1	3	0	282.095	0.115D-22	60	5400.353	4	2	3	3	1	3	141.568	0.310D-24
70	5324.373	4	3	1	4	3	2	379.291	0.159D-22	70	5403.460	5	2	4	5	0	5	324.047	0.196D-23
70	5325.621	7	4	2	7	4	3	925.704	0.110D-23	60	5406.715	4	3	2	3	3	1	210.800	0.221D-23
70	5326.919	5	3	2	5	3	3	500.597	0.221D-23	70	5406.861	6	1	5	6	4	1	445.346	0.137D-23
60	5327.856	2	2	2	1	1	1	36.750	0.184D-24	70	5407.632	9	1	5	6	4	1	223.829	0.396D-22
70	5327.964	4	1	1	2	1	1	78.990	0.148D-22	60	5407.762	9	1	5	6	4	2	881.917	0.591D-24
70	5329.690	4	1	3	2	0	1	752.191	0.105D-23	70	5407.684	2	0	3	2	2	2	204.758	0.832D-23
70	5331.872	2	2	2	4	4	0	604.794	0.788D-23	70	5408.871	6	0	5	6	5	0	221.234	0.135D-22
70	5331.948	6	4	4	4	4	1	751.036	0.316D-23	70	5413.195	4	4	3	3	3	1	881.917	0.301D-24
70	5332.260	4	4	2	4	2	3	298.621	0.929D-23	70	5413.854	4	4	3	3	3	1	172.884	0.396D-22
70	5332.299	6	0	6	5	4	5	604.547	0.263D-23	70	5415.794	4	3	2	3	2	2	204.758	0.968D-24
60	5332.687	6	0	6	5	4	5	325.216	0.218D-23	70	5417.904	6	5	2	6	4	3	444.853	0.433D-24
70	5333.036	4	4	1	4	4	0	482.676	0.597D-23	60	5418.959	5	3	2	2	2	2	324.047	0.139D-24
70	5333.058	7	4	3	0	4	4	921.900	0.376D-24	70	5420.081	4	2	3	3	3	2	210.800	0.255D-22
70	5333.091	7	4	3	0	4	4	482.646	0.179D-22										

TABLE V—Continued

NO	SIGMA	$J'K_a'K_c'$	J	K _a	K _c	E	INTENSITY	NO	SIGMA	$J'K_a'K_c'$	J	K _a	K _c	E	INTENSITY
60	5444.443	7 3 5	6 2 4			601.238	0.602D-25	70	5519.812	13 1 13	12 1 12	1551.195	0.154D-24		
60	5444.385	5 2 3	4 2 3	298.621	0.769D-24	70	5522.561	8 4 5	7 4 5	6 6 0	921.900	0.554D-24			
70	5446.664	5 2 3	4 2 2	314.461	0.742D-23	70	5524.065	7 6 1	6 6 0	1033.195	0.102D-24				
70	5446.613	5 3 2	4 3 1	380.703	0.244D-23	70	5524.065	7 6 2	6 6 1	1033.195	0.306D-24				
70	5446.916	8 1 7	8 1 6	740.999	0.336D-24	70	5525.880	9 3 7	8 3 6	1001.711	0.145D-23				
70	5448.587	6 2 5	5 2 4	114.170	0.574D-23	70	5526.991	3 3 1	2 1 2	78.990	0.226D-24				
70	5449.902	6 2 5	6 0 8	740.918	0.111D-24	60	5527.451	7 4 3	6 3 4	645.385	0.601D-25				
60	5451.465	8 3 6	7 2 5	780.456	0.767D-25	70	5528.123	8 4 4	7 4 3	925.704	0.164D-23				
60	5451.753	3 3 1	2 2 0	69.928	0.619D-25	70	5530.780	4 3 1	3 1 2	172.884	0.665D-24				
70	5452.828	9 3 7	9 1 8	1074.763	0.111D-24	70	5530.903	11 2 10	10 2 9	1288.271	0.460D-24				
70	5453.846	6 1 5	5 1 4	398.361	0.196D-22	70	5530.958	8 3 5	7 3 4	839.555	0.198D-23				
70	5453.960	5 2 4	4 4 1	482.646	0.439D-23	70	5531.138	11 1 10	10 1 9	1287.727	0.194D-24				
70	5454.176	5 4 1	4 4 0	482.676	0.146D-23	70	5532.295	9 2 7	8 2 6	980.229	0.556D-24				
70	5454.263	8 1 8	7 1 7	583.985	0.363D-23	70	5532.531	6 4 2	6 2 5	550.453	0.833D-25				
70	5454.379	8 0 8	7 0 7	583.782	0.109D-22	70	5534.778	7 5 3	7 5 3	1050.994	0.243D-24				
70	5457.798	3 2 1	2 0 2	69.928	0.472D-24	70	5535.352	8 5 3	7 5 2	1051.207	0.727D-24				
70	5465.413	6 3 4	5 3 3	500.597	0.326D-23	70	5539.891	5 2 3	4 0 4	221.234	0.226D-24				
70	5466.287	9 2 8	9 0 9	916.256	0.145D-24	70	5543.783	10 3 6	9 3 7	1211.191	0.196D-24				
70	5466.971	7 2 6	6 2 5	550.453	0.106D-22	70	5543.917	9 4 6	8 4 5	1116.639	0.765D-24				
70	5468.537	9 1 9	8 1 8	740.999	0.567D-23	70	5545.193	12 2 11	11 2 10	1518.798	0.543D-25				
70	5468.572	9 0 9	8 0 8	740.918	0.189D-23	70	5545.345	12 1 11	11 1 10	1518.541	0.163D-24				
70	5470.519	7 1 6	6 1 5	541.179	0.376D-23	70	5546.927	8 6 3	7 6 2	1204.170	0.829D-25				
70	5470.869	7 4 4	7 2 5	780.456	0.137D-24	70	5546.936	8 6 2	7 6 1	1204.177	0.299D-24				
70	5471.677	6 2 4	5 2 3	445.159	0.150D-22	70	5548.183	10 2 8	9 2 7	1198.210	0.637D-24				
70	5473.161	6 3 3	5 3 2	935.730	0.947D-23	70	5552.713	5 3 2	4 1 3	274.803	0.260D-24				
70	5477.334	6 4 3	5 4 2	604.547	0.148D-23	70	5552.856	9 3 6	8 3 5	1047.331	0.370D-24				
70	5478.190	6 4 2	5 4 1	604.704	0.444D-23	70	5553.954	9 4 5	8 4 4	1126.443	0.244D-24				
70	5479.835	4 3 3	4 4 1	223.829	0.167D-24	70	5557.212	9 5 5	8 5 4	1246.380	0.355D-24				
70	5480.643	6 4 3	5 6 2	601.238	0.521D-25	70	5558.775	12 9 5	4 8 5	1247.210	0.118D-24				
60	5481.177	4 3 3	3 0 3	136.338	0.669D-25	70	5559.012	13 2 12	12 2 11	1767.335	0.523D-25				
70	5481.482	10 1 9	10 1 10	1109.801	0.573D-23	70	5559.400	8 7 1	7 7 0	1378.970	0.571D-25				
70	5482.191	10 1 10	9 1 9	916.297	0.889D-24	70	5560.608	11 3 9	10 3 8	1440.291	0.213D-24				
70	5482.218	10 0 10	9 0 9	916.258	0.267D-23	70	5560.704	10 3 2	9 3 1	1415.568	0.103D-24				
70	5484.221	8 2 7	7 2 6	706.599	0.190D-23	70	5563.253	11 2 9	10 2 8	1433.030	0.740D-25				
70	5486.220	8 1 7	7 1 6	701.695	0.588D-23	70	5564.210	10 4 7	9 4 6	1334.465	0.101D-24				
70	5486.660	7 3 5	6 3 4	645.385	0.604D-23	70	5569.410	9 5 6	8 6 3	1399.437	0.136D-24				
70	5488.720	6 5 2	5 5 1	733.681	0.436D-24	70	5573.638	10 3 7	9 3 6	1279.860	0.418D-24				
70	5488.752	6 5 1	5 5 0	733.685	0.131D-23	70	5577.981	12 2 10	11 2 9	1684.437	0.710D-25				
70	5491.147	5 4 2	5 2 3	445.159	0.133D-24	70	5578.983	10 4 6	9 4 5	1355.221	0.270D-24				
70	5494.480	7 2 5	6 2 4	601.238	0.277D-23	70	5580.530	6 3 3	5 1 4	398.361	0.580D-24				
70	5494.543	4 2 2	3 0 3	136.338	0.128D-23	70	5582.595	10 5 5	9 5 4	1468.616	0.144D-24				
70	5495.288	11 1 11	10 1 10	1109.801	0.114D-23	70	5583.369	11 4 8	10 4 7	1574.458	0.106D-24				
70	5495.296	11 0 10	10 0 10	1109.785	0.378D-24	70	5591.864	10 6 4	9 6 3	1619.071	0.592D-25				
60	5496.056	6 4 3	5 3 2	505.730	0.262D-24	70	5592.789	6 2 4	5 0 5	324.047	0.283D-24				
70	5498.761	7 3 3	6 2 3	658.609	0.167D-23	70	5598.644	5 3 3	4 1 4	223.829	0.259D-24				
70	5500.289	7 4 4	6 2 3	751.036	0.304D-23	70	5599.927	11 5 6	10 5 6	1709.575	0.511D-25				
70	5500.628	9 2 6	8 2 7	881.917	0.273D-23	70	5604.937	4 4 0	3 3 2	210.800	0.151D-24				
70	5501.533	9 1 8	8 1 7	879.493	0.925D-24	70	5616.191	7 3 4	6 1 5	541.179	0.855D-25				
70	5502.767	7 4 3	6 4 2	752.191	0.101D-23	70	5622.387	5 4 1	4 2 2	314.461	0.723D-25				
70	5506.840	8 3 6	7 3 5	812.766	0.106D-23	70	5637.825	6 4 2	5 2 3	445.159	0.214D-24				
70	5507.831	12 1 12	11 1 11	1321.454	0.146D-24	70	5637.985	5 4 2	4 2 3	298.621	0.194D-24				
70	5507.838	12 0 12	11 0 11	1321.457	0.438D-24	70	5640.794	6 3 4	5 1 4	325.216	0.543D-25				
70	5511.404	13 3 0	2 2 1	94.789	0.113D-24	70	5653.720	7 4 3	6 2 4	601.238	0.557D-25				
70	5511.907	7 5 3	6 5 2	880.080	0.119D-23	70	5657.711	6 4 3	5 2 4	414.170	0.542D-25				
70	5512.066	7 5 2	6 5 1	880.115	0.396D-24	70	5668.818	8 3 5	7 1 6	701.695	0.695D-25				
60	5512.956	7 4 4	6 3 3	658.609	0.285D-24	70	5673.381	8 4 4	7 2 5	780.456	0.101D-24				
70	5514.656	8 2 8	7 2 5	780.456	0.396D-23	70	5686.699	7 3 5	6 1 6	445.346	0.839D-25				
70	5516.019	10 2 9	9 2 8	1075.911	0.393D-24	70	5697.813	5 5 1	4 3 2	379.291	0.642D-25				
70	5516.520	10 1 9	9 1 8	1074.763	0.119D-23	70	5700.872	7 4 4	6 2 5	550.453	0.102D-24				
70	5517.116	4 4 0	4 2 3	298.621	0.626D-25	70	5716.707	6 5 1	5 3 2	505.730	0.748D-25				
60	5517.630	8 4 5	7 3 4	839.555	0.926D-24	70	5746.602	7 5 3	6 3 4	645.385	0.570D-25				
70	5519.812	13 0 13	12 0 12	1551.195	0.513D-25										

^a The meaning of the different columns is:

NO, Vibrational assignment of the line with 0 = (000), 6 = (110), 7 = (011);

SIGMA, wavenumber of the line in cm^{-1} ;

$J'K_a'K_c'$ and JK_aK_c , rotational quantum numbers of the upper and lower levels respectively;

E, lower level energy in cm^{-1} ;

INTENSITY, computed line intensity k_σ^N in $\text{cm}^{-1}/(\text{molecule cm}^{-2})$.

intensities for 88 lines of H_2^{18}O introduced in the fit is as follows:

$0 \leq \% < 10$, 83.0% of the lines,

$10 \leq \% < 16$, 10.2% of the lines,

$16 \leq \% \leq 23$, 6.8% of the lines,

where $\% = 100 \times |k_\sigma^P(\text{obs}) - k_\sigma^P(\text{calc})|/k_\sigma^P(\text{obs})$.

Using the set of constants listed in Table II we have computed the whole spectrum

TABLE VI
Comparison of the Intensities of H_2^{16}O and H_2^{18}O Lines in the Case of Resonance

		H_2^{16}O			H_2^{18}O		
$(v_1' v_2' v_3')[J' k_a' k_c']$	$(v_1 v_2 v_3)[J k_a k_c]$	σ (cm $^{-1}$)	k_{σ}^N (cm $^{-1}$ /molecule cm $^{-2}$)	ratio	σ (cm $^{-1}$)	k_{σ}^N (cm $^{-1}$ /molecule cm $^{-2}$)	ratio
(110) [330]	← (000) [221]	5403.900	0.361 10^{-20}	0.414	5390.340	0.228 10^{-20}	
(011) [322]	← (000) [221]	5409.349	0.873 10^{-20}		5386.061	0.868 10^{-20}	0.263
(110) [330]	← (000) [423]	5238.440	0.185 10^{-20}	0.222	5225.196	0.289 10^{-20}	
(011) [322]	← (000) [423]	5243.889	0.835 10^{-20}		5220.917	0.627 10^{-20}	0.461
(110) [431]	← (000) [322]	5427.089	0.309 10^{-20}	1.507	5415.794	0.484 10^{-21}	
(011) [423]	← (000) [322]	5433.433	0.205 10^{-20}		5407.884	0.408 10^{-20}	0.119
(110) [431]	← (000) [524]	5217.182	0.135 10^{-20}	0.993	5206.382	0.423 10^{-21}	
(011) [423]	← (000) [524]	5223.526	0.136 10^{-20}		5198.472	0.202 10^{-20}	0.209

^a Converted to pure H_2^{18}O

of the $\nu_1 + \nu_2$ and $\nu_2 + \nu_3$ bands of H_2^{18}O for the normal abundance of this isotope in natural water at a temperature of 296 K for lines having an intensity larger than 0.5×10^{-25} cm $^{-1}$ /molecule cm $^{-2}$ and the results are listed in Table V. The total band intensities (i.e., the sum of the individual line strengths) for the natural abundance at 296 K are $S_v(\nu_1 + \nu_2) = 0.53 \times 10^{-22}$ and $S_v(\nu_2 + \nu_3) = 0.15 \times 10^{-20}$ (in cm $^{-1}$ /molecule cm $^{-2}$).

It is interesting to make a comparison of line intensities for H_2^{16}O and H_2^{18}O when there is resonance. Table VI shows clearly that, for the pairs of resonating levels {(110)[330], (011)[322]} and {(110)[431], (011)[423]}, the ratio of intensities of the two perturbed lines originating from the same lower level can actually be very different from one isotopic species to another and this is a result of the differences of the mixing coefficients of the resonating levels for the two isotopic varieties.

CONCLUSION

Using a Fourier transform spectrum of water in the 1.9-μm region, we have measured about 90 lines intensities of H_2^{18}O in natural abundance. From these data and realistic wavefunctions for the vibration-rotation energy levels of H_2^{18}O , we have determined the constants involved in the expansion of the transformed transition operators corresponding to the $\nu_1 + \nu_2$ and $\nu_2 + \nu_3$ bands of this isotopic species. The whole spectrum of these two bands has been computed giving accurate positions and intensities for the vibration-rotation lines of H_2^{18}O in the region 4900 to 5800 cm $^{-1}$.

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